1. The number of free electrons per 10 mm of an ordinary copper wire is  $2 \times 10^{21}$ . The average drift speed of the electrons is 0.25 mm/s. The current flowing is: A. 0.8 A B. 8 A C. 80 A D. 5 A 2. Which of the following cells is more likely to be damaged due to short circuiting? C. Acid D. Fuel A. Daniel B. Dry 3. A gas expands from 5 litre to 105 litre at a constant pressure  $100N/m^2$ . The work done is C. 8 Joule A. 1 Joule B. 4 Joule D. 10 Joule 4. The Helium nuclei can be formed from A. Hydrogen nuclei by process of chain reaction B. Hydrogen nuclei through nuclear fission C. Hydrogen nuclei through nuclear fusion D. None of these 5. In the atom bomb dropped by Americans in 1945 on Nagasaki, Japan, the fissionable material used was A. Helium 4 B. Plutonium 239 C. Uranium 235 D. Uranium 233 6. The engine of a truck moving a straight road delivers constant power. The distance travelled by the truck in time *t* is proportional to D. *t*<sup>3/2</sup>  $\mathbf{B} t^2$ A. *t*  $C. \sqrt{t}$ 7. The velocity of electron in ground state of hydrogen atom is A.  $2 \times 10^5$  B.  $2 \times 10^6$  C.  $2 \times 10^7$  D.  $2 \times 10^8$ m/s m/s m/s m/s 8. The radius of the first orbit of the electron in a hydrogen atom is  $5.3 \times 10^{-11}$  m; then the radius of the second orbit must be C.  $21.2 \times 10^{-11} \text{ m}$ A.  $15.9 \times 10^{-11} \text{ m}$ D 42.4 x  $10^{-11}$  m B. 10.6 x 10 m 9. A person pushes a rock of  $10^{10}$ Kg mass by applying a force of only 10N for just 4 seconds. The work done is A. 1000 Joule B. 0 J C. nearly zero D. positive 10. One can take pictures of objects which are completely invisible to the eye using camera films which are sensitive to B. sodium light C. visible light D. infra-red rays A. ultra-violet rays 11. Light from a 100 watt filament bulb is passed through an evacuated glass tube containing sodium vapour at a high temperature. If the transmitted light is viewed through a spectrometer, we will observe A.  $D_1$  and  $D_2$  lines of sodium with good B. dark lines where  $D_1$  and  $D_2$  lines should have intensity been observed C. continuous radiation from the bulb only D. the entire emission spectrum of sodium

12. Under the action of				
· · · ·	g a constant acceleration.			
The power is A. zero	B. positive			
C. negative	D. increasing uniformly	V		
C	with time			
	c lens the radius of curva cm, the refractive index		the is 10 cm and the focal as will be	
A. 1.5	B. 1.66	C. 1.33	D. 3	
14. A plane convex len length of lens is	s has radius of curvature	30 cm. If the refractive	index is 1.33, the focal	
A. 10 cm	B. 90 cm	C. 30 cm	D. 60 cm	
•	converging towards a point of the beam $= t$ , refrate point is shifted by	-		
•	B. t $(1 + 1/\mu)$ away	C. t (1 - $1/\mu$ ) nearer	D. t $(1 + 1/\mu)$ nearer	
between the silts and so	creen is doubled. The fri	nge width will be	s halved and the distance	
A. unchanged	B. halved	C. doubled	D. quadrupled	
17. Wavelength of red light is $\lambda_r$ , violet rays is $\lambda_v$ and X -ray is $\lambda_x$ then the order of wavelengths is				
18. The amount of work	ach of mass <i>m</i> , to the roo		$D. \lambda_{r} > \lambda_{v} > \lambda$	
A. n mgh B. mgh/n	C. zero D. <i>ghn/m</i>			
	e state of resonance, wh	ich of the following state	ements is correct ? (cos	
φ)= A.0	B. 0.5	C. 1	D. None of these	
20 In LOD simulations	1:66	14	1.	
A. 80°	se difference between vo B. 90°	C. 145°	D. 0°	
A. 00	<b>D</b> . 70	C. 14J	D. 0	
21. If speed is plotted along x-axis and Kinetic energy against y-axis, then the graph obtained has a shape similar to that of				
A. circle	B. ellipse	C. hyperbola	D. parabola	
through 60°. The torqu	lying parallel to a magne e needed to maintain the	needle in this position w		
A $(\sqrt{3})$ w		Bw		

