GATE question paper - Life Sciences 2009 (XL)

Duration: Three Hours Maximum Marks: 100

Read the following instructions carefully.

- This question paper contains 32 printed pages including pages for rough work. Please check all pages and report discrepancy, if any.
- Write your registration number, your name and name of the examination centre at the specific locations on the right half of the Optical Response Sheet (ORS)

Using HB pencil, darken the appropriate bubble under each digit of your registration number and the letters corresponding to your paper code.

- 4. All the questions in this question paper are of objective type.
- 5. Questions must be answered on Optical Response Sheet (ORS) by darkening the appropriate bubble (marked A, B, c, D) using HB pencil against the guestion number on the left hand side of the ORS. Each question has only one correct answer. In case you wish to change an answer, erase the old answer completely. More than one answer bubbled against a question will be taken as an incorrect response.
- The question paper contains six sections as listed below. Section H (Chemistry) is compulsory section. Choose two more sections from the remaining.

Section	Page No.	Section	Page No.
H. Chemistry	02	K. Botany	13
I. Biochemistry	06	L. Microbiology	18
J. Biotechnology	09	M. Zoology	21

Using HB pencil, mark the sections you have chosen by darkening the appropriate bubbles on the left hand side of the ORS provided. Make sure you have correctly bubbled the sections you have chosen. ORS will not be evaluated if this information is NOT marked.

- 7. There are 18 questions carrying 30 marks in Chemistry section paper (section H), which is compulsory. Question 1 through 6 are 1-mark questions, question 7 through 18 are 2-mark questions, Question 15 and 16 (1 pair) are common data questions with 2-marks each and question 17 and 18 (1 pair) are linked answer questions with 2-marks each. In the remaining sections, each section will contain 21 questions of total 35 marks. Questions 1 through 7 are 1-mark questions. Question 8 through 21 are 2-mark questions.
- Un-attempted questions will carry zero marks. 8.
- 9. Wrong answers will carry NEGATIVE marks. In section, H, for Q.1 to Q.6, 1/3 mark will be deducted for wrong answer. For Q.7 to Q.16, 2/3 mark will be deducted for wrong answer. The question pair (g. 17 and Q. 18) is questions with linked answers. there will be negative marks only for wrong answer to the first question of the linked answer question pair, i.e., for q. 17, 2/3 mark will be deducted for wrong answer. There is no negative marking for Q.18. In all other sections (section I through M), for q. 1 to Q. 7, 1/3 mark will be deducted for each wrong answer and for Q. 8 to Q.21, 2/3 mark will be deducted for each wrong answer.
- 10. Calculator (without data connectivity) is allowed in the examination hall.
- Charts, graph sheets or tables are NOT allowed in the examination hall. 11.
- 12. Rough work can be done on the question paper itself. Additionally, blank pages are given at the end of the question paper for rough work.
- Note from www.questiopapers.net.in: The page no. mentioned above has no relevance as we have utilized maximum page space to accommodate more questions.

H: Chemistry (Compulsory)

Useful data for Section H: Chemistry

 $R = 0.08315 \text{ L bar } K^{-1} \text{mol}^{-1}$; $F = 96490 \text{ C mol}^{-1}$; Atomic Numbers: Fe, 26; Co, 27

Q. 1-Q. 6 carry one mark each.

For a second order reaction, R \xrightarrow{k} p, the relation between half-life time (t_{1/2}) and the initial reactant concentration [R]₀ is

(A)
$$t_{1/2} = \frac{\ln 2}{k}$$

 $t_{1/2} = \frac{2}{k[R]_0}$ (C) $t_{1/2} = \frac{1}{k[R]_0^2}$ (D) $t_{1/2} = \frac{1}{k[R]_0}$

The reversible and irreversible entropy changes of a system on going from state'1' to state '2' are $\Delta S_{12}^{\text{rev}} > \Delta S_{12}^{\text{irrev}}$ respectively. The correct relationship between the two entropy change is

$$\Delta S_{12}^{irrev} > \Delta S_{12}^{rev}$$
 (B)

$$\Delta S_{12}^{irrev} < \Delta S_{12}^{rev}$$
 (C) $\Delta S_{12}^{irrev} = \Delta S_{12}^{rev}$ (D)

$$\Delta S_{12}^{irrev} > -\Delta S_{12}^{rev}$$

Among the following molecules the one that is planar is

(A)
$$(CH_3)_3N$$

$$[SO_4]^2$$

(C)

Among the following molecules the one that exhibits only one isomer is

 $[Fe(H_2O_3)_5OH]^{2+}$

 $Pt(NH_3)_2CI_2$ (B)

 $[CO_3]^{2-}$

(C) $[Pt(NH_3)_3CI_3]^+$ (D)

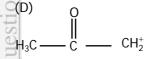
The most stable carbonation among the following is

CH₃OCH₂CH₂⁺

 $CH_3CH = CHCH_2^+$

 $[COCI_3(NH_3)_3]$

(C) CH₃CH₂⁺



The carboxylic acid with the lowest p^{Ka} value is 6.

