| | D | IPLOMA - CO | MMON ENTR | ANCE | |
|---|---|--|--|---|--|
| | | COU | RSE | DAY: | SUNDAY DATE: 30-JUNE-2013 |
| Ch | 1 | CHEM ENGINE | | TIM | IE : 9.00 a.m. to 12.00 Noon |
| MAXIMUM N | IARKS | TOTAL DU | JRATION | MAXII | MUM TIME FOR ANSWERING |
| 180 | <i>,,</i> (, (, (, | 200 Mi | | | 180 Minutes |
| | N VOLL | R DIPLOMA | | | |
| | ET NUM | | QI | JESTION | I BOOKLET DETAILS |
| | | | VERSION | CODE | SERIAL NUMBER |
| | | | A- | 2 | 107082 |
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| Choose of question consider Complete | ne correct titem. In other the best. In ly darken | ase you feel that the any case, choose of shade the relevan | ere is more than o I nly one response | ne correct for each ite | es (options / choices) given under ex response, mark the response which yem. em. ink ballpoint pen against the quest |
| number (| on the OM | R answer sheet. thod of shading the | circle on the OMI | R answer s | heet is as shown below : |
| | | | ① ① ③ | 4 | |
| 4. Use the spa | 1 | | | | Work. Do not use the OMR answer sh |
| 5. After the las | t bell is ru on the OM | IR answer sheet as p | per the instructions | i. | wer sheet and affix your left hand thu |
| 6. Hand over t | ne OMR au | nswer sheet to the re | oom invigilator as i | t is. | nottom sheet renlica (candidate's con- |
| 7. After separa | ting the to home for: | p sheet (KEA copy), self-evaluation. | the invigilator will | ietum me t | pottom sheet replica (candidate's copy |
| O Dranama th | renlica of | the OMR answer st | neet for a minimum | period of (| ONE year. [P |

DOs:

- 1. Check whether the Diploma CET No. has been entered and shaded in the respective circles on the OMR answer
- 2. This question booklet is issued to you by the invigilator after the 2nd bell i.e., after 08.50 a.m.
- 3. The serial number of this question booklet should be entered on the OMR answer sheet.
- 4. The version code of this question booklet should be entered on the OMR answer sheet and the respective circles should also be shaded completely.
- 5. Compulsorily sign at the bottom portion of the OMR answer sheet in the space provided.

DON'Ts:

- 1. THE TIMING AND MARKS PRINTED ON THE OMR ANSWER SHEET SHOULD NOT BE DAMAGED / MUTILATED/SPOILED.
- 2. The 3rd Bell rings at 9.00 a.m., till then;
 - Do not remove the seal / staple present on the right hand side of this question booklet.
 - Do not look inside this question booklet.
 - Do not start answering on the OMR answer sheet.
- 1. This question booklet contains 180 (items) questions and each question will have one statement and four answers. (Four different options / responses.)
- 2. After the 3rd Bell is rung at 9.00 a.m., remove the paper seal / polythene bag of this question booklet and check that this booklet does not have any unprinted or torn or missing pages or items etc., if so, get it replaced by a complete test booklet. Read each item and start answering on the OMR answer sheet.
- 3. During the subsequent 180 minutes:
 - Read each question (item) carefully.
 - Choose one correct answer from out of the four available responses (options / choices) given under each question / item. In case you feel that there is more than one correct response, mark the response which you consider the best. In any case, choose only one response for each item.
 - Completely darken / shade the relevant circle with a blue or black ink ballpoint pen against the question number on the OMR answer sheet.

Correct Method of shading the circle on the OMR answer sheet is as shown below: 4 (3)

- 4. Use the space provided on each page of the question booklet for Rough Work. Do not use the OMR answer sheet for the same.
- 5. After the last bell is rung at 12.00 Noon, stop marking on the OMR answer sheet and affix your left hand thumb impression on the OMR answer sheet as per the instructions.
- 6. Hand over the **OMR answer sheet** to the room invigilator as it is.
- 7. After separating the top sheet (KEA copy), the invigilator will return the bottom sheet replica (candidate's copy) to you to carry home for self-evaluation.



PART-A

It consists of 1 - 40 questions.

- 1. In solving the equations by Cramer's rule for 5x 3y = 1 and 2x 5y = -11, the value of x and y is
 - (1) (3, 2)

(2) (-3, -2)

(3)(2,3)

- (4) (-2, -3)
- 2. If $A = \begin{bmatrix} 2 & 0 & 0 \\ 1 & 2 & 0 \\ 1 & 1 & 2 \end{bmatrix}$ then A adj A is
 - (1) Diagonal

(2) Scalar

(3) Identity

- (4) Zero matrix
- 3. The minor of the element 6 in a matrix $A = \begin{bmatrix} 2 & -3 & 0 \\ 4 & 1 & 6 \\ 3 & 2 & 0 \end{bmatrix}$ is
 - (1) 10

(2) 11

(3) 12

- (4) 13
- 4. The characteristic equation of the matrix $A = \begin{bmatrix} 5 & -3 \\ 2 & 1 \end{bmatrix}$ is
 - (1) $\lambda^2 6\lambda + 11 = 0$
- (2) $\lambda^2 6\lambda 11 = 0$
- (3) $\lambda^2 + 6\lambda + 11 = 0$

- $(4) \lambda^2 + 6\lambda = 0$
- 5. The fourth term in the expansion of $(\sqrt{3} + 2)^7$ is
 - (1) 2520

