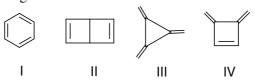
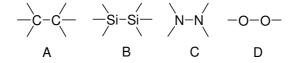
CHEMISTRY

- 1. If the rate of diffusion of O_2 through an orifice is r_1 , then the rate of diffusion of the same volume of H_2 through the same orifice under identical conditions will be
 - A. r_1
 - B. $2r_1$
 - $C. 4r_1$
 - D. 8r₁
- 2. Which of the following having the molecular formula C_6H_6 contain(s) a single set of structurally equivalent hydrogen atoms?



- A. Only I is correct
- B. I, II, and III are correct
- C. Both III and IV are correct
- D. Both I and III are correct
- 3. For equimolar quantities of the following compounds the maximum depression in freezing point of water will be observed in
 - A. Potassium ferrocyanide
 - B. Potassium ferricyanide
 - C. Potassium chloride
 - D. Glucose
- 4. Which of the following ion is expected to be colorless
 - A. Cu^{2+}
 - B. Ti⁴⁺
 - C. V^{3+}
 - D. Fe²⁺
- 5. Which among the following chemical species has the highest E-E bond energy?

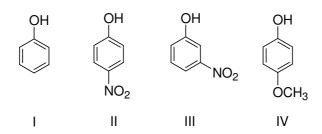


- 6. For an elementary reaction which is the correct statement?
 - A. Order > molecularity
 - B. Order < molecularity
 - C. Order = molecularity
 - D. No relation between order and molecularity

7. Which of the following substrates will have the fastest rate of nucleophilic substitution by an iodide ion?

 CH_3CH_2Br $CH_3CH_2CH_2Br$ $(CH_3)_2CHCH_2Br$ $(CH_3)_3CCH_2Br$ A B C D

- 8. The bond order and magnetic behavior of NO⁺ is
 - A. Bond order = 2; paramagnetic
 - B. Bond order = 2; diamagnetic
 - C. Bond order = 3; paramagnetic
 - D. Bond order = 3; diamagnetic
- 9. Arrange the following compounds in decreasing order of their pK_a values



- A. I > II > III > IV
- B. III > II > IV > I
- C. II > III > I > IV
- D. IV > III > II > I
- 10. On oxidation a compound having the molecular formula $C_4H_{10}O$ can be converted into C_4H_8O . The original compound could be:
 - I. Primary alcohol, II. Secondary alcohol, III. Tertiary alcohol.

Which of the following options is correct?

- A. I, II, and III are correct
- B. Both I and II are correct
- C. Only I is correct
- D. Only II is correct