

**M.Tech. / II Sem.**

**J**

**NUCLEAR SCIENCE AND TECHNOLOGY**

**Paper NST-606— Applied Thermodynamics**

**Time . 3 hours**

**Maximum Marks 70**

*(Write your Roll No on the top immediately on receipt of this question paper)*

*Attempt all the questions.*

*Answer briefly and according to the marks allotted for each question.*

*Use steam table where it is necessary.*

1 (a) Show that the internal energy of the following systems is a function of Temperature only

(I) an ideal gas

(II) an incompressible substance

substance is a function of temperature only

(b) Why would we use an intercooler between compressor stages?

(c) What is reheat cycle? Mention two benefits of a reheat cycle

(d) Two kg water at 200 kPa with a quality of 25% has its temperature raised to 250°C in a constant pressure process. What is the change in enthalpy?

(e) Discuss Mollier diagram and its significances in thermodynamic process

**(4+4+4+4+4) Marks**

2. (a) Calculate volume, enthalpy, internal energy and entropy of 5 kg of steam at 0.8 MPa pressure under following conditions

(I) Dry and saturated

(II) Wet steam having wetness of 37 %

(III) Superheated steam at 250°C

**(2+2+2) Marks**

(b) The pressure at the end of compression of a steam engine having a compressor ratio is 0.6 MPa. If the pressure and temperature at 0.15 of stroke were 0.1 MPa and 70°C respectively, find the law of

compression and the value of final temperature **(6  $\frac{1}{2}$  Marks)**

OR

(a) Explain the principle of boilers and mention the types of boilers

Discuss the locomotive boilers

**(1+1+4) Marks**

(b) Derive the Euler and Bernoulli equation from Steady Flow Energy Equation (S F E E ) for steady flow system. What is the difference between the work for non-flow and flow process? Show with the help of PV diagram?

**(3+2+1  $\frac{1}{2}$ ) Marks**

3. (a) What is a heat exchanger? Derive the heat transfer rate for parallel flow heat exchanger? Why are counter flow heat exchangers superior to parallel flow heat exchanger?

**(2+5+2) Marks**

(b) Use the formulae and steam table to find the relative humidity and specific humidity corresponding to 25°C dry bulb and 21°C wet bulb temperature

**(3  $\frac{1}{2}$ ) Marks**

4. (a) A diesel engine has a state before compression of 95 kPa, 290 K, and a peak pressure of 6000 kPa a maximum temperature of 2400 K. Find the volumetric compression ratio and the thermal efficiency

**(4  $\frac{1}{2}$ ) Marks**

(b) Explain the vapour compression cycle with the help of T-s and p-h diagrams. Explain the effect of superheat and subcooling on this cycle. Derive the expression for COP

or

A fluid undergoes a reversible adiabatic compression according to the law  $pv^n = \text{constant}$ . Determine the change in enthalpy, internal energy and entropy, and the heat transfer and work transfer during this process

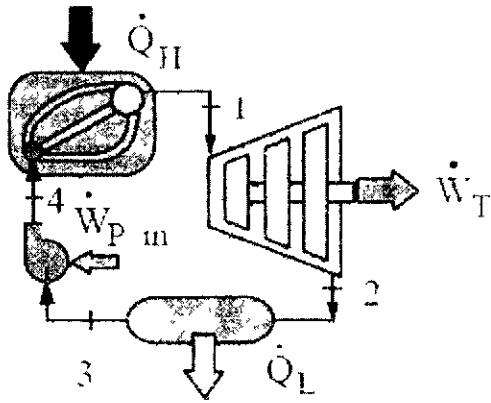
**(8) Marks**

5. Consider a steam turbine power plant operating near critical pressure, as shown in Fig. 1. As a first approximation, it may be assumed that the turbine and the pump processes are reversible and adiabatic. Neglecting any changes in kinetic and potential energies, calculate
- (a) The specific turbine work output and the turbine exit state
- (b) The pump work input and enthalpy at the pump exit state

(c) The thermal efficiency of the cycle

(4+4+4 $\frac{1}{2}$ )

Marks



$$P_1 = P_4 = 20 \text{ MPa}$$

$$T_1 = 700 \text{ } ^\circ\text{C}$$

$$P_2 = P_3 = 20 \text{ kPa}$$

$$T_3 = 40 \text{ } ^\circ\text{C}$$

Fig 1