

ANSWERS TO PAPER – I

PHYSICS

1. $v_T = 108 \text{ km/hr} = 30 \text{ m/s}, \quad f' = f \left[\frac{v + v_T}{v - v_T} \right] = 600 \left[\frac{330 + 30}{330 - 30} \right] \text{ Hz}, \quad f' = 720 \text{ Hz}$

∴ (c)

2. $D = \frac{m}{v} \Rightarrow \Delta D = \left(\frac{\Delta m}{m} + \frac{\Delta v}{v} \right) D, \quad \frac{\Delta v}{v} = \frac{3\Delta r}{r} = 0.06$

$\Delta D = \left(\frac{\Delta m}{m} + \frac{\Delta v}{v} \right) D = 0.1 + 0.06 \Rightarrow 2.25$

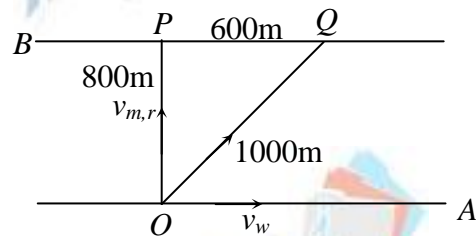
$\Rightarrow \Delta D = 0.4 \quad \therefore D = 0.5 \pm 0.4 \text{ g/cm}^3$

∴ (c)

3. Width of the river = $d = 800 \text{ m}$

∴ $v_w = \frac{600}{5 \times 60} = 2 \text{ m/s}$

∴ (b)



4. $y = \sin \left(\cos 0.02x \right)$, on the comparing this equation, with $y = 2A \sin \left(\omega t \right) \cos \left(kx \right)$

$\omega = 50, k = 0.02$

$v = \frac{\omega}{k} = \frac{50}{0.02} = 2500 \text{ m/s}$

∴ (c)

5. $\frac{\lambda}{4} = 30 \Rightarrow \lambda = 120, \quad v = f_x 120$

$\frac{\lambda}{2} = 59 \Rightarrow \lambda = 118, \quad v = f_y 118$

$\frac{f_x}{f_y} = \frac{59}{60}, \quad f_y - f_x = 10, \quad f_x = 590, \quad f_y = 600$

∴ (b)

6. $x = \sin \omega t, y = 4 \sin \left(\omega t - \frac{x}{2} \right)$ or $x = 4 \sin \omega t, y = 4 \cos \omega t$

So $x^2 + y^2 = 4^2$

Thus particle describes a circle of radius 4

∴ (a)

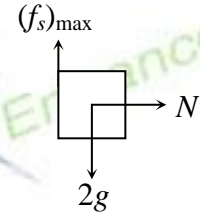
7. $N = 2a$, $f_{s \text{ max}} = 2g$
 $\mu \cdot 2a = 2g \quad \therefore a = \frac{g}{\mu} = 2g$

$T \cos \theta = mg$, $T \sin \theta = ma$

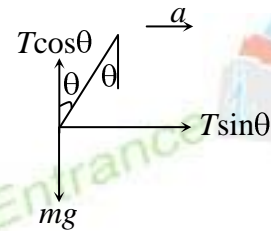
$a = g \tan \theta$, $\tan \theta = \frac{a}{g} = \frac{2g}{g} = 2$

$\theta = \tan^{-1} 2$

∴ (d)



F.B.D. of 2 kg



F.B.D. of 2 kg

8. Rest λ be the mass per unit length of the chain and x be the length of hanging part of the chain.

In case of limiting equilibrium of the chain, we have

$\lambda x g = \lambda (l - x) g \times \frac{1}{4} \Rightarrow x = \frac{l}{5}$

∴ (a)

9. (c)

10. Distance travelled by $B = \frac{10 \times (40 + 60)}{(60 + 300)} = \frac{10 \times 300}{360} = 8 \text{ m (approximately)}$

∴ (a)

11. Applying conservation of linear momentum, we have $mv \cos \theta = \frac{m}{2} v'$

$v' = 2v \cos \theta$

∴ Required distance = $\frac{v^2 \sin 2\theta}{2g} + v \cos \theta \times \frac{v \sin \theta}{g}$

∴ (b)

