

**Kinetic Theory and Thermodynamics**

P. Pages : 4

Time : Three Hours

Max. Marks : 40

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- Note : 1. All questions are compulsory.  
2. Draw neat diagrams wherever necessary.

**Either**

1. a) State any four assumptions of kinetic theory of gases. **2**  
b) State and prove law of equipartition of energy. **3**  
c) Give the kinetic interpretation of temperature. **3**

**OR**

2. p) State Avogadro's law. **1**  
q) Derive Vander Waal's equation of state for real gas. **5**  
r) What are degrees of freedom ? Find degrees of freedom for monoatomic gas. **2**

**Either**

3. a) What is Joule Thomson Effect ? 1  
 b) Describe with neat diagram porous plug experiment for liquification of gases. 4  
 c) Obtain an expression for the coefficient of thermal conductivity of a gas. 3

**OR**

4. p) Describe with neat diagram Method for liquefaction of Hydrogen. 4  
 q) Describe with neat diagram method for liquefaction of Helium. 4

**Either**

5. a) Explain  
     i) Reversible process.  
     ii) Irreversible process with one example of each.  
 b) What are the limitations of first law of thermodynamics. 2  
 c) On the basis of first law of thermodynamics explain (i) free expansion (ii) Cyclic process. 3

**OR**

6. p) State zeroth law of thermodynamics. 1  
 q) State and prove carnot's theorem. 4  
 r) A Carnot's engine works between the steam point and ice point. Find its efficiency. 3

**Either**

7. a) Define extensive and intensive variable. 2  
 b) Derive Clausius - Clapeyron's latent heat equation by using Maxwell's thermodynamics relation. 5  
 c) Define Internal energy function (U) 1

**OR**

8. p) Explain the enthalpy function (H). 2  
 q) Derive Maxwell's thermodynamical relation  

$$\left(\frac{\partial P}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T 3$$
  
 r) Derive Maxwell's thermodynamical relation  

$$\left(\frac{\partial T}{\partial V}\right)_S = - \left(\frac{\partial P}{\partial S}\right)_V 3$$

**Either**

- 9.** a) What is perfectly black body ? **2**  
b) State and explain Stefan's law of radiation. **3**  
c) State and Explain Boltzman's law. **3**

**OR**

- 10.** p) State and prove Wein's displacement law. **3**  
q) State and prove Rayleigh - Jean's law. **3**  
r) What is ultraviolet catastrophe ? **2**

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