A.  $(\sqrt{3})$  w

	ards. A point <i>p</i> lies to the tance and another point <b>(</b>		
24. In a parallel arrang A. less than <i>R</i> <sub>2</sub>	ement if $(R_1 > R_2)$ , the period B. same as $R_2$	ower dissipated in resista C. more than <i>R</i> <sub>2</sub>	ance $R_1$ will be D. none of these
immaterial?	be installed in the supply ce of the material of the se wire	line in a house which or B. the diameter of the f D. none of these	-
motor $I_a$ is given by	ied, $E_a$ is emf drop acros B. $E_a/R_a$	s the armature, the armature, the armature, the armature, the armature $C. V- E_a/R_a$	ture current of a d.c. D. $V/R_a$
A. 1.35	B. 1.50 ent ratio $i_1/i_2$ depends up $i_1 R_1$ $i_2 R_2$ $i_2 R_2$	C. 1.00	having internal resistance erminals of the cell will be D. 1.20
terminals of the cell is A. 2(E - V)V/r 30. Copper and german	connected across a resista found to be V. The intern B. 2(E - V)r/E nium are both cooled to 7 r increases while that of ecreases	nal resistance of the cell C. (E - V) r/V 70 K from room tempera	must be D. (E- V)/r ture, then decreases while that of

31. The potential difference between the points A and B of the electrical circuit given is A. 1.5 V B. 1.0 V 32. A moving coil galvanometer has a resistance of 9.8 $\Omega$  and gives a full scale deflection when a current of 10 mA passes tbrough it. The value of the shunt required to convert it into a mini ammeter to measure current upto 500 mA is A. 0.02 $\Omega$  B. 0.2 $\Omega$  C. 2 $\Omega$  D. 0.4 $\Omega$ 

33. The total electrical resistance between the points *A* and *B* of the circuit shown in the figure is



34. If the plates of a charged parallel plate capacitor are pulled away from each other

A. capacitance increases	B. energy increases	C. voltage increases	D. voltage decreases
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35. A parallel plate capacitor is charged by connecting its plates to the terminals of a battery. The battery remains connected and a glass plate is interposed between the plates of the capacitor, then

A. the charge on plates will be reduced

B. the charge on plates will increase

C. the potential difference between the plates of the capacitor will be reduced

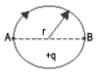
D. the potential difference between the plates of the capacitor will increase

36. A person weighing 70Kg wt lifts a mass of 30 Kg to the roof of a building 10 m high. If he takes 50 sec to do so,then the power spent is

A. 19.6 W B. 196 W C. 300 W D. 50 W

37. Work done in carrying a charge q from A to B along a semi-circle is

A. 2π <i>rq</i>	B. 4π <i>rq</i>
C. π <i>rq</i>	D. 0



38. A particle *A* has charge +q and particle B has charge +4q with each of them having the same mass m. When allowed to fall from rest through same electrical potential difference, the ratio of their speed  $V_A : V_B$  will become

A. 2:1 B. 1:2 C. 1:4 D. 4:1

39. The electric field at a small distance R from an infinitely long plane sheet is directly proportional to

A.  $R^2/2$  B. R/2 C.  $R^{-2}$  D. none of these 40. In the diagram, the electric field intensity will be zero at a distance A. between -q and +2q charge B. towards +2q on the line drawn C. away from the line towards +2q D. away from the line towards -q 41. Wein's displacement law is given by

A.  $\lambda_{m}$  B. T/ $\lambda_{m}$  C.  $\lambda_{m}$  T D. T =  $\lambda_{m}$ = = = = constant constant constant

42. If two electrons are forced to come closer to each to each other, then the potential energy A. becomes zero B. increases C. decreases D. becomes infinite

43. The specific heat at constant pressure is greater than that of the same gas at constant volume because

A. at constant volume work is done in expanding the gas

B. at constant pressure work is done in expanding the gas

C. the molecular attraction increases more at constant pressure

D. the molecular vibration increases more at constant pressure

44. The specific heats of CO<sub>2</sub> at constant pressure and constant volume are 0.833 J/kg.K and 0.641 J/kg.K respectively. If molecular weight of CO<sub>2</sub> is 44, what is the universal constant *R*? A. 4.19 x  $10^7$  erg/cal B. 848.8 J/gm/K C. 8.448 J/mol/K D. 4.19 J/cal

45. The freezing point of the liquids decreases when pressure is increased, if the liquid A. expands while freezing B. contracts while freezing C. does not change in volume while freezing D. none 46. The equation of a transverse wave on a stretched string is given by  $y = 0.05 \sin \pi$  (2t/0.002 -x/0.1) where x and y are expressed in metres and t in sec. The speed of the wave is A.100 B. 50 m/s C. 200 m/s D. 400 m/s m/sec

