

C	LASS:IX		Unified Council
3.	The polygon(s) formed by is/are:	$\mathbf{y} \mathbf{y} = 3\mathbf{x} + 2, \mathbf{y}$	= -3x + 2 and y = -2
	(A) an equilateral triangle		
	(B) an isosceles triangle		
	(C) a right angled triangle		
	(D) a triangle and a trapezoi		
7.	The bottom, side and from known. The product of th	nt areas of a re hese areas is (ectangular box are equal to:
	(A) the volume of the box		
	(B) the square root of the vo	olume	14 3
	(C) twice the volume		1 53
	(D) the square of the volume	e 🦯	2 210
3.	ABCD is a parallelogram	. 'P' is a poin	t on AD such that
	$AP = \frac{1}{3} AD and 'Q' is a po$	int on BC suc	h that $CQ = \frac{1}{3} BC$.
	Then AQCP is a:		
	(A) parallelogram	(B) rhombus	100
	(C) rectangle	(D) square	
-	Two parallel chords of a are respectively, 5 cm an chords lie in a semi-circle chords is:	nd 12 cm in l	ength. If both the
	(A) 8.5 cm (B) 5 cm	(C) 3.5 cm	(D) 3 cm
0.	The degree measure of eac is an integer. Which of the of their measures?		
r	(A) 2:3:4 (B) 3:4:5	(C) 5 : 6 : 7	(D) 6 : 7 : 8
I.	Two adjacent sides of a par One of its diagonals is 20	rallelogram ar cm, then its	re 51 cm and 37 cm. area is:
2	(A) 412 cm^2 (B) 512 cm^2	(C) 612 cm^2	(D) 712 cm^2
2.	If the radius of a circle area is given by a number	is a rational r which is:	number, then its
	(A) rational	(B) irrational	

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(D) $\frac{3S}{4}$

(D) 62.5°

13. If
$$\left(a + \frac{1}{a}\right)^2 = 3$$
, then $a^3 + \frac{1}{a^3}$ equals:
(A) $\frac{10\sqrt{3}}{3}$ (B) $3\sqrt{3}$ (C) 0 (D) $6\sqrt{3}$

14. The solution set of the system of equations

$$\frac{4}{x}$$
 + 5y = 7, $\frac{3}{x}$ + 4y = 5 is:

$$(A)\left(\frac{1}{3},-1\right) \qquad (B)\left(\frac{-1}{3},1\right) \qquad (C)\left(\frac{-1}{3},-1\right) \qquad (D)\left(\frac{1}{3},1\right)$$

15. If the arms of one angle are respectively parallel to the arms of another angle, then the two angles are:

(A) neither equal nor supplementary

- (B) not equal but supplementary
- (C) equal but not supplementary

(D) either equal or supplementary

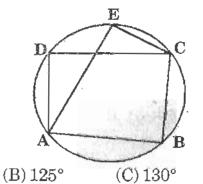
16. ABCD is a parallelogram of area 'S'. E and F are the midpoints of the sides AD and BC respectively. If G is any point on the line EF, then the area of \triangle AGB is equal to:

(A)
$$\frac{S}{2}$$
 (B) $\frac{S}{3}$ (C) $\frac{S}{4}$

17. Mid-points of the sides AB and AC of a \triangle ABC are (3, 5) and (-3, -3) respectively, then the length of the side BC is:

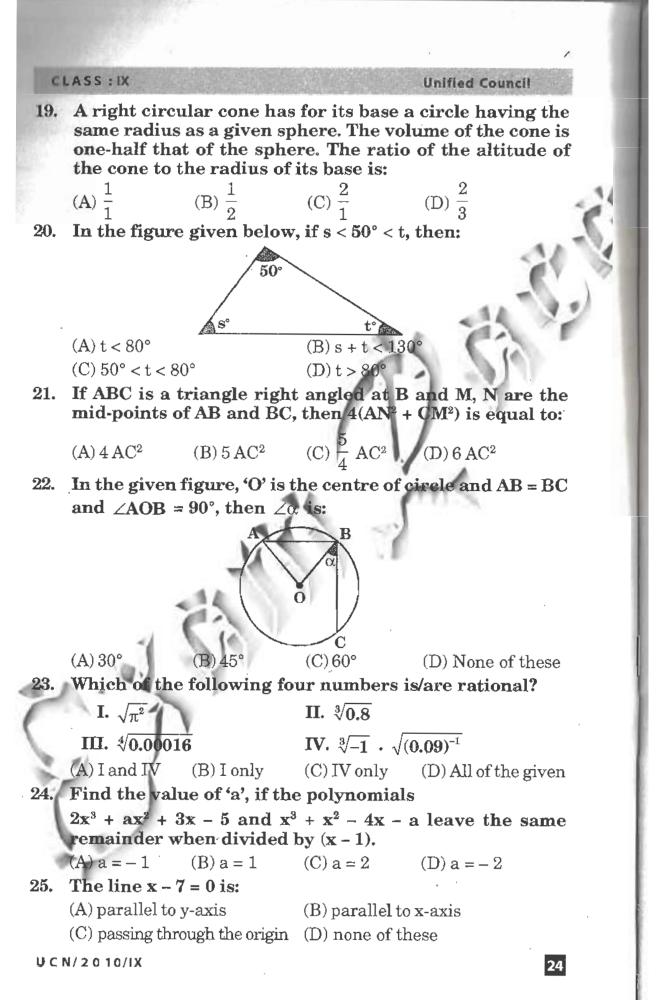
(A) 10 units (B) 15 units (C) 20 units (D) 30 units