12. $v = \sqrt{\frac{2gh}{1 + \frac{k^2}{R^2}}}$

For the solid sphere, $I = \frac{2MR^2}{5} = MK^2 \Rightarrow \frac{k^2}{R^2} = \frac{2}{5}$

$$h = s \sin \theta \quad \therefore \quad v = \sqrt{\frac{5g s \sin \theta}{7}} = \sqrt{\frac{10}{7} g s \sin \theta}$$

\therefore (d)

13. From the principle of conservation mechanical energy, we have

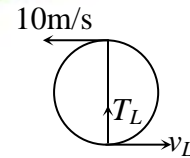
$$\frac{1}{2} m v_L^2 = m \times g \times 20 + \frac{1}{2} \times m \times 100$$

$$v_L^2 = 40g + 100 = 500$$

$$\text{Now, } T_L - mg = \frac{m v^2}{L}$$

$$T_L = \frac{m \times 500}{10} + mg = 600 \text{ N}$$

\therefore (d)



14. During the motion of the missile, its mechanical energy is always conserved.

$$ME = -\frac{GMm}{2R} = -\frac{mgR}{2} \quad \left(\because g = \frac{GM}{R^2} \right)$$

\therefore (c)

15. We know, $g' = g - R\omega^2 \cos^2 \lambda$

$$\text{For equator, } \lambda = 0 \quad \therefore \quad \omega^2 = \frac{g}{R}$$

$$\text{For north pole, } \lambda = 90^\circ \quad \therefore \quad g' = g$$

$$\therefore \quad mg = 100g$$

$$\text{Required weight, } mg' = mg - m \times R \times \frac{g}{R} \times \frac{1}{4} = \frac{3mg}{4} = 750 \text{ N}$$

\therefore (a)

16. Let v be the volume outside the water

For the equilibrium of the body, we have

$$(3 - v) \times 1 \times g = L^3 \times d \times g \Rightarrow v = 400 \text{ c.c.}$$

\therefore (a)

17. We have, $v = \sqrt{2 \times g \times \frac{h}{2}} = \sqrt{gh}$

$$\text{Also, } \sqrt{gh} \times \sqrt{\frac{h}{g}} = 135 \text{ cm, } h = 135 \text{ cm}$$

When the hole is made at a depth $\frac{h}{3}$ below the free surface, velocity of efflux

$$v' = \sqrt{2 \times g \times \frac{h}{3}} = \sqrt{2 \times 1000 \times \frac{135}{3}} = 300 \text{ cm/s}$$

\therefore volume of water coming out/second = $300 \times 9 = 2700 \text{ c/s}$

\therefore (a)

18. We have $\frac{\Delta v}{V} = 10^{-4} \quad \therefore \quad B = \frac{P}{\Delta v/v} = \frac{\rho gh}{\Delta v/v} = \frac{10^3 \times 10 \times 50}{10^{-4}} = 5 \times 10^9 \text{ N/m}^2$

\therefore (b)

19. $T = \frac{U}{A}$

$$U_I = 35 \times 10^{-4} (\pi r^2), \quad U_F = 35 \times 10^{-4} (\pi 4r^2)$$

$$\text{Work done} = U_F - U_I = 2.52 \times 10^{-1} \text{ J}$$

\therefore (b)

20. $-\frac{1}{4\pi\epsilon_0} \frac{q^2}{a} = -\frac{1}{4\pi\epsilon_0} \frac{2q^2}{a} + \frac{1}{2} m v_A^2 + \frac{1}{2} m v_B^2, \quad 2m v_A = m v_B \Rightarrow v_B = 2v_A$

$$\frac{q^2}{4\pi\epsilon_0 a} = \frac{1}{2} m v_A^2 + \frac{1}{2} m (2v_A)^2, \quad \frac{q^2}{4\pi\epsilon_0 a} = \frac{1}{2} m v_A^2 (1 + 4)$$

$$\Rightarrow \frac{q^2}{4\pi\epsilon_0 a} = 3m v_A^2 \Rightarrow 9 \times 10^9 \times \frac{q^2}{a} = 3m v_A^2 \Rightarrow 3 \times 10^9 \times \frac{q^2}{a} = m v_A^2$$

$$K_A = \frac{1}{2} m_A v_A^2 = \frac{1}{2} 2m v_A^2 = m v_A^2 = 3 \times 10^9 \times \frac{q^2}{a}$$

i.e. $K_A = 3 \times 10^9 \frac{q^2}{a}$

\therefore (c)