47. The ratio of velocity of the body to the velocity of sound is called

A. Magic number B. Laplace number C. Natural number D. Mach number

48. Television signals on earth cannot be received at distances greater than 100 km from the transmission station. The reason behind this is that

A. the receiver antenna is unable to detect the signal at a distance greater than 100 km

B. the TV programme consists of both audio and video signals

C. the TV signals are less powerful than radio signals

D. the surface of earth is curved like a sphere

49. A ball is thrown from a height of h m with an initial downward velocity  $v_0$ . It hits the ground, loses half of its Kinetic energy & bounces back to the same height. The value of  $v_0$  is

A.  $\sqrt{2gh}$  B.  $\sqrt{gh}$  C.  $\sqrt{3gh}$  D.  $\sqrt{2.5gh}$ 

50. A thick rope of rubber of density $1.5 \times 10^3$ kg/m <sup>3</sup> and Young's modulus $5 \times 10^6$ N/m <sup>2</sup> , 8m in length, when hung from ceiling of a room, the increase in length due to its own weight is A. 9.6 x 10 <sup>-</sup> B. 19.2 x C. 9.6cm D. 9.6mm <sup>3</sup> m 10 <sup>-5</sup> m 51. Water is falling on the blades of a turbine at a rate 6000Kg/min. The height of the fall is100m. What is the power gained by the turbine?					
A. 10KW	B. 6KW	C. 100KW	D. 600KW		
52. If momentum of a K.E. is that of	lpha-particle, neutron, pr	oton, and electron are th	e same, the minimum		
A. alpha-particle	B. neutron	C. proton	D. electron		
	while lifting a given load f the motor winds the cab B. 15 kW	-			
	electrons are accelerated mass to be respectively <i>e</i>	0 1			
A. $2eV/\sqrt{m}$	B. $\sqrt{(2eV)/m}$	<i>C</i> . 2 <i>m</i> / <i>eV</i>	<i>D.</i> $v^2/8em$		
55. A particle is movin acceleration is	ng on a circular track of r	adius 20 cm with a cons	tant speed of 6 m/s. Its		
A. 0 56. A satellite of the e	B. 180 m/s <sup>2</sup> arth is revolving in a circ v. If gravitational force so e will:	eular orbit	D. 36 m/s <sup>2</sup>		
<ul> <li>A. continue to move with the speed <i>v</i> along the original orbit</li> <li>B. move with the velocity <i>v</i> tangentially to the original orbit</li> <li>C. fall downward with increasing velocity</li> <li>D. ultimately come to rest somewhere on the original orbit</li> </ul>					
57. The kinetic energy K of a particle moving along a circle of radius R depends on the distance covered s as $K = as^2$ . The force acting on the part1cle is					
	ded Nobel Prize for his v t	· · · · ·	D. None of these		
59. One second is defi A. 1650763.73 period C. 1650763.73 period	s of the Krypton clock	B. 652189.63 periods D. 9192631770 period	• 1		

	energy and torque respe B. $MLT^2$ and $ML^2T^2$	ctively are C. $ML^2T^2$ and $MLT^2$	D. $MLT^2$ and $MLT^2$
61. When Benzene dia	zonium chloride reacts v	vith hypophosphorous ac	id, it produces
A. benzene	B. phenol	C. phenylphosphite	D. phenylphosphate
62. The reaction of alip	ohatic primary amine wit	th nitrous acid in cold pro	oduces
A. nitrile	B. alcohol	C. diazonium salt	D. secondary amine
A. acetamide 64. The aldol condensa	B. propionamide ation of acetaldehyde res		D. methyl cyanide
		C. CH <sub>3</sub> CH <sub>2</sub> CHOHCHC reagent at room tempera	CH <sub>3</sub> COOH
A. Butan-l-ol	B. Butan-2-ol	C. 2-Methyl propan-l-o	D. 2-Methyl propan-2- ol
66. The reaction with I	D <sub>2</sub> O, (CH <sub>3</sub> ) <sub>3</sub> CMgCl prod	uces	D. (CD <sub>3</sub> ) <sub>3</sub> COD
A. (CH <sub>3</sub> ) <sub>3</sub> CD	B. (CH <sub>3</sub> ) <sub>3</sub> CO	C. (CD <sub>3</sub> ) <sub>3</sub> CD	
A. 1-Butene	Icoholic potash, l-chloro B. 1-Butanol g agent during nitration o C. NO <sub>2</sub> <sup>-</sup> D. HNO <sub>3</sub>	C. 2-Butene	D. 2-Butanol
69. The number of sign	na and pi bonds in 1-but	en-3-yne are	D. 6 sigma and 4 pi
A. 5 sigma and 5 pi	B. 7 sigma and 3 pi	C. 8 sigma and 2 pi	
70. The most stable can	rbonium ion among the c	cations is	D. none of these
A. sec-butyl	B. ter-butyl	C. n-butyl	
71. How many opticall	y active stereo-isomers a	are possible for butane-2	, 3-diol?
A. 1	B. 2	C. 3	D. 4
72. B.P. and M.P. of in A. high	ert gases are B. low	C. very high	D. very low
73. [CO(NH <sub>3</sub> ) <sub>5</sub> Br] SO <sub>4</sub>	and [CO(NH <sub>3</sub> ) <sub>5</sub> SO <sub>4</sub> ] B	r are examples of which	type of isomerism ?
A. Linkage	B. Geometrical	C. Ionization	D. Optical
74. The valency of Cr	in the complex [Cr(H <sub>2</sub> O)	0 <sub>4</sub> Cl <sub>2</sub> ] <sup>+</sup> is	D. 5
A. 3	B. 1	C. 6	

