[KV 872]

Sub. Code: 5002

B.Sc. (Medical Laboratory Technology) DEGREE EXAMINATION

FIRST YEAR Paper II – BASIC CHEMISTRY

O.P. Code: 725002

Time: Three hours

Maximum: 100 marks

Answer All questions.

I. Essays:

(2X15=30)

- 1. What is an indicator? Explain its function giving two examples. Why does the colour depend upon P^H? What indicator should be used in i) Titration of H2SO4 with Na2Co3 and ii) Acetic acid with NaOH. (2+3+5+5)
- 2. What are nucleic acids? Discuss the structure of DNA. What is meant by replications of DNA? (5+5+5)

II. Write Short Notes on:

(10X 5=50)

- 1. Buffer solution.
- 2. Mutarotation of glucose.
- 3. Biological importance of carbohydrates.
- 4. Classification of proteins.
- 5. Two colour reactions of cholesterol.
- 6. N-terminal analysis of proteins.
- 7. α helix structure of protein.
- 8. Any one coenzyme and its functions.
- 9. Specificity of enzymes.
- 10. Mode of action of enzymes.

III. Short Answer Questions:

(10X2 = 20)

- 1. Define Lewis concept of acids and bases.
- 2. What are intracellular materials present in the animal cell?
- 3. What is golgi apparatus? Mentions two functions.
- 4. What is mitochondrion?
- 5. What are the classifications of carbohydrates?
- 6. What is meant by racemic modification?
- 7. Write any four types of lipids.
- 8. What is the sugar residue present in RNA?
- 9. How can you distinguish ketohexoses for aldohexoses?
- 10. What is the use of Ninhydrin test?