21. Net electrical field strength at L is = 15 N/C

\therefore force on $2C$ charge is = $15 \times 2 = 30 \text{ N}$

\therefore (c)

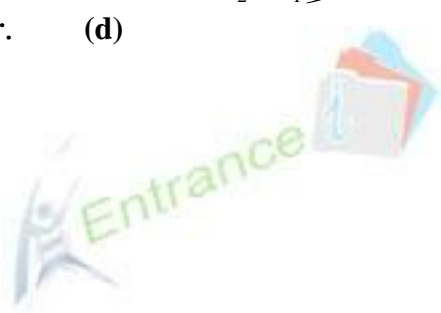
22. $\frac{Q}{6\epsilon_0} = \frac{10^6}{8.85}, \quad Q = \frac{6\epsilon_0 \times 10^6}{8.85}$

Distance between two charges = 15 m

$$F = \frac{1}{4\pi\epsilon_0} \frac{6\epsilon_0 \times 10^6}{8.85} \times \frac{1 \times 10^{-6}}{(5)^2} = 0.24 \text{ N}$$

\therefore (b)

23. $dv = -E \cdot dl = -10 \int_{x_1}^{x_2} dx = -10 \times 8 = -80$
 \therefore (d)



24. $R = 4 \Omega$ for maximum power transfer

$$P = I^2 R = 1 \times 4 = 4 \text{ watt}$$

\therefore (b)

25. Magnetic flux density due to a single wire is

$$B = \frac{\mu_0 I}{4\pi R} [\sin \alpha + \sin \beta] = \frac{10^{-7} \times 2}{\frac{1}{\sqrt{3}}} \left[2 \times \frac{\sqrt{3}}{2} \right] = 6 \times 10^{-7} \text{ T} = 0.6 \times 10^{-6} \text{ T}$$

$$B_{\text{net}} = 3B = 3 \times 0.6 \times 10^{-6} \text{ T} = 1.8 \times 10^{-6} \text{ T}$$

\therefore (a)

26. B at centre $= \frac{\mu_0 I}{2\pi r}$, $I = \frac{v}{r\theta}$
 $\rho \frac{A}{A}$

$$B \text{ at centre} = \frac{\mu_0 v_A}{2\pi \rho r^2}$$

i.e. independent of θ . So B at centre is zero

\therefore (d)

27. Acceleration of rod $\Rightarrow -\frac{BIL}{m} = -2 \text{ m/s}^2$

$$\text{Distance } d = \frac{v^2}{2a} = 1 \text{ m}$$

\therefore (d)

28. $F_e = F_B \Rightarrow \frac{eV}{l} = Bev \Rightarrow B = \frac{V}{lv}$

$$B = \frac{100}{10^{-2} \times 10^5} = 0.1 \text{ T outward}$$

\therefore (b)

29. $\varepsilon = B_A \tan 30^\circ lv = 2 \times 10^{-4} \times \frac{1}{\sqrt{3}} \times \frac{1}{2} \times 20\sqrt{3} \times 10^{-2} = 2 \times 10^{-5} \text{ volt} = 20 \mu\text{v}$

\therefore (a)

30. Energy $= \frac{1}{2} LI^2$

$$L = \text{Energy} \frac{2}{I^2}$$

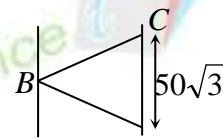
Dimensions of self inductance $= \text{ML}^2 \text{T}^{-2} \text{A}^{-2}$

\therefore (d)

31. $V^2 = (V_L - V_C)^2 + V_R^2$
 $|V_L - V_C| = 6V$
 $V_C = 8V$
 \therefore (d)

32. $\frac{1}{v} + \frac{1}{u} = \frac{1}{f}$, $u = 50$ cm and $f = 30$ cm
 Solving $v = 75$ cm
 \therefore (d)