75. In Nessler's reagen A. $Hg^+$ B. $Hg^{2+}$	t, the ion is C. $HgI_2^{2^-}$ D. $HgI_4^{2^-}$		
	2O, copper is co-ordinate s B. four water molecule		D. one water molecule
77. Which of the follow A. HCl	ving is a weak acid? B. HBr	C. HP	D. HI
78. When $SO_2$ is passe A. the solution turns bl C. $SO_2$ is reduced	d through acidified K <sub>2</sub> Cı ue	$C_2O_7$ solution, B. the solution is decol D. green $Cr_2(SO_4)_3$ is f	
79. Which of the follow A. H <sub>2</sub> O	wing has lowest boiling p B. H <sub>2</sub> S	ooint? C. H <sub>2</sub> Se	D. H <sub>2</sub> Te
<ul><li>80. Nitric oxide is prep</li><li>A. Fe</li><li>81. The laughing gas is</li><li>A. nitrous B. nitric</li><li>oxide oxide</li></ul>	ared by the action of dil. B. Cu C. nitrogen D. nitrogen trioxide pentaoxide	C. Zn	D. Sn
<ul><li>82. Ordinary glass is</li><li>A. sodium silicate</li><li>C. calcium and Sodiun</li></ul>	n silicate	B. calcium silicate D. copper silicate	
83. The chemical name	e of phosgene is		
A. Phosphene	B. Carbonyl chloride	C. Phosphorous oxychloride	D. Phosphorous trichloride
84. Which one of the for $A. BF_3$	ollowing is strongest Lev B. BCl <sub>3</sub>	vis acid? C. BBr <sub>3</sub>	D. BI <sub>3</sub>
85. Three centred bonc A. NH <sub>3</sub>	l is present in B. B <sub>2</sub> H <sub>6</sub>	C. BCl <sub>3</sub>	D. AlCl <sub>3</sub>
86. Plaster of Paris is A. CaSO <sub>4</sub> .H <sub>2</sub> O	B. CaSO <sub>4</sub> .2H <sub>2</sub> O	C. CaSO <sub>4</sub> .1/2 H <sub>2</sub> O	D. CaSO <sub>4</sub> .3/2 H <sub>2</sub> O
87. Rocky impurities p called A. flux B. gangue			
88. Free hydrogen is fo	C C		
A. acids	B. water	C. marsh gas	D. water gas
89. When zeolite, which	h is hydrated sodium alu	minium silicate, is treate	ed with hard water; the