(2) - 2520

(3) 1/2520

(4) - 1/2520

- 6. The value of ($\sin 100^{\circ} + \sin 20^{\circ}$) / ($\cos 100^{\circ} + \cos 20^{\circ}$) is
 - (1) $\sqrt{3}/2$

(2) 1/2

(3) $\sqrt{3}$

- (4) 1
- 7. The value of $(\tan^{-1} 5/6 + \tan^{-1} 1/11)$ is
 - (1) 30°

(2) 60°

 $(3) 90^{\circ}$

- (4) 45°
- 8. If the points (-3, K), (5, 7) and (-11, 1) are collinear, then the value of K is
 - (1) 4

(2) 3

(3) 2

- (4) 1
- 9. The ratio of the line join of the points (2, 3) and (-5, 6) divided by y axis is
 - (1) 5:2

(2) 2:5

(3) 3:2

- (4) 2:3
- 10. Three vertices of a triangle are (-2, 3, 1), (-1, 4, 2) and (-6, 5, 2), then the centroid of the triangle is
 - (1) (-3, 4, 1)

(2) (0, 5/3, 1/3)

(3) (4, 3, 1)

- (4) (-3, -4, -2)
- 11. The volume of a sphere is increasing at the rate of 4π c.c/sec, then the rate of increase of the radius is when the volume is 288 π cc
 - (1) 6 cm/sec

(2) 1/6 cm/sec

(3) 1/36 cm/sec

(4) 36 cm/sec

- 12. $\int \sin^2 x \, dx$ is
 - (1) $\cos x + c$

(2) $x/2 - (\sin 2x)/4 + c$

(3) $x/2 + (\cos 2x)/4 + c$

 $(4) x/2 + (\sin 2x)/4 + c$

13.
$$\int (3x^2 + x - 1)^6 (6x + 1) dx$$
 is

(1) $6(3x^2 + x - 1)^5 + c$

(2) $(3x^2 + x - 1)^6 + c$

(3) $(3x^2 + x - 1)^7 / 7 + c$

(4) $(3x^2 + x - 1)^7 / 21 + c$

14.
$$\int \tan^{-1} x \, dx$$
 is

- (1) $x \tan^{-1} x 1/2 \log (1 + x^2) + c$
- (2) $x \tan^{-1} x + 1/2 \log (1 + x^2) + c$
- (3) $\tan^{-1} x 1/2 \log (1 + x^2) + c$
- (4) $tan^{-1} x + 1/2 log (1 + x^2) + c$

15.
$$\int_{0}^{\pi/2} \sin 3x \cos 2x \, dx$$
 is

(1) 3/5

(2) - 3/5

(3) 5/3

- (4) 5/3
- 16. The constant term in the expansion $(x^2 + 1/x)^{12}$ is
 - (1) 495

(2) 495

(3) 1/495

- (4) 945
- 17. The projection of vector (3, 1, 3) on vector (1, -2, 1) is
 - (1) $2\sqrt{6}/5$

(2) $-2\sqrt{6}/3$

(3) $2\sqrt{6}/3$

- $(4) 2\sqrt{6}/5$
- 18. If vector a = (1, 1, 1) and vector b = (2, 2, 1) then magnitude of vector $a \times b$ is
 - (1) $\sqrt{26}$

(2) $\sqrt{28}$

(3) $\sqrt{24}$

(4) 1

- 19. The cosine of the angle between the vectors (3, -1, 1) and vector (1, 1, -1) is
 - (1) $1/\sqrt{11}$

(2) $-1/\sqrt{33}$

(3) $1/\sqrt{33}$

- (4) $-1/\sqrt{11}$
- 20. The value of $(\sec^6 x \tan^6 x)$ is
 - (1) $1 3 \sec^2 \times \tan^2 x$
 - (2) $1 + \tan^2 \times \sec^2 x$
 - (3) $1 + 3 \sec^2 \times \tan^2 x$
 - $(4) 1 \tan^2 \times \sec^2 x$
- 21. The equation to the straight line passing through (3, 2) and perpendicular to the line 5x + 2y 3 = 0 is
 - (1) 2x 5y 4 = 0
 - (2) 2x 5y + 4 = 0
 - (3) 2x + 5y + 4 = 0
 - (4) 5x 2y + 4 = 0
- 22. The slope of a line passing through the points (-4, -5) and (2, 3) is
 - (1) 3/4

(2) - 3/4

(3) 4/3

- (4) 4/3
- 23. The acute angle between the lines 2x y + 3 = 0 and x 3y + 2 = 0 is
 - (1) 30°

(2) 60°

(3) 90°

- (4) 45°
- 24. The value of $\lim_{n\to\infty} [(3-n)(4-n)(2n-5)]/(4n^3-3)$
 - (1) 1/2