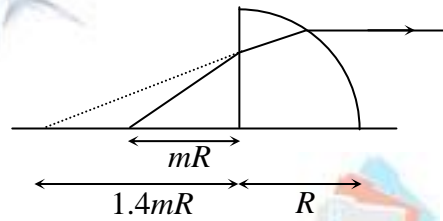
33. Such triangles (OBC) are around 20 in number each will have two new reflections except the first triangle in which we have three reflections.
 Hence 41 reflections in all
 \therefore (a)



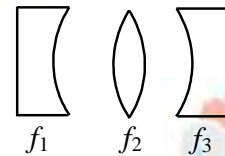
34. (c)

35. $D = (\mu_v - \mu_R) A + (\mu'_v - \mu'_R) \left(-\frac{\mu_y - 1}{\mu'_y - 1} A \right) = 0.04 \text{ } \overset{\circ}{9} + 0.04 \left(-\frac{1.5 - 1}{1.75 - 1} 9^\circ \right)$
 $= 0.04 \text{ } \overset{\circ}{9} \left[1 - \frac{0.50}{0.75} \right] = 0.04 \text{ } \overset{\circ}{3} = 0.12^\circ$
 \therefore (c)

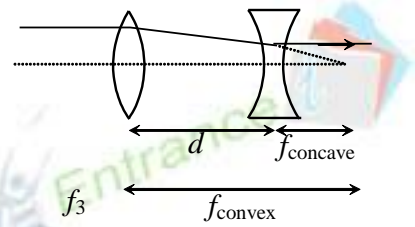
36. Refraction from IInd surface
 $\frac{1.4}{1.4mR + R} = \frac{1.4 - 1}{R}$
 $m = \frac{25}{14}$
 \therefore (a)



37. $\frac{1}{f_{eq}} = \frac{1}{f_1} + \frac{1}{f_2} + \frac{1}{f_3}$
 $\frac{1}{f_{eq}} = \frac{1 - \frac{3}{2}}{20} + \left(\frac{4}{3} - 1 \right) \left(\frac{1}{20} + \frac{1}{40} \right) + \frac{1 - \frac{3}{2}}{80} = -\frac{1}{80}$
 $f_{eq} = -80$ cm
 \therefore (b)



38. $d = 50 - 10 = 40 \text{ cm}$
 \therefore (d)



39. $m = \frac{f_0}{f_e} = 8 \Rightarrow f_0 = 8f_e, \quad f_0 + f_e = L = 45 \text{ cm}$
 $9f_e = 45 \text{ cm}, \quad f_e = 5 \text{ cm}, \quad f_0 = 40 \text{ cm}$
 \therefore (a)

40. $\frac{\lambda}{2} = t(\mu - 1) \Rightarrow t = \frac{\lambda}{2(\mu - 1)}$
 \therefore (b)

41. $\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$

Because image is erect and 4 times means lens is convex and object between pole and focus.

$$-\frac{1}{4u} + \frac{1}{u} = \frac{1}{20}$$

$$\frac{3}{4v} = \frac{1}{20}$$

$$u = 15 \text{ cm}$$

\therefore (a)

42. $\frac{1}{10} = \left(\frac{3}{2} - 1\right) \left(\frac{1}{R_1} + \frac{1}{R_2}\right) \quad \dots \text{ (i)}$

$$-\frac{1}{20} = \left(\frac{3}{\mu} - 1\right) \left(\frac{1}{R_1} + \frac{1}{R_2}\right) \quad \dots \text{ (ii)}$$

$$\mu = 2.0$$

\therefore (c)

43. $y = \frac{Dn\lambda}{d}, \quad n_1\lambda_1 = n_2\lambda_2, \quad \frac{n_1}{n_2} = \frac{\lambda_2}{\lambda_1} = \frac{6}{7}, \quad n_1 = 6, \quad n_2 = 7$

$$y = \frac{100 \times 6 \times 7000 \times 10^{-8}}{0.1} = 42 \times 10^{-2} \text{ cm} = 0.42 \text{ cm}$$

\therefore (c)

