Register	Timou 21	d lan.	
Number			

# SCIENCE (Theory) — Paper I (Physics and Chemistry)

Time Allowed :  $2\frac{1}{2}$  Hours ]

I.

[ Maximum Marks: 100

Instruction: Check the question paper for fairness of printing. If there is any lack

of fairness, inform the Hall Supervisor immediately.

## (PHYSICS)

(Marks: 50)

### SECTION - A

		SECTION		
		Answer all the	questi	ions.
	3			$10 \times 1 = 10$
Cho	ose th	ne correct answers :		$10 \times 1 = 10$
1.	100	0 watt is equal to		
	a)	746 HP	b)	74600 HP
	c)	1.34 HP	d)	3.6×10 <sup>3</sup> HP.
2.	A co	one balanced on its tip is an exa	mple f	for aver edges (e
	a)	stable equilibrium	b)	unstable equilibrium
	c)	neutral equilibrium	d)	none of these.
3.	The	centre of gravity of the displace	d liqu	id is called
	a)	centre of gravity	b)	centre of buoyancy
	c)	centroid	d)	metacentre.
4.	The	S.I. unit for heat is		
	a)	joule	. p)	°C and stane testares soft . At
	c)	*F with white dispesses arts of	d)	K. The change from the .X

	5	5. Th	e fixed temperati	ure at which a liqui	d changes into var	oour is called
		. a)	melting point	b		
		c)	evaporation	d)	condensation.	
	6	. Pre	esbiopia can be co	onnected using		
		a)	convex lens	b)	concave lens	
		c)	bifocal lens	d)	cylindrical lens	S bowolfA batt
	7.	The	dip at the equat	or is		
		a)	0°	b)	45°	
		c)	60°	d)	90°.	
	8.	The	pipe instrument	without reed is		
		a)	clarinet	b)	trumpet	
*		c)	organ	<u>,</u> d)	harmonium.	
	9.	Flen	ning's Right Hand	l Rule is used in		
		a)	motor	b)	dynamo	
		c)	transformer	d)	microphone.	
	10.	The i	radiations not de	flected by electrical	and magnetic field	ds are
			alpha rays	b)	beta rays	
•		c)	cathode rays	d)	X-rays.	
II.	Con	nplete	the following usin	ng appropriate word	d / words / expres	sions: 5 - 1 - 5
	11.	The n	nathematical exp	ression for Joule's	Law of Heating	0 7 1 2 3
	12.	In Tai	mil Nadu the nuc	elear power plant is	situated in	
	13.	The ı	init for frequency	/ is		
	14.	The cr	titical angle for gl	lass is		
	15.	The cl		solid state to the g		tly on heating is

#### SECTION - B

Answer any five of the following in one or two sentences each:

 $5 \times 2 = 10$ 

- 16. State the law of conservation of energy.
- 17. Define metacentre.
- 18. Compare open organ pipe and closed organ pipe.
- 19. Mention the element of earth's magnetic field.Give the reasons for the following:
- 20. A fuse is used in electrical appliances.
- 21. Water should be continuously circulated around the anode of an X-ray tube.

  Give two uses or practical applications of the following:
- 22. IR rays.
- 23. Specific heat.

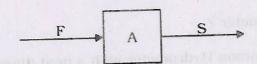
#### SECTION - C

Answer any five of the following, choosing at least one question from each Part:

 $5 \times 5 = 25$ 

#### PART - I

24.



In the diagram F is the force acting on a body A and S is the displacement.

a) What is the work done?

7

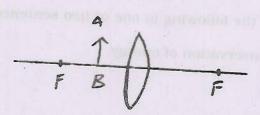
b) Give the S.I. unit of work.

c) If the body does not move, what is the work done?

A

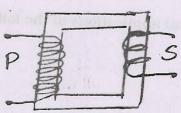
d) If the mass of the body is 10 kg, and the distance moved by it is 20 ms, calculate the work done.

25.



a) Name the optical instrument.
b) What is the nature of the image?
c) Complete the ray diagram.
d) Mention any one of its uses.

26.



	a)	Name the device. Why is it called so?
	b)	Name the principle on which it works.
27	c)	Mention one of its uses.
	d)	What is turns ratio?
		PART - II
27.	a)	What is a Lactometer?
	b)	Describe the common Hydrometer with a neat diagram.
•	c)	Mention any one use of Hydrometer.
28.	Des	cribe an experiment to determine the melting point of naphthalene.
29.	State	e the laws of transverse vibrations of a stretched string and the
		train builte Hors viol move, what is the work done, com-
30.	Desc	ribe the construction and working of a dip cicle.