18. Given a circle and a quadrilateral ABCD inscribed in it as shown. If $\angle B = 125^\circ$, then $\angle E$ is equal to:





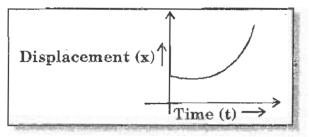
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С	ASS : IX	PHYSICS				
26.		nrticle, what does decrease i ime mean ?	n			
	(A) The particle is move	ing towards its initial position.				
	(B) The acceleration of	the particle is approaching zero.				
	(C) The particle is mov	ing with uniform speed.				
	(D) The particle is mov	ing with uniform motion.				
27. 28.	of 4 bullets per secon force must be applie (A) 20 N (B) 12.5 Assertion : The buoya	ass 12 kg fires 25 g bullets at the rat nd with a velocity of 500 m s ⁻¹ . What ed to the gun to hold it in position (5 N (C) 50 N (D) 75 N ant force of water on a submerge eater than on a steel cube of equa	at ? d			
	of the liquid displace					
	 (A) Both assertion and reason are true and reason is the correct explanation of assertion. 					
	 (B) Both assertion and reason are true, but reason is not the correct explanation of assertion. (C) A structure in false 					
	(C) Assertion is true, reason is false.					
29.	(D) Assertion is false, I Which of the follow tional force existing	ving statements regarding gravits between two bodies is TRUE ?	a-			
~	(A) First body exerts attractive force on second body while second body exerts repulsive force on first body.					
1	(B) The gravitational fo	orce is zero when they are kept in vacuur	n.			
7	(C) Force exerted by first body on second is not equal to the force exerted by second body on first.					
1	force exerted by se					
30.	A wave source proc the time period of t		d			
	(A) 1 second	(B) 0.01 second				
	(C) 10 second	(D) 0.1 second	,			
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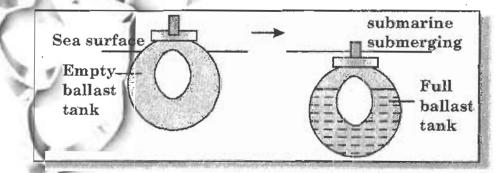
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31. The graph below represents a particle moving along a straight line, such that its displacement 'x' varies with time 't' as $x = at^2 + bt + c$ where a, b, c are constant.



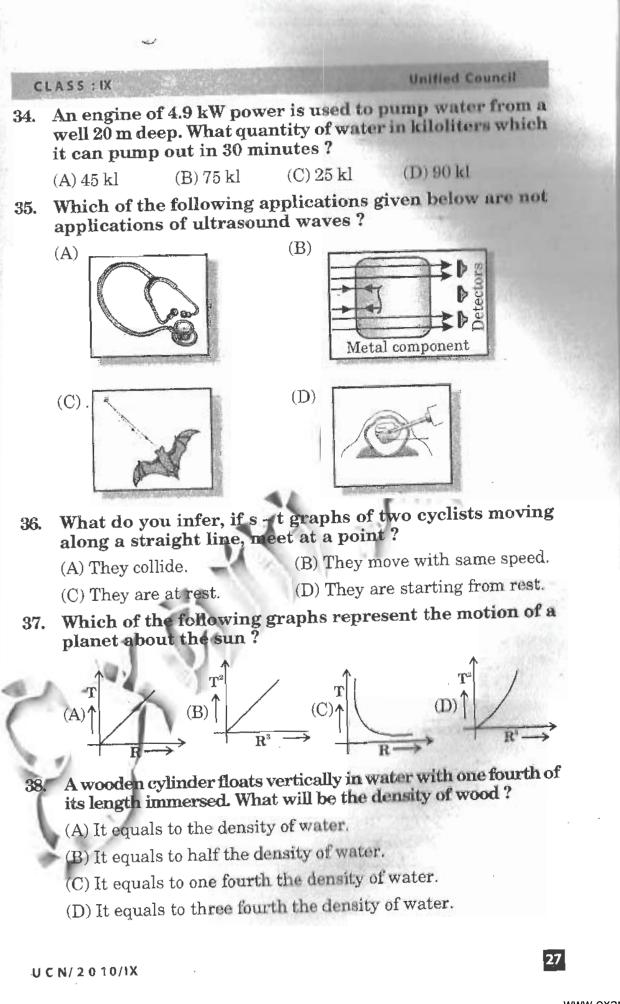
Which of the following statements regarding the above graph is false ?