(2) 1/2

 $(3) \ 3/2$

(4) - 3/2

- 25. The value of $\lim_{x\to -3} (x^4 81) / (x^3 + 27)$ is
 - (1) 3

(2) - 3

(3) 4

(4) - 4

- 26. $\int_{0}^{2} (x-1)(x-2) dx$ is
 - (1) 2/3

(2) - 2/3

(3) 3/2

- (4) 3/2
- 27. The area bounded by the curve $y = 2x^2$, the x axis and the ordinates at x = -1 and x = 2 is
 - (1) 6 sq units
 - (2) 3 sq units
 - (3) 3 sq units
 - (4) 6 sq units
- 28. The differential equation formed by eliminating a and b from $x + y = ae^{x} + be^{-x}$ is
 - (1) $d^2y/dx^2 + y = 0$
 - (2) $d^2y/dx^2 y = 0$
 - (3) $d^2y/dx^2 x y = 0$
 - (4) $d^2y/dx^2 + x y = 0$
- 29. The solution of the differential equation $\frac{dy}{dx} = \frac{1 + y^2}{1 + x^2}$ is
 - (1) $tan^{-1} y + tan^{-1} x + c = 0$
 - (2) $\log (1 + y^2) + \log (1 + x^2) + c = 0$
 - (3) $tan^{-1} y tan^{-1} x + c = 0$
 - (4) $\log (1 + y^2) \log (1 + x^2) + c = 0$

30. If
$$\begin{vmatrix} x+2 & 5 \\ 0 & x-2 \end{vmatrix} = 0$$
, then $x =$

(1) 1

(2) 2

(3) 3

(4) 0

31. If x cot 45° cos 60° = $\sin 60^{\circ}$ tan 30° then the value of x is

(1) $\sqrt{3}$

(2) $\sqrt{3}/2$

(3) 1/2

(4) 1

32. If $\tan x = 15/8$ and x is in the III quadrant then the value of $(2 \sin x - 3 \cos x) / (2 \cos x + 3 \sin x)$ is

(1) 61/6

(2) - 61/6

(3) - 6/61

(4) 6/61

33. The value of $\{[\sin(2\pi - \theta) + \cos(-\theta)]/[\tan(-\theta) + \cot(2\pi + \theta)]\} - \{[\sin(\pi/2 + \theta) + \cos(3\pi/2 - \theta)]/[\cot(\pi + \theta) + \tan(2\pi - \theta)]\}$ is

(1) 0

(2) - 1

(3) + 1

(4) - 2

34. If $\sin A = 5/13$ and $\sin B = 4/5$ then the value of $\cos (A - B)$ is

(1) 65/56

(2) 56/65

(3) 16/65

(4) - 16/65

35. On simplification the value of $(\cos^3 A - \cos 3 A) / \cos A + (\sin^3 A + \sin 3 A) / \sin A$ is

(1) 3

(2) 1

(3) 2

(4) 0

- 36. $d/dx \left(\sqrt{\sin^2 x} is\right)$
 - (1) cos x

(2) sin 2x

(3) $\cos^2 x$

- (4) $\sqrt{\cos x/\sin x}$
- 37. $d/dx tan^{-1} \sqrt{(1-\cos 2x)/(1+\cos 2x)}$ is
 - (1) 1

(2) 0

(3) tan x

(4) cos x

- 38. If $y = \sin x^x$ then dy/dx is
 - (1) x log sin x
 - (2) cos x^x
 - (3) $\sin x^x (x \cot x + \log \sin x)$
 - (4) $\cos x^x (x \tan x + \log \sec x)$
- 39. $d/dx \left(\sin h^{-1} x \right)$ is
 - (1) $1/\sqrt{1+x^2}$

(2) $1/\sqrt{1-x^2}$

(3) $1/\sqrt{x^2-1}$

- (4) $1/\sqrt{x^2+1}$
- 40. The equation to the normal to the curve $y = 5x^2 + 4x 11$ at the point (-1, 2) is
 - (1) x 6y + 11 = 0
 - (2) x + 6y 11 = 0
 - (3) 6x y + 11 = 0
 - (4) 6x + y 11 = 0

PART-B

It consists of **41 – 80** questions.

| 41. | Which of the following is dimensional phy- | sical quantity? |
|-----|---|---|
| | (1) pressure | (2) strain |
| | (3) mechanical advantage | (4) sp.gravity |
| 42. | The principle of vernier is | |
| | (1) $n VSD = (n + 1) MSD$ | (2) $(n-1) VSD = n MSD$ |
| | (3) $n MSD = (n-1) V SD$ | (4) $(n-1) MSD = n VSD$ |
| 43. | | 60 division on sleeve. The reading when the a wire the reading is PSR = 3 PSD and |
| | (1) 1.62 cm | (2) 0.162 cm |
| | (3) 0.162 mm | (4) 16.2 mm |
| 44. | The extension of the material by itself with (1) within elastic limit | out increase of load takes place |
| | (2) beyond elastic limit | |
| | (3) beyond yield point | |
| | (4) at breaking point | |
| 45. | If the strain in a wire is 0.1%, then the char | nge in the length of the wire of length 5 m is |
| | (1) 5×10^{-2} m | (2) 5×10^{-3} m |
| | (3) 5×10^{-4} m | (4) 5×10^{-3} cm |
| 16. | A force of 10 N acting on a body fixed at a pline of force is 2 m. Then the moment of the | point the distance from the fixed point to the e force is N-m. |
| | (1) 0.002 | (2) 0.02 |
| | (3) 2 | (4) 20 |
| | SPACE FOR BO | MICH MODE |



47. By Lami's theorem, P Q R are three forces acting in equilibrium and angle between PR, PQ, QR, are α , β , γ respectively then which of the following is correct?