> (A) C₆H₅CO_{2H}

(B) CH₃CO₂H

(C) p-CH₃O-C₆H₄CO₂H (D) p-CH₃-C₆H₄CO₂H

Q. 7 to Q.18 carry two marks each.

If the ground state ionization energy of the hydrogen atom is denoted by ε , then the energy required 7. to ionize an electron from the 3d energy level of the hydrogen atom is

2ε (A)

8. Given the following standard electrode potentials at 25°C:

$$\frac{1}{2}Fe^{2+} + e^{-} = \frac{1}{2}Fe(s); E^{0} = -0.440V$$

$$Fe^{3+} + e^{-} = Fe^{2+}; E^{0} = +0.771V$$

The standard electrode potential at 25°C for

$$Fe^{3+} + 3e^{-} = Fe(s)$$

is

-0.036 V(A)

(B) -0.331 V (C) -0.662 V (D) -2.422 V

Identify the major product Q formed in the following reaction: 9.

 $(CH_3)_2CHCH_2Br \rightarrow_{EtOH}$

(A) CH₃CH(OE_t)CH₂CH₃ (B) (CH₃)CHCH₂OE_t

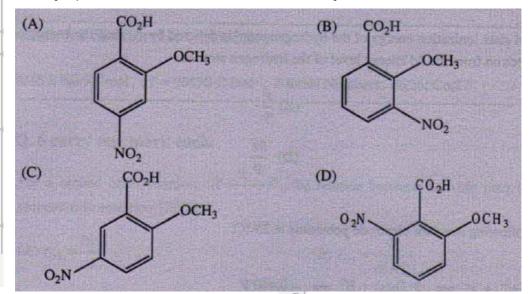
(C) $(CH_3)_3COE_t$ (D) $(CH_3)_2C=CH_2$

- The reaction of AgNO₃ with KCI in an aqueous environment leads to an insoluble product P. Treatment of P with an excess of KCI leads to its dissolution because of the formation of Q. P and Q respectively are
 - (A) AgCI and $[AgCI_2]^-$

(B) $[AgCI_2]^-$ and AgCI

(C) AgCI and $[AgCI_3]^-$

- (D) $[AgCl_2]^{-}$ and $[AgCl_3]^{2-}$
- 11. The major product formed in the nitration of o-methoxybenzoic acid is



12. Match the following:

http://www.questionpaper

P. [CoCl ₄] ²⁻	1. √15 BM
Q. [Fe(H ₂ O) ₆] ³⁺	2. oBM
R. [Fe(CN) ₆] ⁴⁺	3. √15 BM

(B)

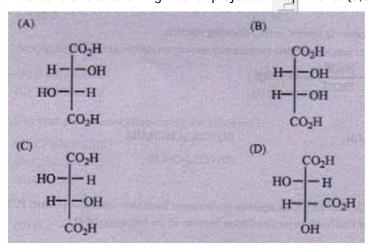
- (A) p-2, Q-3, R-1
- p-2, Q-1, R-3

(C)

p-1, Q-3, R-2

(D)

- p-3, Q-2, R-1
- **13.** Which one of the following Fischer projections represents (S,S)-tartaric acid?



- 14. Which one among the following compounds loses a proton most readily in a basic medium?
 - (A) Cyclopentadiene

(B) Cycloheptatriene

(C) Cyclopropene

(D) 1,3-Cyclohexadiene

Common Data Questions

Common Data for Questions 15 and 16:

For the reaction, P(g) = 2Q(g). The equilibrium constant with a standard state pressure of 1 bar is 0.25 . Assume ideal gas behaviour.

- 15. The total pressure (in bar) needed for 50% conversion of P into Q is
 - (A) 0.1250
- (B) 0.1875
- (C) 0.5000
- (D) 0.7500
- 16. The amount of P that will be converted to Q at a total pressure of 0.5 bar is approximately
 - (A) 13%
- (B) 25%
- (C) 33%
- (D) 55%

Linked Answer Questions

Statement for Linked Answer Question 17 and 18:

The reaction of BF₃ with NaBH₄ leads to the formation of a stable gaseous boron compound P. The compound P reacts with Me₃N to give Q.