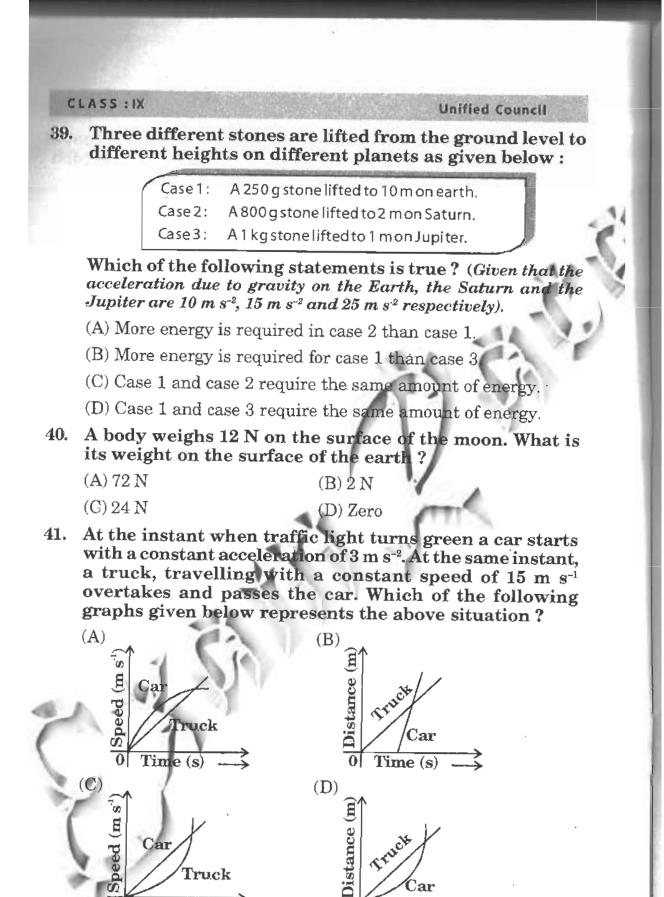
- (A) The body moves with an uniform acceleration of 2a.
- (B) The final velocity of the body is (b + 2at)
- (C) The initial displacement of the body is c.
- (D) The initial velocity of the body is (b + c)
- 32. Identify which of the statements given below are applicable for an object to move in a circle ?
 - (A) Object must continually slow down.
 - (B) Object must accelerate.
 - (C) Object must be acted on by balanced forces.
 - (D) Object must move with uniform velocity.
- 33. Figures (a) and (b) show that a submarine can either sink or float even though the upthrust acting on it is the same. What is the relationship between the weight of submarine and the upthrust when the submarine sinks in water ?



(A) Weight of submarine < upthrust.

- (B) Weight of submarine = upthrust.
- (C) Weight of submarine > upthrust.
- (D) Weight of submarine \leq upthrust.





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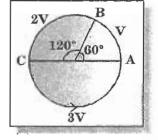
Time (s)

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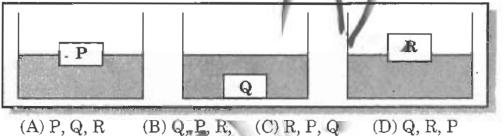
Time (s)

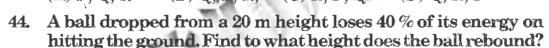
42. Sohail cycles on a circular track in anticlockwise direction as shown in figure. He travels with a speed 'V' to cover the path AB, next with speed '2V' from B to C and with a speed of '3V' from C to A. What is his average speed for the total journey?