$$(1) \frac{P}{\sin\beta} = \frac{Q}{\sin\gamma} = \frac{R}{\sin\alpha}$$

(2)
$$\frac{P}{\sin \gamma} = \frac{Q}{\sin \alpha} = \frac{R}{\sin \beta}$$

(3)
$$\frac{P}{\sin\alpha} = \frac{Q}{\sin\beta} = \frac{R}{\sin\gamma}$$

$$(4) \frac{P}{\sin\alpha} = \frac{Q}{\sin\gamma} = \frac{R}{\sin\beta}$$

- 48. If the line of action of the force passes through the point of rotation, then the moment of force is
 - (1) Maximum

(2) Less than one

(3) Greater than one

- (4) Zero
- 49. 1 Kilo calorie of heat is equal to _____ joule.
 - (1) 4.186

(2) 41.86

(3) 418.6

- (4) 4186
- 50. The correct relation between °F and K scale is
 - (1) 5K = 9 (F 32)
 - (2) 9K = -5(F 32)
 - (3) $K = \frac{9}{5} (F 32) 273$
 - (4) $K = \frac{5}{9} (F 32) + 273$
- 51. Two coherent sources $2\times 10^{-4}\,$ m apart are illuminated by the light of wave length 5000×10^{-10} m. The distance between the source and screen is 0.2m, then fringe width is
 - (1) 0.05×10^{-3} m
 - (2) 5×10^{-3} m
 - (3) 0.5×10^{-3} m
 - (4) 50×10^{-3} m



- 52. Resolving power of microscope is
 - (1) Equal to the resolution of the microscope
 - (2) Reciprocal to the resolution of the microscope
 - (3) Reciprocal to the focal length of the microscope
 - (4) Product of wave length and semi vertical angle
- 53. Which of the following phenomenon confirm that light is transverse wave?
 - (1) Diffraction
 - (2) Interference
 - (3) Refraction
 - (4) Polarization
- 54. In Field emission
 - (1) High positive voltage is used
 - (2) Secondary electrons are used
 - (3) High energy is used
 - (4) High radiations are used
- 55. Which of the following is not true?
 - (1) Photoelectric emission is an instantaneous process
 - (2) Photoelectric emission do not takes place below threshold frequency
 - (3) The K.E. of the photoelectron depends on the wavelength of incident radiation
 - (4) Number of photoelectrons emitted is directly proportional to the intensity
- 56. Poisson's ratio is the ratio of
 - (1) $\frac{Lateral\ strain}{Linear\ strain}$

(2) Linear strain

(3) Lateral strain
Volume strain

(4) Volume strain

| | | 13 | C |
|-----|---|--|------------------|
| 57. | The pressure at a depth of 100 m belo | ow the surface of water density 1000 kgm | ⁻³ is |
| | (1) $98 \times 10^5 \text{Nm}^{-2}$ | (2) $9.8 \times 10^4 \text{Nm}^{-2}$ | |
| | (3) $980 \times 10^4 \text{ Nm}^{-2}$ | (4) $98 \times 10^4 \mathrm{Nm}^{-2}$ | |
| 58. | When two capillary tube of different of height of the liquid is | diameters are dropped vertically in a liqu | id, the |
| | (1) More in the tube of larger diameter | ər | |
| | (2) More in the tube of smaller diame | eter | |
| | (3) Lesser in the tube of smaller diam | neter | |
| | (4) Same in both the tubes | | |
| 59. | The property by virtue of which a liquilayers is | uid opposes relative motion between its di | fferent |
| | (1) Viscosity | (2) Elasticity | |
| | (3) Surface tension | (4) Inertia | |
| 60. | The maximum amount of force acting | for a short duration is known as | |
| | (1) Momentum | (2) Inertia | |
| | (3) Power | (4) Impulse | |
| 61. | Absolute zero is the temperature of theoretically zero. | a gas at which, the of | gas is |
| | (1) Mass | (2) Weight | |

62. When the particle is in SHM having amplitude 'r', then its velocity is

(1)
$$v = \omega (r^2 - y^2)$$

(2)
$$v = \omega \sqrt{r^2 - y^2}$$

(3)
$$v = r\omega^2$$

(3) Volume

(4)
$$v = r\omega^3$$

(4) Density



| 63. | Ripples in water are the example for | | |
|---------------------------------------|--------------------------------------|---|--|
| | (1) | Transverse wave | |
| | (2) | Longitudinal wave | |
| | (3) | Sound wave | |
| | (4) | Ultrasonic wave | |
| 64. | The | e length of one ventral segment in stat | ionary wave is equal to |
| | (1) | Full wavelength of the wave | |
| | (2) | Twice the wavelength of the wave | |
| | (3) | Half a wavelength of the wave | |
| | (4) | Quarter a wavelength of the wave | |
| 65. | | tretched string under a tension T vibra | ates with a frequency f. When the tension is becomes |
| | (1) | same | (2) doubled |
| | (3) | tripled | (4) zero |
| 66. | The | appearance of additional frequencies | in scattered beam of light is known as |
| | (1) | Raman effect | |
| | (2) | Coherent scattering | |
| | (3) | Incoherent scattering | |
| | (4) | Bipolar scattering | |
| 67. | Two | properties of LASER are | |
| | (1) | Highly monochromatic and extremely | intense |
| | (2) | Highly chromatic and extremely fast | |
| | (3) | Very high frequency and extremely hi | gh wave length |
| · · · · · · · · · · · · · · · · · · · | (4) | Very high power and extremely low a | mplitude |
| | | SPACE FOR RO | OUGH WORK |



