Enrolment No.

GUJARAT TECHNOLOGICAL UNIVERSITY

B.E. Sem-III Regular / Remedial Examination December 2010

Subject code: 132603

Subject Name: Thermodynamics of Elastomers and Polymers Date: 16 /12 /2010 Time: 10.30 am – 01.00 pm Total Marks: 70

Instructions:

- 1. Attempt all questions.
- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.
- 4. Assume suitable data wherever necessary.

Q.1 Answer the following

- (i) Write down any two limitations of first law of thermodynamics.
- (ii) Define the terms: (1)Heat of reaction (2) Heat of formation
- (iii) List the any two important properties of a refrigerant.
- (iv) Explain the following terms:(1)Degree of Freedom or Variance (2)Polymorphism
- (v) What do you mean by Eutectic System? Give an example of it.
- (vi) Write any two characteristics of a chemical equilibrium.
- (vii) How the proximate analysis of air-dried coal is done?

Q.2 (a) Explain the polymer solutions in detail with respect to its thermodynamics . 07

- (b) Solve the following sums:
 - (i) 2 gas of CO₂ is contained in a piston cylinder assembly at a pressure of 65bar & a temperature of 300°K. The piston has a mass of 5000kg and a surface area of 1m². The friction of the piston on the walls is significant & can not be ignored. The atmospheric pressure is 1.01325bar. The latch holding a piston in position is suddenly removed & the gas is allowed to expand. The expansion is arrested when the volume is double the original volume. Determine the work appearing in the surroundings.
 - (ii) A man whose weight is 600N takes 2 minutes for climbing up a staircase.03 What is the power developed in him, if the stair case is made up of 20 stairs each 0.18m in height?

OR

	(b)	Derive any three equations which show Maxwell's thermodynamic equations.	07
Q.3	(a)	Explain the factors which are taken into consideration during selection of coal.	07
	(b)	Write a short note on Effect of crosslinking on solubility	07
		OR	
Q.3	(a)	Give the difference between Low temperature carbonization & High temperature carbonization.	07
	(b)	Discuss in about solubility parameter.	07
Q.4	(a)	How the variation in heat of polymerization of various monomer arises? Explain it in detail.	08
	(b)	Write a short note on Eutectic System.	06
		OR	

.

14

Q.4	(a) (b)	How the estimation of heat of polymerization is carried out? Derive the equation of phase rule.	07 07
Q.5	(a) (b)	 Derive the equation of law of mass action by using Vant Hoff equilibrium box. A boiler is fired with a coal with composition: C= 75%, H= 9%, S= 2%, O= 4%, N= 3%, ash= 7%. Calculate: (1) Gross & Net calorific value of 1kg of coal(Latent heat of steam=587kcal/kg) (2) Minimum theoretical air required for combustion of 1kg of coal(by weight and by volume) (3) Percentage composition of dry flue gas if 25% excess air is used. 	07 07
		OR	
Q.5	(a)	State the Le-Chatelier-Braun principle and discuss the factors which are taken into consideration in application of this principle.	07
	(b)	An ideal gas undergoes the following sequence of mechanically reversible process:	07

(1) From an initial state of 70°C and 1bar, it is compressed adiabatically to 150°C

(2) It is then cooled from 150°C to 70°C at constant pressure.

(3) Finally it is compressed isothermally to its original state.

Calculate W,Q, Δ U, Δ H for each of three processes & for entire cycle.(Take C_p=7/2*R, C_v=5/2*R, r=1.4)