44. $E = 5 - \phi_x \Rightarrow \phi_x = 5 - E$, $(E - 2) = 6 - \phi_y \Rightarrow \phi_y = 8 - E$

$$\frac{P_1}{P_2} = \frac{\lambda_2}{\lambda_1}, \frac{E}{E-2} = \frac{P_1^2}{P_2^2} = 3 \Rightarrow E = 3, E = 6, E = 3$$

$$\phi_x = 2, \phi_y = 5, \quad \phi_x : \phi_y = 2 : 5$$

\therefore (b)

45. $T = \frac{T_1 T_2}{T_1 + T_2} = \frac{40}{3} \therefore$ in 80 years, $\frac{1}{64}$ of the initially substance will remain undecay.

$$\therefore \text{fraction which will decay is } 1 - \frac{1}{64} = \frac{63}{64}$$

\therefore (a)

46. $Q = \text{change in binding energy} = 4.6 \text{ MeV}$

\therefore (a)

47. $\frac{dT}{dt} = 10, \frac{dH}{dt} = C, \frac{\Delta H}{dt} = ms_1$

$$\therefore \frac{\Delta H}{dt} = ms_1 \times 10, \frac{\Delta H}{dt} = ms_2 \times \frac{20}{8} \quad \therefore \frac{s_1}{s_2} = \frac{1}{4}$$

$$\frac{mL_1}{mL_2} = \frac{C \times 6}{C \times 8} = \frac{L_1}{L_2} = \frac{3}{4}$$

\therefore (c)

48. $PV^\gamma = \text{constant}, P \propto T^{\frac{\gamma}{\gamma-1}} \therefore -\frac{r}{r-1} = 4 \Rightarrow r = 4/3$

$$\Delta H = 5 \times C_p \times 15 = 75 \times C_p = 300R$$

\therefore (a)

49. $\eta = \left[1 - \frac{(73^\circ + 27^\circ \text{C})}{273 + 227^\circ} \right] = \frac{2}{5}, \quad \frac{2}{5} = \frac{300}{Q}, \quad Q = 150 \times 5 = 750 \text{ J}$

\therefore (b)

50. $\frac{\Delta H}{\Delta U} = \frac{nC_p \Delta T}{nC_v \Delta T} = \gamma, \quad \Delta U = \frac{\Delta H}{\gamma} = \frac{700}{7} \times 5 = 500 \text{ J}$

\therefore (d)

51. $\frac{\text{K.E}}{\text{mole}} = \text{constant} \therefore 16 : 1$

\therefore (a)

$$52. \quad C_{v_{mix}} = \frac{\frac{1}{2} \times \frac{3}{2} R + \frac{1}{4} \times \frac{5}{2} R}{\frac{1}{2} \times \frac{1}{4}} = \frac{11}{6} R, \quad C_{p_{mix}} = \frac{11}{6} R + R = \frac{17}{6} R$$

$$\Delta Q = \frac{3}{4} \times \frac{17}{6} R \times 50 = 212.5 \text{ cal}$$

∴ (c)

$$53. \quad \frac{30}{10} = K \left(\frac{150}{2} - 15 \right) \Rightarrow 3 = K \times 60, \quad K = \frac{1}{20}$$

$$\frac{10}{t} = \frac{1}{20} \times (5 - 15) \Rightarrow t = \frac{200}{10} = 20 \text{ min}$$

∴ (a)

$$54. \quad \frac{dH}{dt} = \frac{KA}{l} (\theta_1 - \theta_2), \quad \frac{20}{5} = \frac{KA}{l} (\theta_1 - \theta_2)$$

$$\frac{160}{t} = 4 \frac{KA}{l} (\theta_1 - \theta_2) \quad \therefore \frac{160}{t} = 4 \times \frac{20}{5}$$

$$t = 10 \text{ min}$$

∴ (b)

$$55. \quad t \propto (\text{thickness})^2$$

$$\frac{t_1}{t_2} = \frac{4}{16} = \frac{1}{4}, \quad t_2 = 32 \text{ hour} \quad \therefore \Delta t = t_2 - t_1 = 32 - 8 = 24 \text{ hour}$$

∴ (c)

$$56. \quad KE = \frac{1}{2} m \omega^2 \left(A^2 - \frac{A^2}{4} \right) = \frac{3}{8} m \omega^2 A^2, \quad E = \frac{3}{8} m \omega^2 A^2, \quad P.E = \frac{1}{2} m \omega^2 A^2 = \frac{4}{3} E$$