68. To form a galvanic cell

| | (1) difference in concentration of electrolyte is required | | |
|-----|--|--|--|
| | (2) difference in concentration of frequency is required | | |
| | (3) difference in concentration of amplitude is required | | |
| | (4) both (2) and (3) | | |
| 69. | pH value is not having its application in | | |
| | (1) determination of quality of soil | | |
| | (2) determination of quality of textile dye | es | |
| | (3) determination of quality of chemical | s | |
| | (4) determination of quality of electron | | |
| 70. | The prefix "mega" stands for | | |
| | (1) 10 ³ | $(2) 10^{-3}$ | |
| | ·(3) 10 ⁻⁶ | (4) 10 ⁶ | |
| 71. | A bullet of mass 0.01 kg is fired from a rift the recoil velocity of rifle is | le of mass 20 kg with a speed of 10 m/s , then m/s. | |
| | (1) -1 | (2) -0.05 | |
| | (3) -200.01 | (4) -0.005 | |
| 72. | Final velocity of a body thrown downwar | rds is | |
| | (1) Maximum | (2) Minimum | |
| | (3) No change | (4) Zero | |
| 73. | A person throws a sand bag from a boat | at rest in a pond then boat moves | |
| | (1) In the same direction | (2) In the opposite direction | |
| | (3) In a perpendicular direction | (4) In circular direction | |
| 74. | Two equal forces at a point, the square product of the forces. Then the angle be | e of their resultant is equal to three times the etween the forces is equal to | |
| | (1) 30° | (2) 45° | |
| | (3) 60° | (4) 90° | |
| | SPACE FOR | ROUGH WORK | |

| 75. Equilibrant i | is a force | 3 |
|-------------------|------------|---|
|-------------------|------------|---|

- (1) Which brings a body in equilibrium
- (2) Which moves the body along the resultant force
- (3) in zig-zag movement of the body
- (4) Which moves the body in opposite direction to equilibrant force

| 76. | The best value of reverberation time for speech listener | |
|-----|--|--|
|-----|--|--|

(1) 0.5 to 1.5 s

(2) 0.15 to 0.5 s

(3) 0.05 to 0.15 s

(4) 0.5 to 5 s

77. 3 strings of equal lengths but stretched with different tensions are made to vibrate, if their masses per unit length are in the ratio 3:2:1 and frequencies are same then the ratio of the tensions _____

(1) 1:2:3

(2) 2:3:1

(3) 1:3:2

(4) 3:2:1

78. Newton's formula for velocity of sound was corrected by

(1) Boyle

(2) Charles

(3) Laplace

(4) Hertz

79. Light waves are composed of both electric and magnetic field is proposed by

- (1) Newton's corpuscular theory
- (2) Huygen's wave theory
- (3) Maxwell's theory of light
- (4) Plank's theory

80. If 'a' and 'b' are the amplitudes of two interfering waves then for destructive interference the amplitude 'R' is

(1) R = ab

(2) R = a/b

(3) R = a - b

(4) R = a + b



PART-C

It consists of 81-180 questions:

- 81. Bernoulli's equation for steady, frictionless, continuous flow states that the ____at all sections is same.
 - (1) total pressure

(2) total energy

(3) velocity

- (4) pressure head
- 82. Small pressure differences in liquids is measured using an
 - (1) U-tube manometer

(2) inclined tube manometer

(3) pitot tube

(4) variable area meter

- 83. Steady flow occurs when
 - (1) conditions change steadily with time
 - (2) conditions are the same at adjacent points at any instants
 - (3) conditions do not change with time at any point
 - (4) rate of velocity is constant
- 84. Reynolds number is ratio of
 - (1) viscous force to gravity force
 - (2) inertial force to viscous force
 - (3) viscous force to inertial force
 - (4) inertial force to gravity force
- 85. Cavitation occurs in a centrifugal pump when
 - (1) the suction pressure < vapour pressure of the liquid
 - (2) the suction pressure > vapour pressure of the liquid
 - (3) the suction pressure = vapour pressure
 - (4) the suction pressure = developed head

| 86. | The vacuum pump in any compressor vatmospheric and discharges at atmosphe | which takes suction at pressureric pressure. |
|-----|---|--|
| | (1) equal to | (2) above |
| | (3) below | (4) none of these |
| 87. | The pumps are commonly empl | oyed in industry for handling viscosity liquids. |
| | (1) gear pumps | (2) rotary pumps |
| | (3) plunger pumps | (4) centrifugal pumps |
| 88. | The head of centrifugal pumpc | continuously as the capacity is decreased. |
| | (1) decreases | (2) increases |
| | (3) becomes less | (4) becomes more |
| 89. | The removal of air from the suction line ar | nd pump casing is known as |
| | (1) air binding | (2) priming |
| | (3) NPSH | (4) suction head |
| 90. | In a single effect evaporator, the economy | y is |
| | (1) 1 | (2) < 1 |
| | (3) > 1 | (4) 0 |
| 91. | In extractive distillation, solvent is added to | o alter the of the mixture. |
| | (1) viscosity | (2) temperature |
| | (3) composition | (4) relative volatility |
| 92. | Molecular distillation is | The state of the s |
| | (1) high temperature distillation | (2) for heat sensitive materials |
| | (3) very low pressure distillation | (4) both (2) and (3) |