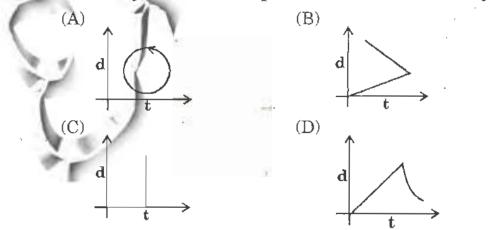
43. Three objects are introduced into the same liquid as shown below. Which of the following is in descending order of relative density of the objects?



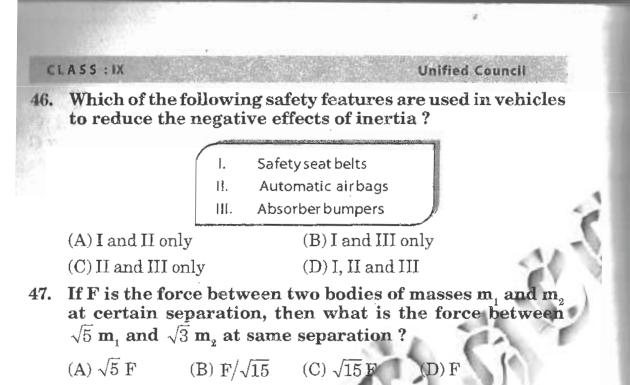


(A) 28 m (B) 8 m (C) 12 m (D) 20 m

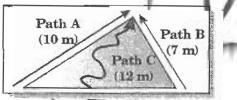
45. The motion of an object is plotted by four distance – time graphs. Which of the following graphs given below correctly describe the possible motion of the object?



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48. There are three paths leading to the top of the hill as shown. Assuming that the friction of the ground is negligible, which of the following statements is true ?

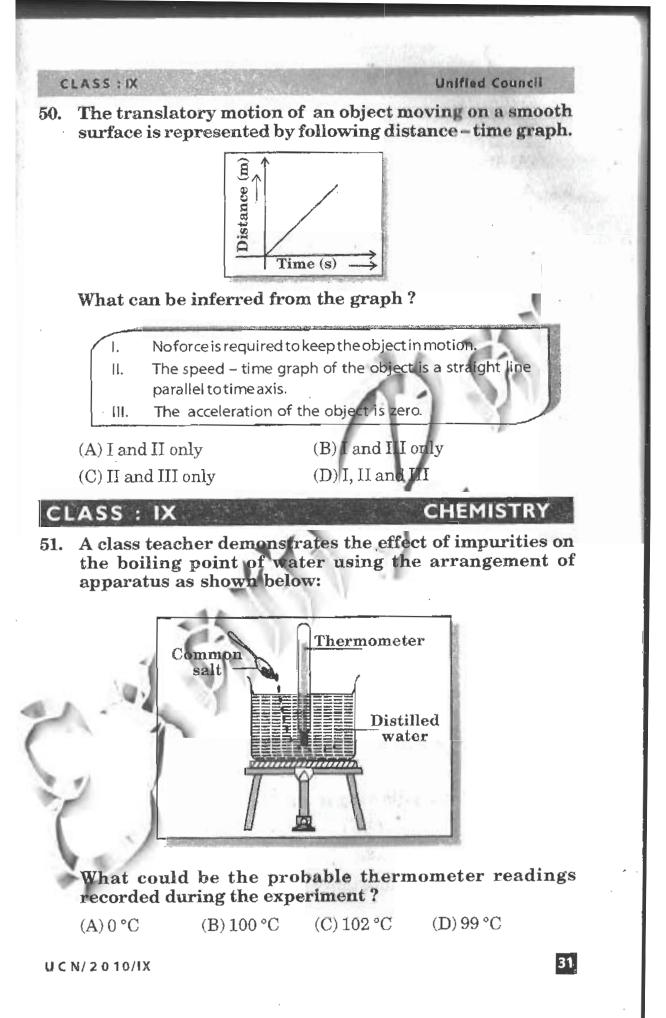


- (A) Path C requires the more energy to reach the top.
- (B) Path B requires the least energy to reach the top.
- (C) Path B requires more energy than path A to reach the top.
- (D) All the three paths require the same amount of energy to reach the top.
- 49. Observe the diagram carefully. A car braked suddenly near a cliff. Explain the motion of the driver.
 - (A) The driver remain stood up to look at how near his car was to the edge of the cliff.
 - (B) The driver was thrown forward when the car stopped, as he still has tendency to remain in motion.



- (C) The driver tried to lean forward to balance himself.
- (D) The driver was moved in the backward direction due to inertia of motion.