∴ (c)

$$57. \quad A_1 = 4, \quad A_2 = 8 \sin \left(4\pi t + \frac{\pi}{3} \right) \quad \therefore \quad A_2 = 8, \quad \frac{A_1}{A_2} = \frac{1}{2}$$

∴ (b)

$$58. \quad y = A \cos \frac{2\pi}{6} t, \quad \frac{\sqrt{3}}{2} A = A \cos \frac{2\pi}{6} t, \quad t = \frac{4}{3} \text{ sec}$$

∴ (a)

$$59. \quad T = 2\pi \sqrt{\frac{R^3}{KQ}}, \quad T' = \frac{T}{2} = 2 \text{ sec}$$

∴ (a)

$$60. \quad f = 202, \quad f - 202 = 5, \quad f = 207, \quad \omega = 414 \pi$$

∴ (a)

CHEMISTRY

61. (b) With increase in temperature; the intermolecular attraction decreases.

62. (b) $C(s) + O_2(g) \longrightarrow CO_2(g)$; $\Delta H = -94.6 \text{ kcal}$... (i)

$CO(g) \longrightarrow C(s) + \frac{1}{2}O_2(g)$; $\Delta H = +26.4 \text{ kcal}$... (ii)

$CO(g) + \frac{1}{2} O_2(g) \longrightarrow CO_2(g)$; $\Delta H = -68.2$

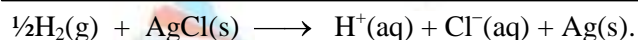
Equation (ii) is reverse of the equation of heat of formation of $CO(g)$.

63. (a) $\Delta H = \Delta U + \Delta n_g RT$

$$\Rightarrow \Delta H - \Delta U = \Delta n_g RT = \frac{-3 \times 8.314 \times 298}{1000} = -7.4327 \text{ kJ mol}^{-1}$$

64. (c) At anode: $\frac{1}{2}H_2(g) \longrightarrow H^+(aq) + e^-$

At cathode: $AgCl(s) + e^- \longrightarrow Ag(s) + Cl^-(aq)$



65. (c) Rate of reaction = $\frac{-d[SO_2]}{2dt} = \frac{-d[O_2]}{dt} = \frac{+d[SO_3]}{2dt}$

Hence, the rate of reaction will be $2.5 \times 10^{-4} \text{ mol/L sec}$.

66. (d) $B(OH)_3 + H_2O \rightleftharpoons B(OH)_4^- + H^+$. It's a monobasic acid.

67. (b) Because after giving one electron, it gains inert gas configuration.

68. (a)

69. (c)

71. (d)

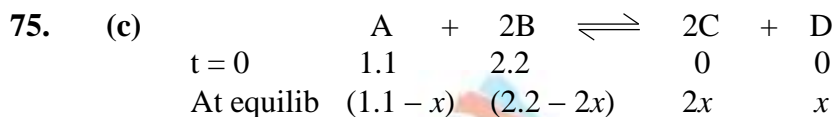
70. (b)

72. (b)

73. (a) Rubidium belongs to IA group and 5th period, hence its outermost configuration is $5s^2$, this implies that

$$n = 5, l = 0, m = 0, S = +\frac{1}{2}, -\frac{1}{2}$$

74. (c)



According to question [C] = 2x = 0.2

$$\Rightarrow x = 0.1$$

$$K_c = \frac{[C]^2 [D]}{[A][B]^2} = \frac{4x^2 \cdot x}{(1.1-x)(2.2-2x)^2} = \frac{x^3}{(1.1-x)^3} = 0.001.$$

76. (b) In I_3^- , $3x = -1$, or $x = -\frac{1}{3}$.

77. (c)

78. (b) Element belonging to groups 15 and 3rd period must have $5e^-$ in 3rd orbit, hence electronic configuration is $1s^2, 2s^2, 2p^6, 3s^2, 3p^3$.

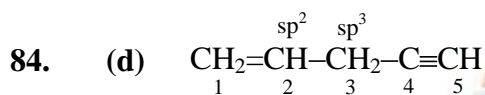
79. (c) In BCl_3 , Boron has only six electrons in outer most orbit.

