Register Number	1.7	

			50	TENCE	
			(Engli	ish Versi	on)
Tim	e All	owed : $2\frac{1}{2}$ Hou	ırs]		[Maximum Marks : 100
				14	
			P	ART - I	
			(Marks	: 20 × 1 =	= 20)
		N.B.: i)	Answer all th	e question	ns.
		ti) Choose and v	vrite the c	correct answer.
		ii	i) Each question	n carries o	one mark.
1. The angle of contact between pure water and clear gla				d clear glass is	
	a)	90°		b)	0*
	c)	140°	3 4 3	d)	180*
2.	Wh	ich of the follo	wing substances	has least	specific heat capacity?
	a)	Aluminium		b)	Mercury
	c)	Copper		d)	Lead.
3.	The	boiling point of	of water inside th	ne pressui	re cooker is
	a)	100°C		b)	0°C
	c)	120°C	1 1 1 1 1 1 1 1 1	d)	– 120°C.

4.	During change of state the temperature of the substance						
	a)	increases	b) decreases			
	c)	does not change	d)	increases or decreases.	14		
5.	W	hen an object reflects all	colours simulta	ineously, it appears			
	a)	white	b)	blue			
	c)	red	d)	green.			
6.	Th	e resistance of a conduc	ctor carrying a	current 3A with a potential differenc	e		
		V between its two ends					
	a)	5 ohm	b)	45 ohm			
	c)	$\frac{1}{5}$ ohm	d)	30 ohm.			
7.	Wh	nen a radioactive nucleus	s disintegrates	by emitting alpha particle its atomic			
		mber	100	y apart paracic its atomic	2		
	a)	increases by two	b)	decreases by two			
	c)	increases by four	d)	decreases by four.			
8.	The	catalyst which increases	the decomposi	ition of hydrogen peroxide is			
	a)	Iron	b)	Platinum			
	c)	Nickel	d)	Manganese dioxide.			
9.	Glas	ss is attacked by					
	a)	HCI	b)	H ₂ SO ₄			
	c) .	HF	d)	HNO 3.			

10.	The substance that converts hard water to soft water is							
	a) sodium carbonate	b)	sodium hydrogen carbonate					
	c) calcium carbonate	d)	magnesium carbonate.					
11.	In Solvay process, the salt that se	eparates out	in carbonating tower is					
	a) NH 4 HCO 3	b)	NaHCO 3					
	c) Na ₂ CO ₃	d)	CaCl ₂ .					
12.	The purest form of iron is							
	a) cast iron	b)	wrought iron					
	c) steel	d)	haematite.					
13.	The functional unit of skeletal m	uscle is						
	a) neuron	b)	nephron					
	c) sarcomere	d)	cell.					
14.	Spoilage of milk is caused by							
	a) Enterobacter	b)	Azotobacter					
	c) Streptococcus	d)	Lactobacillus.					
15.	The genetic code has	codons.						
	a) 50	b)	40					
	c) 35	d)	64.					
16.		he genetic ir	nformation for the sequence of amino					
	acids is	b)	tRNA					
	a) mRNA	and an	RNA.					

17.	Wh	ich of t	he follow	ing purif	les blood	?	
	a)	Onion	is ,			b)	Garlic
	c)	Turm	eric			d)	Cinchona.
18.	B.C	.G. vac	cination g	gives imn	nunization	n again	st
	a)	Polio				b)	Tetanus
	c)	Tuber	culosis			d)	Diphtheria.
19.	Wile	d asses	are confi	ined to			* =
	a)	Gir fo	rest			b)	Rann of Kachchh
	c)	Sunde	erbans			d)	Nilgiri hills.
20.	Whi	ch part	of the p	lant is af	fected by	chloro	sis disease ?
	a)	Roots		oll ag		b)	Stem
	c)	Leave	8			d)	Fruits.
		w ²					
					PART	- п	
				(M	larks : 10	× 1 =	10)
	N.B.	. : i)	Answer	all the q	uestions.		
		ii)	Each qu	estion ca	arries one	mark.	
		iii)	Answer	should b	e in a wo	rd or i	n few words or in one line
21.	Why	do the	sky diver	rs use a	spread-ea	gle pos	sition while falling?
22.	Name the heater used in calorimeter experiment.						
23.	What	t is the	commerc	cial unit	of electric	energy	7?

24. What is the atomic number of the element $_{92}$ U 235 ?

- 25. Why is cooked food spoiled quickly in summer season?
- 26. What happens when washing soda is kept in air for a long time?
- 27. What is the functional group of Aldehydes?
- 28. Why is the skin slimy in frog?
- Mention the disease in which the blood glucose is too high.
- 30. Mention any one primary nutrient.

PART - III

 $(Marks : 15 \times 2 = 30)$

- N. B: i) Answer any fifteen questions.
 - ii) Each question carries two marks.
 - iii) Students should answer the Question Nos. 32 and 38 compulsorily. These two questions are not included in the option.
- Define Sublimation.
- 32. Calculate the centripetal force required by a car of mass 500 kg which takes a round turn of radius 50 metre with a velocity of 20 ms⁻¹.
- 33. Why do substances expand on heating?
- Define dispersion of light.
- 35. Write any two differences between compound microscope and astronomical telescope.

- 36. State Fleming's left hand rule.
- 37. What is a nuclear reactor?
- 38. The hydrogen ion concentration of a solution is 1×10^{-4} mol. l^{-1} . Find the hydroxyl ion concentration of the solution.
- 39. What is called setting of cement?
- 40. Write any four types of glass.
- 41. What is flux?
- 42. Why is sulphuric acid called the king of chemicals?
- 43. How is soap prepared?
- 44. Write any two uses of ethanoic acid.
- 45. How will you identify (a) Anopheles (b) Culex?
- 46. What is template DNA?
- 47. How are chromosomes classified based on the position of the centromere?
- 48. Draw a paddy seed and label its parts.
- 49. What is naturopathy?
- 50. Write any two uses of Vinca rosea.
- 51. What are the common air pollutants?
- 52. Mention the three methods of prawn culture.

PART - IV

$(Marks : 8 \times 5 = 40)$

- N. B.: i) Answer eight questions by choosing at least two questions from each Group.
 - ii) Each question carries five marks.
 - iii) Draw the diagrams wherever necessary.

GROUP - A

- a) State and explain Bernoulli's theorem.
 - Mention any two applications of Bernoulli's theorem.
- Describe the construction and working of a compound microscope.
- State Ohm's law and explain how it can be verified.
- 56. Write the applications of radio-isotopes in the field of medicine.

GROUP - B

- 57. a) Define the rate of chemical reaction.
 - b) Define Ionic product of water.
 - State Lowry and Brönsted theory.
- Describe how Aluminium is extracted from its ore by electrolysis process.
- 59. What is fermentation? How is ethanol prepared by fermentation?

GROUP - C

- 60. Describe the asexual reproduction in Penicillium.
- 61. Explain Watson and Crick model of DNA with the help of a diagram.
- 62. What are the two types of stem cells ? Explain briefly.
- 63. Describe the control measures of noise pollution.
- 64. Give an account of cultivation of paddy.