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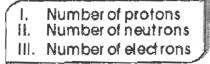


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52.	What is the smallest consists chemical identity ?	stituent of a matter that retains				
	(A) Atom (B) Molecule	e (C) Ion (D) Radical				
53.	At what conditions a ga into liquid ?	s can be cooled and converted				
	(A) At its critical temperatu	re by decreasing pressure on it.				
	(B) Above its critical temperature by increasing pressure on it.					
	(C) Above its critical temperature by decreasing pressure on it.					
	(D) Below its critical temper	rature by increasing pressure on it.				
54.	Which of the following pairs of colloidal solutions have dispersed phase as liquid and dispersing medium as gas?					
	(A) Fog, mist	(B) Butter, milk				
	(C) Fog, smoke	(D) Smoke, foam				
55.	The table below shows the melting and boiling points of substances P, Q, R and S:					
	Substance Melting poin	it (°C) Boiling point (°C)				
	P -210	-196				
	R 250	360 800				
	S -8	60				
		n a liquid state over a wide range				
	(A) P (B) Q	(C) R $(D) S$				
56.	Which of the following is	s a true solution ?				
	(A) Copper in gold	(B) Sulphur in water				
ς.	(C) Milk	(D) KCl in sulphur dioxide				
57.	Which of the following is	s a physical change ?				
1	(A) Magnetisation of iron	(B) Curdling of milk				
1	(C) Burning of a candle	(D) Cooking of food				
58.	Identify a pure substanc	e from the following ?				
	(A) Steel	(B) Magnalium				
	(C) Ammonia	(D) Gun powder				
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59.	Identify the pair of substances having same formula uni mass ?
	(A) Calcium chloride, potassium carbonate.
	(B) Calcium oxide, hydrochloric acid.
	(C) Carbon monoxide, ammonia.
	(D) Carbon dioxide, nitrous oxide.
60.	Which of the following ions is not divalent?
	(A) Sulphate (B) Nitrate (C) Hydrogen phosphate (D) Carbonate
51 .	Find the number of moles of sodium nitrate which contains 1.5 moles of oxygen atoms.
	(A) 0.5 (B) 1.5 (C) 2.0 (D) 1.0
	P and Q. If the ratio of valency of element X in P to elemen X in Q is 3 : 5 respectively. What could be the probable compounds P and Q?
	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
	(D) $H_{2O} = H_{2O_2}$ (D) $CH_4 = C_3H_8$
3.	
:3.	(D) CH ₄ C ₃ H ₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in
3. 4.	 (D) CH₄ C₃H₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in 245 g of KClO₃? (A) 2:2:3 (B) 1:1:3 (C) 39:36:48 (D) 1:2:6
4.	 (D) CH₄ C₃H₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in 245 g of KClO₃? (A) 2:2:3 (B) 1:1:3 (C) 39:36:48 (D) 1:2:6 On which of the following factors, molecular arrangement
4.	 (D) CH₄ C₃H₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in 245 g of KClO₃? (A) 2:2:3 (B) 1:1:3 (C) 39:36:48 (D) 1:2:6 On which of the following factors, molecular arrangement of a substance depends ?
4.	 (D) CH₄ C₃H₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in 245 g of KClO₃? (A) 2:2:3 (B) 1:1:3 (C) 39:36:48 (D) 1:2:6 On which of the following factors, molecular arrangement of a substance depends ? (A) Temperature and pressure.
4.	 (D) CH₄ C₃H₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in 245 g of KClO₃? (A) 2:2:3 (B) 1:1:3 (C) 39:36:48 (D) 1:2:6 On which of the following factors, molecular arrangement of a substance depends ? (A) Temperature and pressure. (B) Concentration and temperature.
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4.	 (D) CH₄ C₃H₈ What is the ratio of number of atoms of respective elements potassium, chlorine and oxygen present in 245 g of KClO₃? (A) 2:2:3 (B) 1:1:3 (C) 39:36:48 (D) 1:2:6 On which of the following factors, molecular arrangement of a substance depends? (A) Temperature and pressure. (B) Concentration and temperature. (C) Temperature, pressure and concentration. (D) Volume and pressure. Carbon and oxygen react to produce carbon dioxide. What is the weight of oxygen required to convert 1.5 g of carbon

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66. Two atoms ${}_{92}P^{235}$ and ${}_{92}P^{238}$ are similar in terms of :



(A) I only (B) I and II only (C) I and III only (D) I, II and III

67. Which of the following conclusions cannot be drawn on the basis of Rutherford's atomic model ?

- (A) Total mass of the atom is concentrated at the centre of atom.
- (B) Nucleus is located inside the atom containing positively charged particles.
- (C) Most of the atom is empty in space

(D) Electrons revolve around the nucleus in stationary circular orbits.