80. (c)

81. (a)

82. (a)

83. (b) There is no sp^3 hybridised α -hydrogen in the compound $O=C_6H_4=O$.



85. (b) Compounds containing at least 3 α -hydrogens (in case of carbonyl compounds) or 3 β -hydrogens (in case of alcohols) will give iodoform test.

86. (d)

87. (c)

88. (a)

89. (c)

90. (c)

91. (d) For FCC, $4r = a\sqrt{2}$
 $\Rightarrow r = \frac{a\sqrt{2}}{4}$

$$\text{Closest distance} = 2r = \frac{a}{\sqrt{2}}.$$

92. (d) Rate = $k[\text{N}_2\text{O}_5]$

$$\therefore [\text{N}_2\text{O}_5] = \frac{2.4 \times 10^{-5}}{3 \times 10^{-5}} = \mathbf{0.8}.$$

93. (c) $W = 27$; $E = \frac{\text{atomic weight}}{\text{valency}} = \frac{27}{3} = 9$

$$Z = \frac{E}{F} = \frac{9}{96500} = \mathbf{9.326 \times 10^{-5}}.$$

94. (c) Milli equivalent of NaOH = $100 \times 0.5 = 50$

$$\text{Milli equivalent of HCl} = \frac{1}{5} \times 100 = 20$$

$$\text{Milli equivalent of H}_2\text{SO}_4 = \frac{1}{10} \times 100 = 10$$

$$\text{Total milli equivalent of acid} = 20 + 10 = 30$$

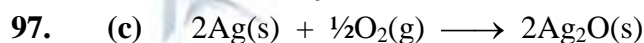
$$\text{Total milli equivalent of NaOH} = 50$$

$$\text{Left milli equivalent of NaOH} = 50 - 30 = 20 \text{ i.e., alkaline.}$$

95. (c)

96. (c) $E_n = \frac{-13.6 \times Z^2 \text{ eV}}{n^2}$ [Z = 1, n = 3]

$$= \frac{-13.6}{9} = \mathbf{-1.5 \text{ eV.}}$$



$$\Delta H = \Delta U + \Delta n_g RT$$

$$\Delta H = \Delta U - \frac{1}{2} RT$$

$$\Rightarrow \mathbf{\Delta H < \Delta U.}$$

98. (c) $E = -\frac{RT}{2F} \ln \frac{1}{[\text{H}^+]^2}$

$$= -0.0591 \log \frac{1}{[\text{H}^+]} = -0.0591 \log 10 = -0.0591.$$

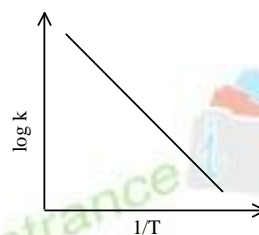
99. (d) According to Arrhenius equation

$$\log k = \log A - \frac{E_a}{2.303R} \cdot \frac{1}{T}$$

Plot of $\log k$ vs $\frac{1}{T}$ is a straight line

$$\text{Slope} = -\frac{E_a}{2.303R}$$

$$\text{Intercept} = \log A.$$

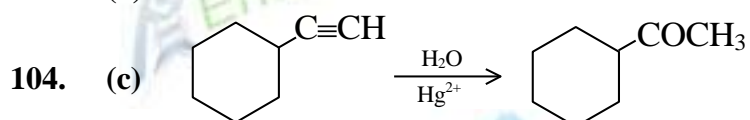


100. (c) The bond between chlorine and benzene has a partial double bond character.

101. (d)

102. (a)

103. (a)



105. (d)

106. (d)

107. (a)

108. (b)

109. (a)

110. (b)

111. (d) -CHO or >C=O can be reduced to $\text{-CH}_2\text{OH}$ or >CHOH respectively by H_2 + catalyst. LiAlH_4 or NaBH_4 .

112. (c)

113. (a)

114. (c) $P = P_A^\circ x_A + P_B^\circ x_B$
Let mole fraction of benzene = x_A
Let mole fraction of toluene = $(1 - x_A)$
 $380 = 500 x_A + (1 - x_A) 200$
 $x_A = 0.60$.

115. (a)

116. (a)

117. (c)

118. (a)

119. (b)

120. (a)