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B. Tech
SCC 2001

Second Semester Examination, 2004

Chemistry-I

Full Marks : 70

Time : 3 hours

Answer Section—A which is compulsory and
any five from Section—B

The figures in the right-hand margin indicate marks

(Standard data : $h = 6.626 \times 10^{-34}$ Js,
 $R = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$, $c = 3 \times 10^8 \text{ m s}^{-1}$,
 $1 \text{ amu} = 1.6605 \times 10^{-27} \text{ kg}$, Atomic weight of
Cl = 35)

SECTION—A

1. Write short answers to the following
questions, preferably in *one* sentence for each
question : 2 × 10

(a) Write down the relationship between the
energy and the velocity of a photon.

(T. Over)

~~(b)~~ Arrange the following in the decreasing order of energy :

(i) UV-visible ν

(ii) IR ν

(iii) Microwave ν

(iv) X-ray ν

~~(c)~~ Which of the following pairs is iso-electronic ?

(i) N_2 and O_2

(ii) NO^+ and N_2

(iii) B_2 and C_2

~~(d)~~ The lattice constant of an FCC lattice is 'a'. What is the length of the face-diagonal of the lattice ?

~~(e)~~ 'Spontaneous processes are irreversible in nature.' Is this statement true ?

~~(f)~~ C_p of an ideal gas is $5 \text{ cal mol}^{-1} \text{ K}^{-1}$. Calculate its C_v . Assume the gas to be ideal.