| | | 19 | Cł |
|-----|---|---|----|
| 93. | Mc Cabe Thiele- method uses | _ for material and energy balance. | |
| | (1) molar units | (2) weight fractions | |
| | (3) any type of units | (4) both (1) and (2) | |
| 94. | Steam distillation is used to separate | · | |
| | (1) azeotropes | | |
| | (2) high boiling substances from non-v | olatile impurities | |
| | (3) heat sensitive materials | | |
| | (4) mixtures with low relative volatility | | |
| 95. | Moisture in a substance exerting an equipolation liquid at the same temperature is called | ilibrium vapour pressure less than that of pur moisture. | е |
| | (1) bound | (2) unbound | |
| | (3) critical | (4) free | |
| 96. | 1 bar is almost equal to atm | osphere. | |
| | (1) 1 | (2) 10 | |
| | (3) 100 | (4) 1000 | |
| 97. | 1 gram mole of methane contains | | |
| | (1) 6.023 x 10 ²³ atoms of hydrogen | | |
| | (2) 4 gram atoms of hydrogen | | |
| | (3) 3.01×10^{23} molecules of methane | | |
| | (4) 3 grams of carbon | | |
| 98. | Number of gram equivalent of solute diss | solved in 1 liter of solution is called its | |
| | (1) normality | (2) molarity | |
| | (3) molality | (4) none of these | |
| | | | |



| 99. | The total volume occupied by a gaseou component volumes is | s mixture is equal to the sum of the pure |
|------|--|---|
| | (1) Dalton's law | (2) Amagot's law |
| | (3) Gas Lussac's law | (4) Avogadro's law |
| 100. | In a temperature recorder thermocouple is | s an example of |
| | (1) primary element | (2) secondary element |
| | (3) functioning element | (4) manipulated element |
| 101. | When damping co-efficient is unity the sys | stem is |
| | (1) over damped | (2) critically damped |
| | (3) under damped | (4) highly fluctuating |
| 102. | Response of a linear control system for a | change in set point is called |
| | (1) servo problem | (2) frequency response |
| | (3) regulator problem | (4) transient response |
| 103. | A negative gain margin expressed in deci | bels means |
| | (1) a stable system | (2) unstable system |
| | (3) critically damped system | (4) none of these |
| 104. | Desirable characteristic of an instrument is | S |
| | (1) high drift | (2) high fidelity |
| | (3) high measuring lag | (4) poor reproducibility |
| 105. | Continuous measurement of moisture comeasuring | ntent of paper in paper industry is done by |
| | (1) electric resistance through the paper | |
| | (2) thermal conductivity through the pape | r' |
| | (3) magnetic susceptibility | |
| | (4) both (2) and (3) | |
| | | |

SPACE FOR ROUGH WORK

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|------|---|------|--|
| 106. | Whether thermo chemical reaction occursenthalpy change is constant. This is the | | |
| | (1) Laplace-Lavoisier | (2) | Roult's |
| | (3) Hess's | (4) | Gibb's |
| 107. | The order of a reaction with respect to the g, in the experimentally determine | | • |
| | (1) nature | (2) | temperature |
| | (3) concentration | (4) | pressure |
| 108. | Buffers are the solutions which have the proof small amounts of acids or alkalies. | per | ty of changes in pH on addition |
| | (1) assisting | (2) | resisting |
| | (3) supporting | (4) | allowing |
| 109. | An aqueous or molten solution which allow and the solution decomposes into products | | The state of the s |
| | (1) neutrolysis | (2) | analysis |
| | (3) synthesis | (4) | electrolysis |
| 110. | The splitting of a compound into two fragmer fragment is known as | ents | , and carries one electron each , such a |
| | (1) free radical | (2) | cation |
| | (3) anion | (4) | carbanion |
| 111. | Cement clinker is reduced to fine size by | | |
| | (1) roll crusher | (2) | ball mill |
| | (3) tube mill | (4) | hammer mill |
| 112. | Mixer used rubber compounding is | | |
| | (1) mixer-extruder | (2) | banburry mixer |
| | (3) muller mixer | (4) | paddle mixer |
| | | | |



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|------|--|--|
| 113. | Size of ultra fine particles can be expresse | ed in terms of |
| | (1) centimeter | (2) screen size |
| | (3) micron | (4) surface area per unit mass |
| 114. | Mesh indicates the number of holes per | |
| | (1) square inch | (2) linear inch |
| | (3) square foot | (4) linear foot |
| 115. | Which of the following cannot be recommer | ended for transportation of abrasive material? |
| | (1) belt conveyor | (2) apron conveyor |
| | (3) flight conveyor | (4) chain conveyor |
| 116. | To handle smaller quantity of fluid at higher | er discharge pressure, use |
| | (1) reciprocating pump | (2) centrifugal pump |
| | (3) volute pump | (4) rotary vacuum pump |
| 117. | Viscosity of a gas varies with ten | mperature. |
| | (1) exponentially | (2) linearly |
| | (3) logarithmically | (4) both 1 and 2 |
| 118. | Pick out the Hagen-Poiseulli's equation | |
| | (1) $\Delta P/\rho = 4.f L/D .V^2/2g_c$ | (2) $\Delta P = 32\mu LV/g_c D^2$ |
| | (3) $\Delta P/L = 150(1-\epsilon)/\epsilon^3 \cdot \mu \cdot Vo^2/g_c^2 \cdot D_p$ | (4) $\Delta P = 32\mu L^2 V/g_c D$ |
| 119. | Bernoulli's equation accounts for | |
| | (1) various momentums | (2) various masses |
| | | |