sodium ions are excha	unged with		
A. $H^+$	B. $K^+$	C. $SO_4^{2-}$	D. Mg <sup>2+</sup>
metal deposited on ca	thode is $(Al = 27)$	gh aluminium chloride, t	
A. 0.27 g	B. 0.3 g	C. 2.7 g	D. 0.9 g
•	-	influence of an electric fi	eld is known as
A. Electro-osmosis	B. Brownian moveme	ent C. Cataphoresis	D. Dialysis
92. In a colloidal state	, particle size ranges fro	m	
A. 1 to 10 $A^{\circ}$	B. 20 to 50 $A^{\circ}$	C. 10 to 1000 $A^{\circ}$	D. 1 to 280 A <sup>o</sup>
A. 1.05 <sup>-1</sup>	irst order reaction is 69. B. 0.15 <sup>-1</sup> ion of a strong acid and	35. The value of rate con C. 0.015 <sup>-1</sup>	stant of the reaction is D. 0.0015 <sup>-1</sup>
A. 13.7 B. 9.6	C. 6 D. 11.4		
Kcal/mol Kcal/mol	Kcal/mol Kcal/mol		
95. In exothermic read	rtions		
A. $H_R = H_P$	B. $H_R > H_P$	$C. H_R < H_P$	D. None of the above
96. Which is a buffer			
A. $CH_3COOH +$	B. $CH_3COOH +$	C. $CH_3COOH + NH_4C$	Cl D. NaOH + NaCl
CH <sub>3</sub> COONa 97. The pH of 0.01 M	CH <sub>3</sub> COONH <sub>4</sub> solution of HCl is		
A. 1.0	B. 2.0	C. 10.0	D. 11.0
		action go fastest to compl	
A. $k = 10^2$	B. $k = 10^{-2}$	C. k = 10	D. k = 1
	· · · ·	ating will give 28 kg of <b>(</b>	
A. 1000 kg	B. 56 kg	C. 44 kg	D. 50 kg
100. The percentage o	f ovugan in NaOH is		
A. 40	B. 16	C. 18	D. 10
	of $CO_2$ and 14 g of $N_2$ ,		2.10
	fraction of $CO_2$ in the		
mixture?			
A. 1/5 B. 1/3	C. 1/2 D. 1/4		
102. The molarity of a	a solution of Na <sub>2</sub> CO <sub>2</sub> has	ving 5.3 g/250 ml of solu	tion is
A. 0.2 M	B. 2 M	C. 20 M	D. 0.02 M
	at 1 atm pressure. To co	mpress it to 1/2th of its in	nitial volume, pressure to

103. A gas is initially at 1 atm pressure. To compress it to 1/2th of its initial volume, pressure to be applied is

A. 1 atm	B. 4 atm	C. 2 atm	D. 1/4 atm
104. The value of <i>R</i> in c	palorie/degree/mole is		
A. 0.0831	B. 8.31	C. 8.31 x $10^7$	D. 1.987
	210101		2.1
105. Which of the follow	wing possesses zero resi	stance at 0 K?	
A. Conductors	B. Semi-conductors	C. Super-conductors	D. Insulators
106. CsCl has lattice of	• •		
A. ccp	B. fcc	C. bcc	D. hcp
107. In the reaction betw	ween sodium and chlorir	ne to form sodium chlori	de.
A. sodium atom is	B. sodium ion is	C. chlorine atom is	D. chloride ion is
reduced	reduced	reduced	reduced
108. Octahedral molecu	lar shape exists in		
<u>hybridisation.</u>	-		
A. $sp^3d$ B. $sp^3d^2$	C. $sp^3d^3$ D. $sp^2d^2$		
109 NH <sub>2</sub> and BF <sub>2</sub> form	an adduct readily becau	ise they form	
A. a co-ordinate bond	•	C. an ionic bond	D. a hydrogen bond
	D. a covalent bond	c. un tonic bond	D. u nyulogen bollu
110. Diagonal relationsl	hip exists between		
•	B. Na and Mg	C. K and Mg	D. Al and Mg
C	s the highest electro-neg	e	C
A. F	B. He	C. Ne	D. Na
7 <b>1.</b> I	D. IIC	0.110	D. Nu
112. Loss of a -particle	is equivalent to		
A. loss of two neutrons	-	B. loss of two protons of	only
	and loss of two protons	-	•
	-		
113. Stable compounds	in + 1 oxidation state an	e formed by	
A.B	B. Al	C. Ga	D. Th
114. Sodium hexametap	phosphate is used as		
11 il bouluit ilenuiteup			D. an iron exchange
A. a cleansing agent	B. an insecticide	C. a water softner	resin
115. The strongest acid	ic		
A. B.	C. D.		
$\text{ClO}_3(\text{OH})$ $\text{ClO}_2(\text{OH})$			
	50(011)2 502(011)2		
116. Which one among	the following pairs of ic	ons cannot be separated b	y H <sub>2</sub> S in dilute
hydrochloric acid?		-	
$A D^{3+} C^{4+}$	D $A1^{3+}$ $II_{\sim}^{2+}$	$C T n^{2+} C u^{2+}$	<b>D</b> $N^{2+}_{+}$ $C^{2+}_{+}$

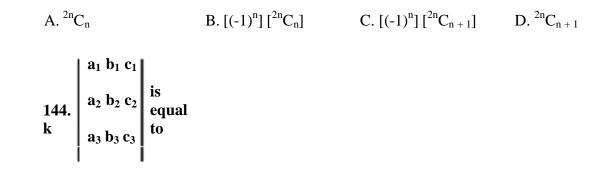
A.  $Bi^{3+}$ ,  $Sn^{4+}$  B.  $Al^{3+}$ ,  $Hg^{2+}$  C.  $Zn^{2+}$ ,  $Cu^{2+}$  D.  $Ni^{2+}$ ,  $Cu^{2+}$ 

