Seat No.: _____

Enrolment No._____

GUJARAT TECHNOLOGICAL UNIVERSITY

B.E. Sem-III (Rubber Technology) Examination December 2009

Subject code: 132601 Date: 17 / 12/ 2009			Subject Name: Basic Rubber Science Time: 11.00 am – 1.30 pm Total Marks: 70			
Instructions: 1. Attempt all questions. 2. Make suitable assumptions wherever necessary. 3. Figures to the right indicate full marks.						
Q.1	(a) (b)	Discuss the characteristic properties	of rubber.	06		
	(0)	(i) Write in brief about 'segmental n(ii) Explain the conditions, which ar behave as rubber.	notion' in rubber. e necessary for any material to	02 06		
Q.2	(a)	Answer the following (i) What do you mean by isoelectric point? Discuss in detail with even	point and electro-osmosis	04		
		(ii) Write the preparation of colloida	Il solution by reduction method	03		
	(b)	State the experimental laws of frie these laws for applications to rubber	ction. Give the modification to	07		
	(b)	Write a short note on diffusion in ela	astomers.	07		
Q.3	(a)	Answer the following(i) Give the importance of density de the method for determining relation powder and also list the sources of the sourc	etermination in rubber. Describe ve density of carbon black of error in it.	05		
		(ii) Write a short note on 'bulk mod collision'.	ulus', 'shape factor' and 'elastic	04		
	(b)	Explain the working of 'Lectone du be the surface tension if mass of liqu circumference of ring is 3cm.	Nouy tensiometer'. What will uid raised is 5grams and the	05		
OR						
Q.3	(a)	 Answer the following (i) Explain the phenomenon of total and destructive interference and plight in glass if refractive index of the second secon	internal reflection, constructive polarization. Find the velocity of f glass is 1.5.	05		
	(b)	(ii) Describe different modes of heat Discuss on transmissibility of oscilla	transfer. ations and vibration isolation.	04 05		
Q.4	(a)	Answer the following (i) Give the difference between add	ition polymerization and	04		
		(ii) Write a short note on 'initiator'.		04		

(b)	Explain the bulk polymerization technique with it's advantages and	06
	disadvantages.	

OR

Q.4	(a)	Answer the following	
		(i) Which are the various modes of addition of incoming monomers to growing chain in propagation step?	04
		(ii) Explain the terms 'organic polymer' and 'inorganic polymer' respectively with suitable examples and their structures.	04
	(b)	Write a short note on cationic polymerization.	06
Q.5	(a)	Answer the following	
		(i) Give the difference between 'crystalloids' and 'colloids'.	02
		(ii) Write the applications of emulsions.	02
		(iii) Explain the Tyndall effect.	02
	(b)	Discuss the purification of colloidal solution.	08
		OR	
Q.5	(a)	Answer the following	
		(i) Write about Schulze-Hanry law.	02
		(ii) What do you mean by 'Micelles'? Write it's examples also.	02
		(iii) Draw the figure for formation of gel showing colloidal	02
		dispersion and gel structure.	-
	(b)	Explain in detail about the difference between lyophillic sol and	08
		lvophobic sol.	