31. Describe the Coolidge tube and the production of X-rays.

# (CHEMISTRY)

(Marks: 50)

## SECTION - A

Answer all the questions.

Cho	$10 \times 1 = 10$					
1. The ratio by weight of hydrogen and oxygen in water is						
	a)	1:8 and annuboted add	b)	8:1		
	c)	2:1	d)	1 : 2. [salottem] of	9 .0	
2.	The	molar volume is litr	es.			
	a)	2.24	b)	22.4		
*	c)	224	d)	none of these.		
3.		is an electrovalent compo	ound.	(amerijātā )		
	a)	HCl and the amount of the	b)	NaCl	6	
	c)	NH <sub>3</sub>	d)	CH <sub>4</sub> .		
4.		can be stopped by a thi	ick sh	neet of lead.		
	a)	Alpha	b)	Beta		
	c)	Gamma	d)	All of these.		
5.		is used in dating ancie	ent sp	pecimens.		
	a)	Na <sup>23</sup>	b)	Na <sup>24</sup>		
	c)	C <sup>12</sup>	d)	$C^{14}$ .		
6.	Tyn	dall effect is exhibited by				
	a)	true solution	b)	colloidal solution		
	0)	suspension	d)	all of these.		

17. What are metalloids?

7.	is the most stable form of phosphorus.							
	a)	Scarlet			b)	Violet		
	c)	White			d)	Red.		
8.	·		is used t	o neutralize a	cidity	in the stomac	ch.	
	a)	Sodium	hydroxide		b)	Sodium chlo	oride	
	c)	Sodium	carbonate		d)	Sodium bica	arbonate.	
9.	The	functiona	al group R-0	OH is				
	a)	acid			b)	alcohol		
	c)	ether	22.0		d)	ester.		
10.	•••••		does not	decolourise a	alkalir	ne potassium	permanga	anate.
	a)	Methane			b)	Ethene	E	
	c)	Ethyne			d)	None of these	e.	
Con	plete	the follow	ring, using a	appropriate w	ord/w	vords/express	sions :	5 × 1 = 5
11.			is a liqui	d non-metal.				
12.	The	promoter t	used in the	contact tower	is			
13.	Gain	of electron	ns is					
14.	Dete	rgents are	salts of				1	
15. Carbon combining with itself to form long chains is called								
				SECTION - B				
Answer any five questions in one or two sentences each: $5 \times 2 = 10$								
16.	What	is radioac	tivity?					

Complete and balance :

- 18.  $Fe_2O_3 + CO \rightarrow$
- 19.  $Ag+H_2SO_4 \rightarrow$

Give reasons:

- 20. White phosphorus is banned in the match industry.
- 21. Methanol is more poisonous than ethanol.

  Give two practical applications:
- 22. Aluminium.
- 23. Ethylene.

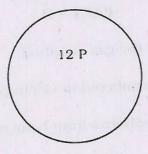
#### SECTION - C

Answer any five of the following, choosing at least one question from each Part:

 $5 \times 5 = 25$ 

#### PART - I

24. Study the following diagram and answer:



a) What is the atomic number?

1

b) What is the mass number?

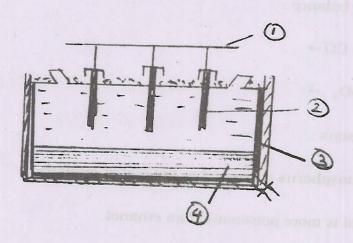
-

c) Name the element.

1

d) Complete the structure.

2



Label the parts. a) Why is charcoal added? b) Write the Electrode reactions. c) 26. You are provided with a conical flask, delivery tubes, thistle funnel, trough gas jar and beehive shelf: How will you set up the apparatus for the preparation of acetylene? 3 Why is sand added? b) Mention one of its uses. c) PART - II 27. State the gas laws and derive the gas equation. 3 + 228. What are the points to be remembered in electroplating? 29. Explain the extraction of phosphorus from bone ash. 30. Give any five differences between organic and inorganic compounds.

31. Explain the manufacture of soap by hot process.