117. The alkane would have only the primary and tertiary carbon is

A. Pentane	B. 2-methylbutane	C. 2, 2- dimethylpropane	D. 2, 3-dimethylbutane		
118. The product of re A. ethane	action of alcoholic silver B. ethene	nitrite with ethy1 bromi C. nitroethane	de is D. ethyl a1coho1		
119. Formy1 chloride	has not been so prepared nulation?	. Which one of the follow	wing can function as		
A. HCHO + HCl	B. HCOOCH <sub>3</sub> + HCl	C. CO + HCl	D. HCONH <sub>2</sub> + HCl		
120. Amongst the follo A. Benzylarnine 121. If the roots of $x^2$ - consecutive integers, th A. 4 B. 3	hen $b^2$ - 4c is equal to	npound is C. Acetanilide	D. p-Nitroaniline		
122. Condition that the perpendicular is	e two lines represented by	y the equation $ax^2 + 2hx$	$y + by^2 = 0$ to the		
A. $a = -b$	B. ab = 1	C. a = b	D. ab = -1		
123. If $A \subseteq B$ , then A A. B <sup>c</sup>	$ \cap  \textbf{B is equal to} \\ \textbf{B. A}^{c} $	С. В	D. A		
124. In order that the f A. $f(0) = 0$	function $f(x) = (x + 1)^{\cot x}$ B. $f(0) = e$	is continuous at x = 0, f C. f(0) = 1/e	(0) must be defined as D. none of the above		
125. The eccentricity of	of the ellipse $16x^2 + 7y^2 =$	= 112 is			
A. 4/3	<b>B.</b> 7/16	C. 3/√ 7	<b>D.</b> 3/4		
126. If $z_1, z_2, z_3$ are three complex numbers in A.P., then they lie on A. a circle B. an ellipse C. a straight line D. a parabola 127. If $[(a^2 + 1)^2]/(2a - i) = x + iy$ , then $x^2 + y^2$ is equal to					
A. $[(a^2 + B. [(a + 1)^4]/(4a^2 + 1)^2]/(4a^2 + 1)^2]/(4$	C. $[(a^2 - 1)^2]/(4a^2 - 1)^2$ D. none of the above $(1)^2$				
128. The vertices of a A. (3/2, 2)	triangle are (0, 0), (3, 0) B. (0, 0)	and (0, 4). Its orthocentro C. (1, 4/3)	e is at D. none of the above		
	of the conic $9x^2 - 16y^2 =$				
A. 5/4	B. 4/3	C. 4/5	D. √ 7		
120 The wartiese of a	(1)	and (2 0) The as and	ates of its outly control and		

130. The vertices of a triangle are (0, 3), (-3, 0) and (3, 0). The co-ordinates of its orthocentre are

A. (0, 2) B. (0, -3) C. (0, 3) D. (0, -2) 131. If t is the parameter for one end of a focal chord of the parabola  $y^2 = 4ax$ , then its length is C. a  $[t - (1/t)]^2$ D. a  $[t + (1/t)]^2$ B. a [t + (1/t)]A. a [t - (1/t)] 132. The value of  $\cos^2 \theta + \sec^2 \theta$  is always A. equal to 1 B. less than 1 D. greater than 1, but less than 2 C. greater than or equal to 2 133. The number of points of intersection of 2y = 1 and y = sin x,  $-2\pi \leq x \leq 2\pi$  is A. 2 **B**. 3 C. 4 D. 1 134. If  $\sin \theta_1 + \sin \theta_2 + \sin \theta_3 = 3$ , then  $\cos \theta_1 + \cos \theta_2 + \cos \theta_3 =$ **B**. 1 A.0 C. 2 D. 3 135. The number of solutions in  $0 \le x \le \pi/2$  of the equation  $\cos 3x \tan 5x = \sin 7x$  is A. 5 **B**. 7 C. 6 D. none of the above 136. One end of a diameter of the circle  $x^2 + y^2 - 4x - 2y - 4 = 0$  is (5, -6), the other end is B. (-9, -4) A. (4, -9) C. (4, 9) D. (9. -4) 137. The set of values of m for which both the roots of the equation  $x^2 - (m + 1)x + m + 4 = 0$  are real and negative consists of all m, such that A.  $-3 \ge m \text{ or } m \ge 5$ B.  $-3 < m \le 5$ C. -  $4 < m \le -3$ D.  $-3 < m \le -1$ 138. Let  $P_n(x) = 1 + 2x + 3x^2 + \dots + (n + 1)x^n$  be a polynomial such that n is even. Then the number of real roots of P(x) = 0 is A. 1 B. n C. 0 D. none of the above 139. The next term of the sequence 1, 3, 6, 10, ..... is A. 16 B. 13 C. 15 D. 14 140. If H is the harmonic mean between P and Q, then H/P + H/Q is B. PQ/(P + Q)D. none of the above A. (P + Q)/PQC. 2 141. A class is composed of two brothers and six other boys. In how many ways can all the boys be seated at a round table so that the two brothers are not seated besides each other? A. 4320 B. 3600 D. 1440 C. 720 142. The binomial coefficient of the 4th term in the expansion of  $(x - q)^5$  is C. 10 A. 15 B. 20 D. 5 143. For  $x \neq 0$ , the term independent of x in the expansion of  $(x - x^{-1})$  is equal to