(3) different forms of mechanical energy (4) different forms of pressure

| 120. | Enamels and paints are generally | |
|--|---|-------------------------|
| | (1) reheopectic | (2) pseudo-plastic |
| | (3) thixotropic | (4) dilatent |
| 121. | Multipass heat exchangers are used | · . |
| | (1) because of simplicity of fabrication | |
| | (2) for low heat load | |
| | (3) to obtain higher heat transfer co-efficient | ent of shorter tube |
| | (4) to reduce the pressure drop | |
| 122. | In counter flow compared to parallel flow_ | |
| | (1) LMTD is greater | |
| | (2) less surface area is required for a give | en heat transfer rate |
| | (3) both (1) and (2) | |
| | (4) more surface area is required for a given | ven heat transfer rate |
| 123. | Kg of liquid evaporated per hour in an eva | porator is defined as |
| | (1) capacity | (2) economy |
| | (3) steam load | (4) rate of evaporation |
| 124. Rate of heat transfer per unit is heat flux. | | eat flux. |
| | (1) area | (2) length |
| | (3) volume | (4) width |
| 125. | Fouling factor is | |
| | (1) a dimensionless quantity | |
| | (2) does not provide safety factor for des | ign |
| (3) accounts for additional resistances to heat flow | | heat flow |
| | (4) accounts for no resistance to heat train | nsfer |
| CDACE FOR DOUGH WORK | | |

| CH | 2 | 24 | |
|------|--|---|-------|
| 126. | Rotary dryers cannot handle m | materials. | |
| | (1) free flowing | (2) dry | |
| | (3) sticky | (4) granular | |
| 127. | Refractory bricks are usually dried in a _ | drier. | |
| | (1) tray | (2) tunnel | |
| | (3) conveyor | (4) festoon | |
| 128. | Moisture contained by a substance in e moisture. | excess of the equilibrium, moisture is co | alled |
| | (1) unbound | (2) free | |
| | (3) critical | (4) bound | |
| 129. | Detergent solution is dried to a powder in | in a | |
| | (1) spray drier | (2) tunnel drier | |
| | (3) tray drier | (4) rotary drier | |
| 130. | pH value of an alkaline solution is | | |
| | (1) 7 | (2) > 7 | |
| | (3) < 7 | (4) constant over wide range | |
| 131. | On-off control is a special case of | control. | |
| | (1) PID | (2) PD | |
| | (3) PI | (4) P | |
| 132. | Thermocouple measures | | |
| | (1) current | (2) voltage | |
| | (3) flow rate | (4) temperature | |
| | | | |

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| 133. | In ammonia reactor pressure measuremen | nt is done by using |
|----------------------|---|---------------------------|
| | (1) U-tube manometer | (2) Bellow gauge |
| | (3) Bourdon gauge | (4) Pirani gauge |
| 134. | Analysis of natural gas is done by | |
| | (1) orsat apparatus | (2) spectrometer |
| | (3) chromatography | (4) emission spectrometer |
| 135. | Emf generated in a thermocouple is of the | order of |
| | (1) nano volts | (2) milli volts |
| | (3) micro volts | (4) macro volts |
| 136. | Point velocity is measured by using | |
| | (1) orifice meter | (2) venturimeter |
| | (3) pitot tube | (4) velocity meter |
| 137. | Fluid used in hydraulic controller is | |
| | (1) water | (2) air |
| | (3) steam | (4) oil |
| 138. | 38. Pressure of 0.01 Psi (absolute) can be measured by | |
| | (1) ionization gauge | (2) pirani gauge |
| | (3) Mcloid gauge | (4) bourdon gauge |
| 139. | 39. Flapper nozzle is a controller. | |
| | (1) pneumatic | (2) electronic |
| | (3) hydraulic | (4) electric |
| 140. | D. A compound which when dissolved in water yields hydroxyl ions is | |
| | (1) an acid | (2) an alkali |
| | (3) salt | (4) non aqueous solution |
| SPACE FOR ROUGH WORK | | |

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|---|---|---------------------------------------|---------------------------------|-------------------------|
| 141. | Cyclic organic compounds with conjugate hydrocarbons. | d do | ouble bonds are considered as _ | |
| | (1) aliphatic | (2) | aromatics | |
| | (3) unsaturated | (4) | saturated | |
| 142. | Trivial name of hydroxy benzene is | · · · · · · · · · · · · · · · · · · · | _ | |
| | (1) benzyl alcohol | (2) | toluene | |
| | (3) phenol | (4) | benzaldehyde | |
| 143. | Petroleum is chiefly | | | |
| | (1) alkanes | (2) | cycloalkanes | |
| | (3) aromatics | (4) | mixture of (1), (2) and (3) | |
| 144. Product formed by chlorination of methane in the presence of light at HCI. | | the presence of light are | and | |
| | (1) CH ₃ CI | (2) | CH ₂ Cl ₂ | |
| | (3) CHCl ₃ | (4) | All of these | |
| 145. | Derivatives of benzene are commonly call | ed _ | | |
| | (1) phenois | (2) | esters | |
| | (3) carotenoids | (4) | benzenoids | |
| 146. | Fluid energy mill comes in the category of | | | |
| | (1) grinder | (2) | crusher | |
| | (3) cutting machine | (4) | ultrafine grinder | |
| 147. Which of the following gives the crushing energy required to create new | | gy required to create new surfac | e ? | |
| | (1) Taggarts rule | (2) | Fick's law | |
| | (3) Rittinger's law | (4) | None of these | |
| | | | | |

