

APRIL - 1993

[RS 537]

SECOND B.Pharm. DEGREE EXAMINATION.

(Old Regulations)

Paper II -- ADVANCED PHARMACEUTICAL  
ORGANIC CHEMISTRY

Time : Three hours      Maximum : 100 marks

Answer SIX questions.

Question No.1 and 5 are compulsory.

Give chemical equations wherever necessary.

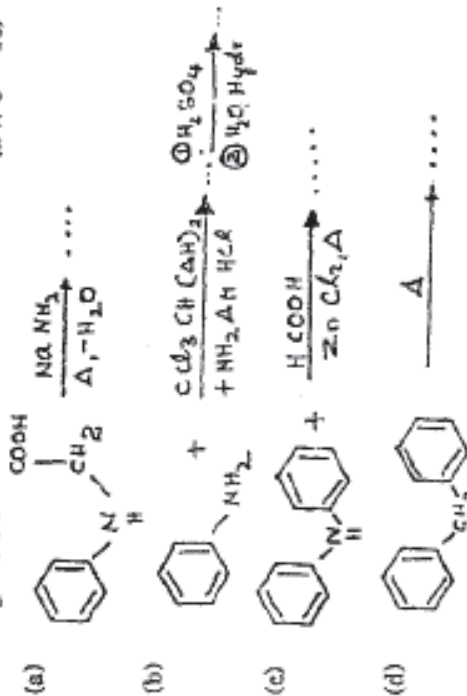
1. (a) Give a general classification of Heterocyclic compounds with examples, add a note on their nomenclature.  
(b) Explain the synthesis of quinoline and main reaction.  
(c) Name the compounds of quinoline of medical importance and uses. (8 + 6 + 4)
2. (a) Discuss the synthesis and chemistry of Anthracene, with important reactions.  
(b) Give the names and structures of medically important compounds of Anthracene. (12 + 4)
3. (a) Discuss the stereochemistry of (i) Diphenyl molecule and (ii) Nitrogen compounds.  
(b) Write a brief account of Hybridization of orbitals. (10 + 6)

4. Write short notes on :

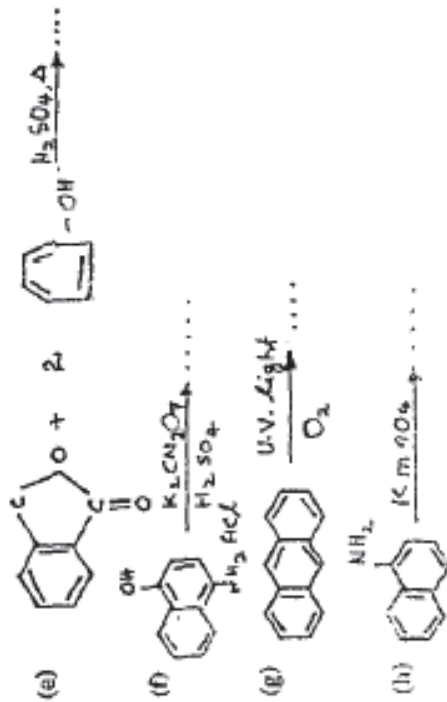
- (a) Optical isomerism and optical activity.
- (b) Geometrical isomerism.
- (c) Steric hindrance in organic compounds. (4 x 4 = 16)
- (d) Conformational analysis.

- 5. (a) What is inversion of configuration? Explain with examples using Walden Inversion reaction.
- (b) How is resolution of racemic mixtures achieved?
- (c) What is asymmetric synthesis? Give example. (6 x 3 = 18)

6. Explain the reactions, conditions and name the products in the following cases :

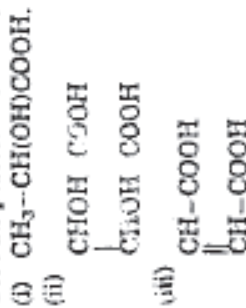


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- 7. (a) What is Keto-enol interconversion? How does it differ from resonance structures?
- (b) Why Mesotartaric acid does not show optical activity? How does it differ from dil. tartaric acid?
- (c) What is Diels Alder Reaction? Explain its mechanism with example. (4 + 6 + 6)

8. (a) Give all possible isomers of the following compounds :



- (b) Discuss the stereochemistry of cyclohexane derivatives. (8 x 2 = 16)

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[ R S 5 4 2 ]

SECOND B.Pharm. DEGREE EXAMINATION.

(New Regulations)

Paper II — ADVANCED PHARMACEUTICAL ORGANIC  
CHEMISTRY

Time : Three hours Maximum : 90 marks

Two and a half hours Sections A and B : 60 marks.  
for Sections A and B

Answer Sections A and B in separate answer books.  
Answer Section C in the answer sheet provided.

SECTION A — (2 × 15 = 30 marks)

Answer any TWO questions.

1. What do you understand by the term chirality ? Explain with suitable examples. What are the factors that make a molecule dissymmetric ? Give one method for asymmetric synthesis.
2. Outline two methods each for the synthesis of the following :  
(a) Indole  
(b) Imidazole  
(c) Isoquinoline.
3. What type of compounds exhibit geometric isomerism ? Explain how the configuration of geometric isomers can be established in the case of oximes, with suitable examples.
4. What is resolution ? Give different methods that are employed with suitable examples. Write a note on Walden inversion.

SECTION B — (6 × 5 = 30 marks)

Answer any SIX questions.

5. Give three reactions for the following :  
(a) Thiophen  
(b) Acridine  
(c) Pyridine.
6. Explain and compare the aromaticity of the following :  
(a) Pyridine  
(b) Pyrimidine  
(c) Pyrrole.
7. Discuss in brief the conformational isomers arising out of cyclic compounds.
8. Explain the following terms with suitable examples :  
(a) Absolute configuration.  
(b) R & S forms.  
(c) Epimers.
9. Give the preparation and reactions of  
(a) Triphenyl methane.  
(b) Phenothiazine  
(c) Thiazole.
10. Write a note on stereochemistry of biphenyls.
11. Write down the Fisher projection formula for lactic acid. Indicate the different isomers of tartaric acid by drawing structures. What is a diastereomer ? Define and explain.
12. Describe the various conformational isomers of n-butane and with energy diagram and Newman's projection formula explain their stabilities.
13. Give the synthesis of phenanthrene. Give examples of medicinally useful compounds belonging to this group.