	$a_1$	$b_1$	$kc_1$	
	a <sub>2</sub>	kb <sub>2</sub>	c <sub>2</sub>	
A.	ka <sub>3</sub>	<b>b</b> <sub>3</sub>	c <sub>3</sub>	

	ka <sub>1</sub>	$kb_1$	$kc_1$	
В.	ka <sub>2</sub>	kb <sub>2</sub>	$kc_2$	
D.	ka <sub>3</sub>	kb <sub>3</sub>	kc <sub>3</sub>	

C. 
$$\begin{vmatrix} ka_1 & b_1 & c_1 \\ ka_2 & b_2 & c_2 \\ ka_3 & b_3 & c_3 \end{vmatrix}$$

A. 2/3 B. 8/3 C. 16/3 D. 1/3

146. If 
$$|A| =$$

 a b c  
 x y z  
 p q r
 l
 and  $|B| =$ 
 $| q -b y |$ 
 -p a -x
 r -c z
 l
 , then

 A. | A | = 2 | B | B. | A | = | B | C. | A | = - | B | D. none of the above

147. Equation of the sphere with centre (1, -1, 1) and radius equal to that of sphere  $2x^2 + 2y^2 + 2z^2 - 2x + 4y - 6z = 1$  is A.  $x^2 + y^2 + z^2 - 2x + 2y - 2z + 1 = 0$  B.  $x^2 + y^2 + z^2 + 2x - 2y + 2z + 1 = 0$ C.  $x^2 + y^2 + z^2 - 2x + 2y - 2z - 1 = 0$  D. none of the above 148. Equation of the line passing through the point (1, 1, 1) and parallel to the plane 2x + 3y + 3z + 5 = 0 is A. (x - 1)/1 = (y - 1)/2 = B. (x - 1)/-1 = (y - 1)/1 (z - 1)/1 = (z - 1)/-1C. (x - 1)/3 = (y - 1)/2 = D. (x - 1)/2 = (y - 1)/3 = (z - 1)/1

149. If a, b, c are constants such that a and c are of opposite signs and r is the correlation coefficient between x and y, then the correlation coefficient between ax + b and cy + d is A. (a/c)rB.r C. - r D. (c/a)r150. From a deck of 52 cards, the probability of drawing a court card is A. 3/13 **B**. 1/4 C. 4/13 D. 1/13 151. A binomial probability distribution is symmetrical if p, the probability of success in a single trial, is A. > 1/2B. < 1/2C. < q, where q = 1 - p D. = 1/2152. The binomial distribution whose mean is 10 and S.D. is  $2\sqrt{2}$ is B.  $(4/5 + 1/5)^{1/50}$  C.  $(4/5 + 5/1)^{50}$ A.  $(4/5 + 1/5)^{50}$ D. none of the above 153. tan (cot  $^{-1}x$ ) is equal to B.  $\cot(\tan^{-1}x)$ C. tan x D. none of the above A. π /4 - x 154. If f(x) is an odd periodic function with period 2, then f(4) equals A. - 4 **B**. 4 C. 2 D. 0 155. The function  $f(x) = [(x^3 + x^2 - 16x + 20)]/(x - 2)$  is not defined for x = 2. In order to make f(x) continuous at x = 2, f(2) should be defined as A. 0 **B**. 1 C. 2 D. 3 156. Let f and g be differentiable functions satisfying g'(a) = 2, g(a) = b, and fog = 1 (identity function). Then f'(b) is equal to A. 0 C. 1/2 D. none of the above B. 2/3 157. A cone of maximum volume is inscribed in a given sphere. Then the ratio of the height of the cone to the diameter of the sphere is A. 3/4 B. 1/3 C. 1/4 D. 2/3 158. The function is decreasing in the interval A. -  $\infty < x < -10/3$ B.  $0 < x < \infty$ C. -3 < x < 3D. -10/3 < x < 0**159.** Suppose that f''(x) is continuous for all x and tf'(t) dt = 0,f(0) = f'(1). If then the value of f(1) is D. none of A. 3 B. 2 C. 9/2 the above 160. Integrating factor of differential equation  $\cos x (dy/dx) + y \sin x = 1$  is B. sec x A. sin x C. tan x D.  $\cos x$ 

