### Elements of Mechanical Engineering (ME-101, Dec.2005)

Time: 3 Hours

# Max. Marks: 60

Note: Question No. 1 is compulsory. Attempt five questions from section A and B, taking at least two questions from each section.

## Section-A

- 1. (a) State the Zeroth law of Thermodynamics.
  - (b) Discuss whether (Pdv + Vdp) represent the system properties or not?
  - (c) How enthalpy is related to the internal energy?
  - (d) State the principle of entropy, draw T-S diagram for it.
  - (e) A close system contains air at 1 bar, 190 K and occupies 0.02 m<sup>3</sup> of volume. This system is heated at constant volume till pressure become 4 bar. For the problem draw P-V and T-S diagram only.
  - (f) Draw P-V and T-S diagram for an engine operated by Dual cycle.
  - (g) State the condition of reversibility of lifting machine.
  - (h) Draw a stress-strain curve for a mild steel specimen.
  - (i) What is fatigue, define?
  - (j) Which of the following material has the maximum ductility?
    - (a) Mild steel (b) Copper (c) Zinc (d) Aluminium

## Section-B

- 2. (a) State the condition for a process to be reversible.
  - (b) Classify the thermodynamic system and explain them with the help of simple diagrams.
  - (c) A non flow reversible process occurs for which pressure and volume are correlated by the expression  $P=(V^2 + 6V)$ , where P is in bar and V is m<sup>3</sup>. What amount of work will be done when volume changes from 2 to 4 m<sup>3</sup>?
- 3. (a) Explain the corollaries of first law.
  - (b) A close system having a mass of 50 Kg has an initial velocity of 10 m/s. Subsequently its velocity increases to 30 m/s and its elevation also rises by 40 m. During this process the system receives 30000 J of heat and 4500 J of work. If the system delivers 0.002 KWh of electrical energy, determine the change in internal energy of the system.
- 4. (a) Differentiate between different polytropic process, draw common P-V diagram showing expression follow the law PV<sup>n</sup>=C for n=0, n=1, n=n.
  - (b) 3 Kg of air kept at an absolute pressure of 100 Kpa and temperature of 300 K is compressed polytropically until the pressure and temperature becomes 1500 KPa and 500 K respectively. Evaluate the polytropic exponent, the final volume and the work of compression. Take gas constant, R=287 J/Kg K.
- 5. (a) Explain revised heat engine Carnot cycle with the help of P-V diagram.
  (b) Three real heat engines have the same thermal efficiency and are connected in series. The first engine absorbs 2400 KJ of heat from a thermal reservoir at 1250 K and the third engine rejects its waste of 300 KJ to a sink at 150 K, determine the work output from each engine.

## Section-C

- 6. (a) Explain the working principle of an air standard cycle with the help of P-V and T-S diagram.
  - (b) A four stroke engine working on diesel cycle has a piston diameter 25 cm, stroke of 40 cm and clearance volume of 1200 cc. The fuel injection takes place for 5% of stroke. If the induction pressure corresponding to 1 bar and engine turns 5 rev/s, find the air standard efficiency of the engine and power developed.
- 7. (a) Define input, output and efficiency of a simple lofting machine.
- (b) A simple lifting machine raised a load of 360 N through a distance of 200 mm. The effort a force of 60 N is moved 1.8 m during the process. Calculate velocity ratio, mechanical advantage and efficiency of the machine.
- 8. Write notes on the followings
- (a) Torsion and bending (b) Stress and strain (c) Sudden and impact load (d) Modulus of rigidity and Bulk modulus.
- 9. (a) What is creep? Explain the different stages of creep with the help of common diagram.
- (b) Differentiate between torsion and bending.