68. Assertion: ²⁰Ne and ²²Ne are isotopes.

Reason: Noble gases do not exist as isotopes as they are inert.

- (A) Both assertion and reason are true and reason is the correct explanation of assertion.
- (B) Both assertion and reason are true, but reason is not the correct explanation of assertion.
- (C) Assertion is true, reason is false.
- (D) Assertion is false, reason is true.
- 69. What is the ratio of number of neutrons present in potassium atom and magnesium atoms with mass numbers 39 and 24?

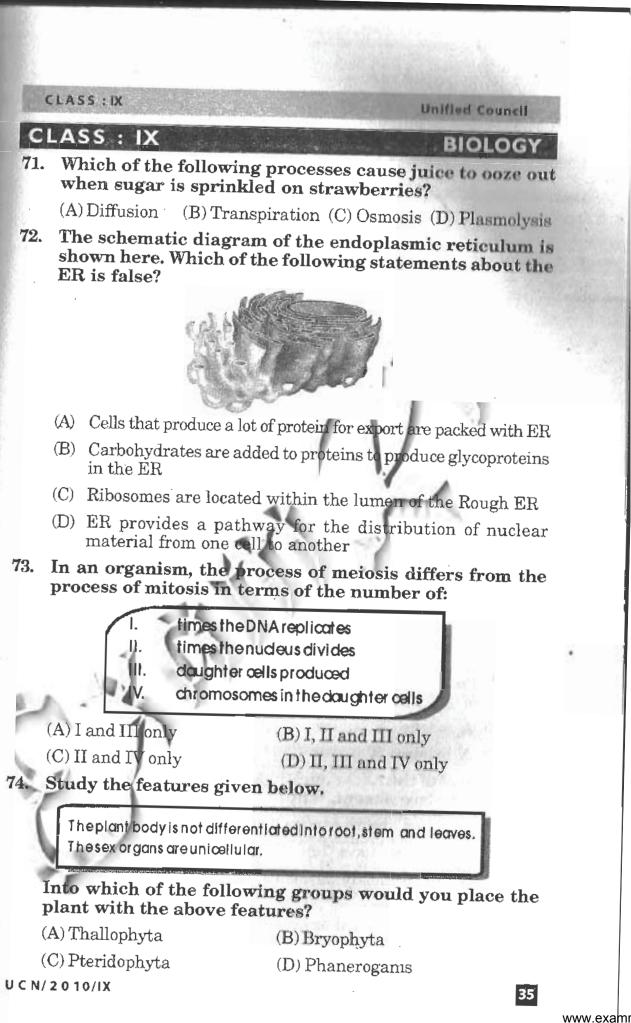
 $(A) 19: 12 \qquad (B) 5: 3 \qquad (C) 5: 6 \qquad (D) 4: 3$

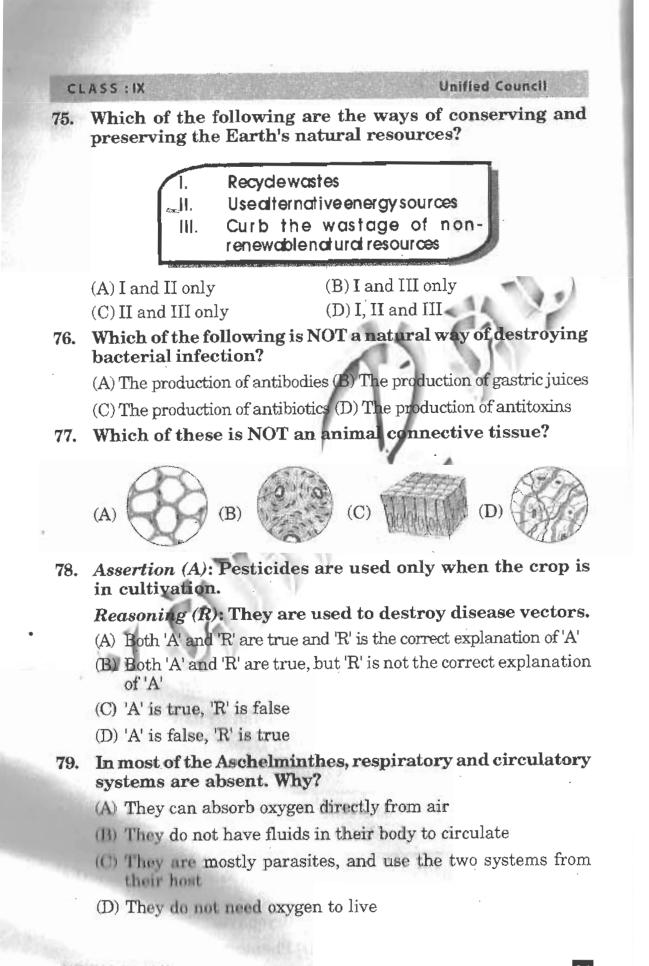
The isotopes of an element X are $_{z}X^{A}$, $_{z}X^{A+1}$, $_{z}X^{A+2}$ respectively. What is the ratio of number of nucleons in these respective isotopes ?