| 148. | For transportation of pasty material, one will use | |
|----------------------|--|----------------------|
| | (1) apron conveyor | (2) belt conveyor |
| | (3) screw conveyor | (4) bucket conveyor |
| 149. | Ultra fine grinders separate principally by | |
| | (1) slow compression | (2) impact |
| | (3) attrition | (4) cutting action |
| 150. | Newton's law of viscosity relates | |
| | (1) shear stress and velocity | |
| | (2) velocity gradient and pressure intensi | ty |
| | (3) shear stress and rate of angular defor | mation |
| | (4) pressure gradient and rate of angular | deformation |
| 151. | Which of the following is a dimensionless | parameter ? |
| | (1) angular velocity | (2) specific weight |
| | (3) kinematic viscosity | (4) Reynold's number |
| 152. | 52. The maximum pressure difference for transportation of gases is produced by | |
| | (1) vacuum pumps | (2) blowers |
| | (3) fans | (4) compressors |
| 153. | 53. The co-efficient of discharge of an orifice-meter is a function of | |
| | (1) Reynold's number at the orifice | |
| | (2) ratio of orifice to pipe diameter | |
| | (3) both (1) and (2) | |
| | (4) none of the above | |
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(3) less than the sum of vapour pressures of the components

(2) less than that computed for ideality

(4) equal to that computed for ideality

| | | 3 | |
|------|--|---------------------------|--|
| 160. | 160. For a binary mixture with low relative volatility | | |
| | (1) use steam distillation | | |
| | (2) use molecular distillation | | |
| | (3) use high pressure distillation | | |
| | (4) an azeotrope may be formed during of | distillation | |
| 161. | 161. The quantity of the heat required to evaporate 1 kg of a saturated liquid is called | | |
| | (1) specific heat | (2) 1 K cal | |
| | (3) 1 cal | (4) latent heat | |
| 162. | Dry air is a mixture of | | |
| | (1) vapors | (2) gases | |
| | (3) both (1) and (2) | (4) neither (1) nor (2) | |
| 163. | 33. Mass number of an atom is the sum of the numbers of | | |
| | (1) neutrons and protons | (2) protons and electrons | |
| | (3) neutrons and electrons | (4) both (1) and (2) | |
| 164. | 64. Molality is defined as the number of gram moles of solute per of solvent. | | |
| | (1) litre | (2) kg | |
| | (3) gram mole | (4) gram | |
| 165. | 65. The total number of atoms in 8.5 gm of NH_3 is x 10^{23} . | | |
| | (1) 9.03 | (2) 3.01 | |
| | (3) 1.204 | (4) 6.02 | |
| 166. | 166. To measure the temperature of a red hot object we use | | |
| | (1) thermistor | (2) thermometer | |
| | (3) radiation pyrometer | (4) optical pyrometer | |
| | | | |

| | | i seggi lien en | |
|------|--|-------------------------------------|--|
| 167. | Notches are used to measure fluid flow ra | ates in | |
| | (1) closed channels | (2) open channels | |
| | (3) vertical pipe lines | (4) horizontal pipe lines | |
| 168. | 88. Flow rate of a liquid containing heavy solids can be best measured by | | |
| | (1) segmental orifice | (2) concentric orifice | |
| | (3) rotameter | (4) eccentric orifice | |
| 169. | The corrector plate in a vapor pressure the | ermometer is element. | |
| | (1) primary | (2) secondary | |
| | (3) functioning | (4) manipulating | |
| 170. | Response of a system to a sinusoidal input | ut is called | |
| | (1) impulse response | (2) unit step response | |
| | (3) frequency response | (4) both (1) and (2) | |
| 171. | A mole is the amount of substance which | contains atoms/molecule/ions. | |
| | (1) 6.023 x 10 ²³ | (2) 6.022 x 10 ²² | |
| | (3) 6.23×10^{22} | (4) 6.23×10^{23} | |
| 172. | 72. A gas law which relates the volume of a gas to the number of molecules of the gas is | | |
| | (1) Avogadro's law | (2) Boyle's law | |
| | (3) Charle's law | (4) Gay – Lussac's law | |
| 173. | The process due to which an acid complet | tely reacts with a base is known as | |
| | (1) oxidation | (2) reduction | |
| | (3) neutralization | (4) combustion | |
| | | | |

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| 174. | A solution whose pH below 7 is | | |
| | (1) blood | (2) milk | |
| | (3) lime water | (4) ammonia solution | |
| 175. | . Amount of heat absorbed or liberated when diamond is converted into graphite t the enthalpy of is considered. | | |
| | (1) formation | (2) solution | |
| | (3) combustion | (4) transition | |
| 176. | 76. The characteristic functional group for addition reaction is | | |
| | (1) single bond | (2) double bond | |
| | (3) triple bond | (4) multiple bond | |
| | 77. Two and more organic compounds having same molecular formula but different structure due to presences of same substituents are isomers. | | |
| | (1) chain | (2) functional | |
| | (3) optical | (4) position | |
| 178. | 178. For obtaining an instant temperature is used along with oxygen. | | |
| | (1) methane | (2) ethane | |
| | (3) ethyne | (4) benzene | |
| 179. | 179. When acetylene gas is passed through a red hot tube is produced. | | |
| | (1) methane | (2) ethane | |
| | (3) ethyne | (4) benzene | |
| 180. | 30. For coarse reduction of hard solids , use | | |
| | (1) impact | (2) attrition | |
| | (3) compression | (4) cutting | |
| | | | |
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