161. If $\int_{0}^{\frac{\pi}{3}} \frac{dx/(1+4x^2)}{\pi/8}$ , then the value of a is			
Α.π /2	B. 1/2	C. π /4	D. 1
<ul><li>162. The maximum va</li><li>A. 2/e</li><li>163. If one root of the is 4, while the equation equal roots, then the va</li><li>A. 49/4 B. 4/49</li></ul>	B. $1/e$ equation $x^2 + px + 12 =$ $px^2 + px + q = 0$ has		D. e
164. The sum of the se A5/6	eries 1/2 + 1/3 + 1/6 + B1/2	to 9 terms is C. 1	D3/2
165. The sum of all tw A. 2475	o digit numbers, which a B. 2530	re odd is C. 4905	D. 5049
166. How many ten dia A. ${}^{10}C_1 + {}^{9}C_2$	git numbers can be forme B. 2 <sup>10</sup>	ed by using the digits 3 a C. ${}^{10}C_2$	nd 7 only? D. 10!
167. If x and y are real A. non-negative	and different and $u = x^2$ B. zero	$+4y^2 + 9z^2 - 6xyz - 3zx$ C. non-positive	- 2xy, then u is always D. none of the above
168. If a be a non-zero vector, then which of the following is correct?A. a. $a = 0$ B. $a \cdot a > 0$ C. $a \cdot a \ge 0$ D. $a \cdot a \le 0$ 169. If two vectors a and b are parallel and have equal magnitudes, thenB. they are not equalD. $a \cdot a \le 0$ A. they are equalB. they are not equalC. they may or may notD. they do not have the same direction			
170. In a triangle, the lengths of the two larger sides are 10 and 9 respectively. If the angles are in A.P., then the length of the third side can be A. $5 \pm \sqrt{6}$ B. $3\sqrt{3}$ C. 5 D. none of the above			
	$a + 4y + 6 = 0, \sqrt{2x} + \sqrt{3x}$ B. concurrent		
172. The pole of the straight line $9x + y - 28 = 0$ with respect to the circle $2x^2 + 2y^2 - 3x + 5y - 7 = 0$ is			
A. (3, 1)	B. (1, 3)	C. (3, -1)	D. (-3, 1)

173. If the sets A and B are defined as  $A = \{ (x, y) : y = e^x, x \in R \}, B = \{ (x, y) : y = x, x \in R \},$ then  $\mathbf{C}.\mathbf{A} \subseteq \mathbf{B}$  $\mathbf{D}. \mathbf{B} \subseteq \mathbf{A}$  $A. A \cup B = A$ **B**. **A**  $\cap$  **B** =  $\phi$ 174. The  $\frac{2a}{5}$  { f(x)/[f(x) + f(2a) value of the - x)] }dx is equal integral ō to D. none of C. 3a A.a B. 2a the above 175. The slope of the normal at the point ( $at^2$ , 2at) of the parabola  $y^2 = 4ax$  is B.t C. - t A. 1/t D. -1/t 176. If z is any complex number such that  $|z + 4| \le 3$ , then the greatest value of |z + 1| is A. 2 B. 6 C. 0 D. - 6 177. The equation  $\cos x + \sin x = 2$  has A. only one solution B. two solutions C. no solution D. infinite number of solutions 178. The most general value of  $\theta$ , which satisfies both the equations  $\tan \theta = -1$  and  $\cos \theta =$  $1/\sqrt{2}$  will be B.  $n\pi + (-1)^n (7\pi / 4)$  C.  $2n\pi + (7\pi / 4)$ A.  $n\pi + (7\pi / 4)$ D. none of the above 179. A spherical ball of radius r placed on the ground subtends an angle of  $60^{\circ}$  at a point A of the ground. Then the distance of the point A from the centre of the ball is D. none of C. 4r A. 3r B. 2r the above 180. In a triangle ABC,  $a^2 \cos 2B + b^2 \cos 2A + 2ab \cos (A - B)$  is equal to  $B.c^2$ A.c C. 2c D. none of the above