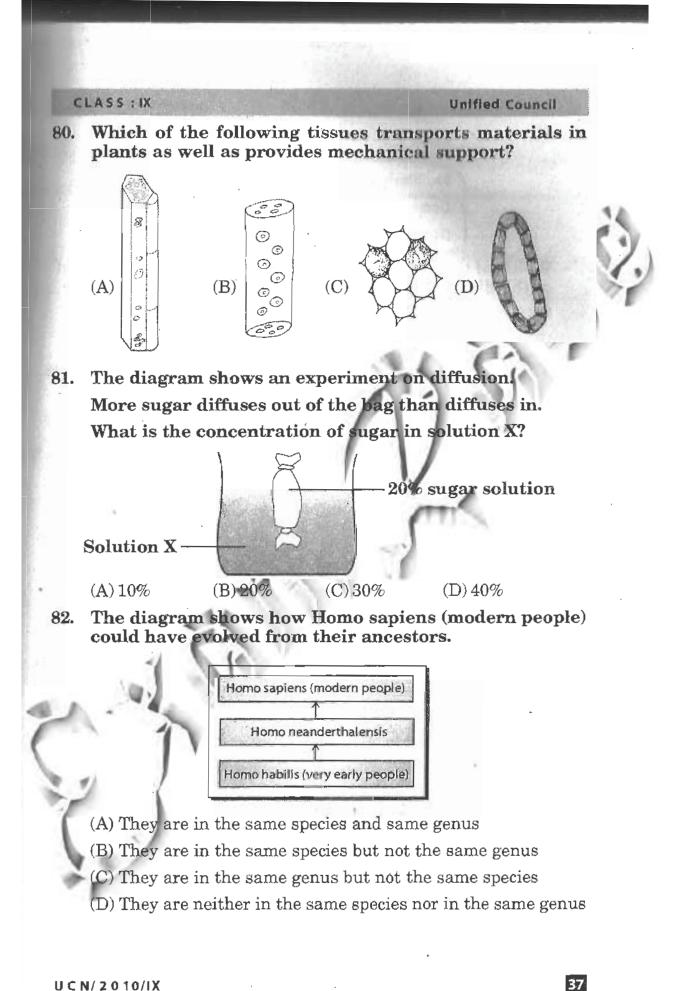
(A) A: (A + 1): (A + 2) (B) 1: 1: 1

(C) (A-Z): ((A+1)-Z): ((A+2)-Z) (D) 1:2:3

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83. The following symptoms were observed in a patient.

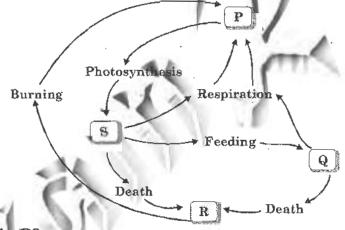
Swollen lymph glands

Decreased thrombocyte count

Decreased immunity and weight loss

Which of the following could be the reason for the spread of the disease with the above symptoms?

- (A) Contaminated food and water
- (B) Inhaling injected droplets released by injected person
- (C) Transfusion of blood from infected person to healthy person
- (D) Through infected mosquito bite
- 84. The figure given below shows a part of carbon cycle. P, Q, R and S are carbon compounds.



What is P?

(A) Carbon compounds in animals (B) Carbon compounds in plants

(C) Carbon dioxide in the air (D) Coal and oil

85. Neha observed the following observations while looking into a permanent slide.

Cells are long and cylindrical Light and dark bands are present giving striated appearance

It would be a slide of:

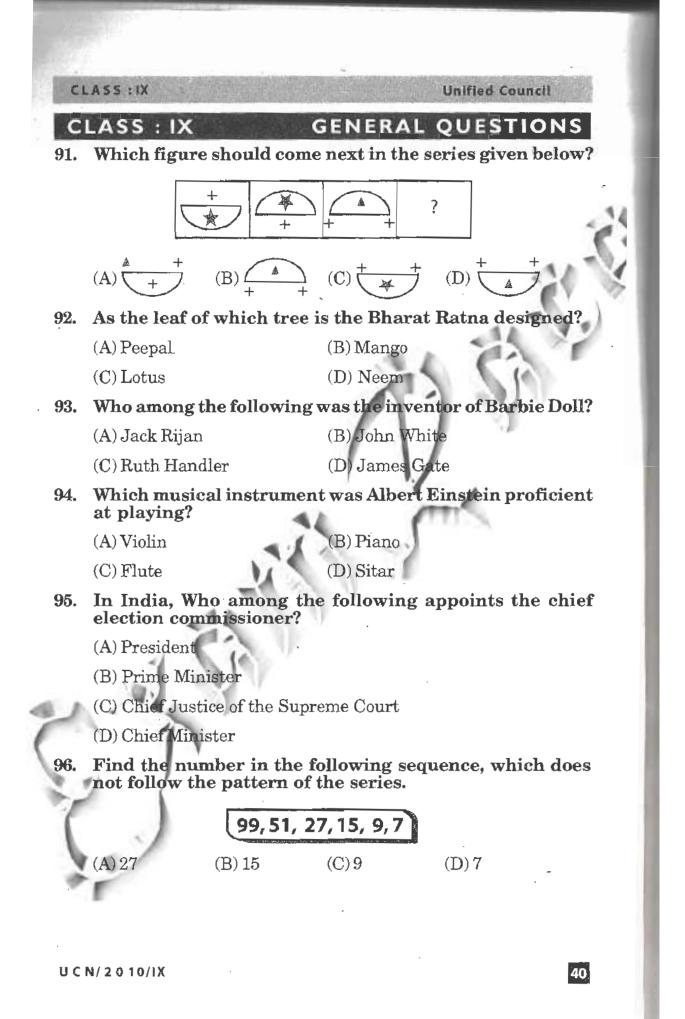
(A) skeletal muscle fibre(C) ligament

(B) smooth muscle fibre(D) visceral muscle

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6. Which of the following is a result of n	atural selection?
(A) Seedless grapes	A STATE OF STREET
(B) Insecticide - resistant mosquitoes	
(C) Disease - resistant crop plants	
(D) Cattle for high milk yield	
7. Application of nitrogenous manure to	a plant causes:
(A) early flowering	52
(B) growth retardation due to toxicity of an	nmonia
(C) early fruiting	
(D) vigorous vegetative growth	1. PA
The chromosomes which are present i male human are:	n the neurons of
(A) $44 + XX$ (B) $22 + Y$	1
(C) 22 + X $(D) 44 + XY$	1 3
Which of the part labelled in the figure	e riven above act
as a nerve insulator besides providing	
dendrite and axon?	mark
P	mm - 5
YK Q R S	Æ
A CONTRACTOR OF THE OFFICE OFF	E.
	for .
The state	
$(A) P \qquad (B) Q$	
$(C) R \qquad (D) S$	
Why are houseflies NOT considered to be	biological vectors
(A) They do not spread diseases	
(B) They do not transmit the disease-causin	ng organism directl
into our body	
(C) They are parasites and not vectors	
(D) None of these	
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Which of the follo given below?	owing should come next in the serie
ABDG,	GDFI, EFHK, ?
(A) GHJM	(B) HILN
(C) HIMN	(D) HIKM
Which famous arch at New Delhi?	itect designed the Baha'i Lotus Templ
(A) Le Corbusier	(B) Edwin Lutyens
(C) Laurie Baker	(D) Fariborz Sahba
In which year did	India become a member of the UN?
(A) 1945	(B) 1947
(C) 1951	(D) 1953
symbolized by the (A) Asia (C) America	lowing is NOT a geographic area circles on the Olympic flag? (B) Australia (D) Russia

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1. B	2. D	3. A	4. A	5. D	6. B	7. D	8. A
9. C	10. D	11. C	12. B	13. C	14. A	- 15. D	16. C
17. C	18. A	19. C	20. D	21. B	22. B	23. C	24. A
25. A	26. A	27. C	28. D	29. D	30. B	31. D	32. B
33. C	34. A	35. A	36. A	37. B	38. C	39. D	40. A
41. D	42. A	43. B	44 C	45. D	46. D	47. C	48. D
49. B	50. D	51. C	52. B	53. D	54. A	55. C	56. A
57. A	58. C	59. D	60. B	61. A	62. B	63. B	64. C
65. D	66. C	67. D	68. C	69. B	70. A	71. C	72. C
73. D	74. A	75. D	76. C	77. C	78. A	79. C	80. B
81. A	82. C	83. C	84. C	85. A	86. B	87. D	88. D
89. D	90. B	91. D	92. A	93. C	94. A	95. A	96. D
97. A	98. D	99. A	100. D		•	40 	
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