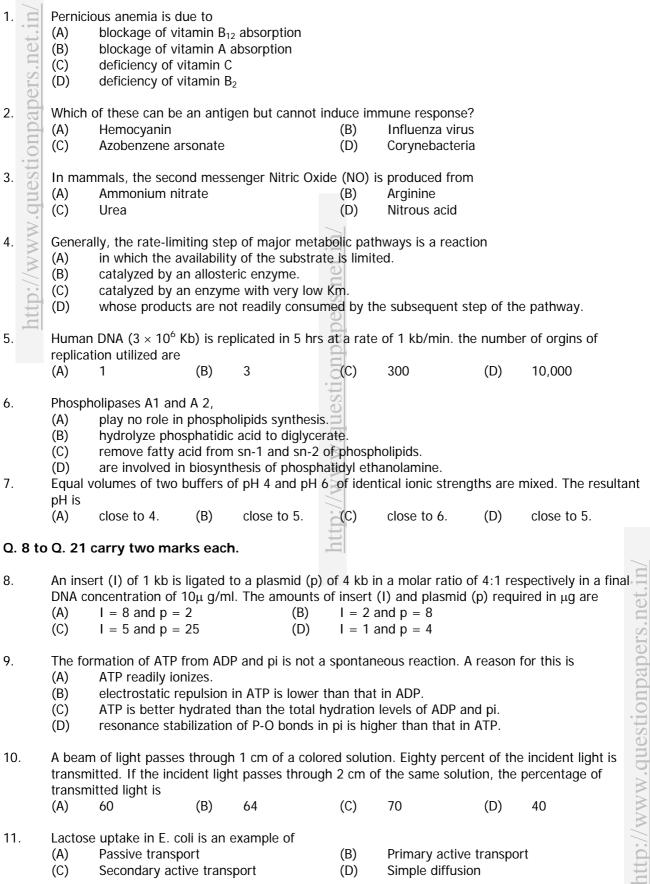
- 17. Identify P among the following:
 - (A) BH₃
- (B) $Na[B_3H_3]$
- (C) B_2H_6
- (D) B_4H_{10}

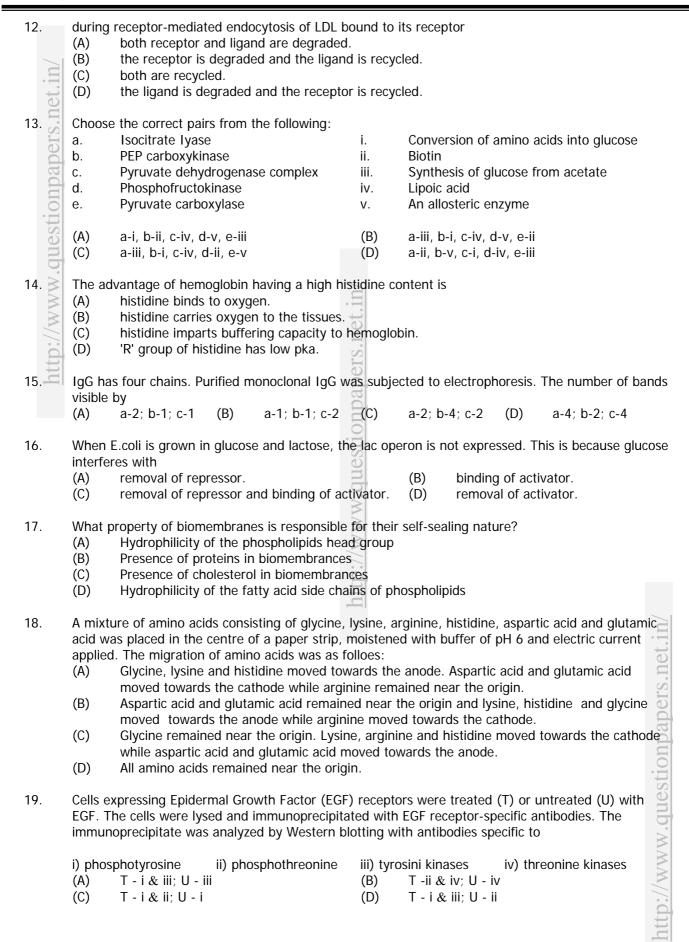
- 18. The compound Q is
 - (A) $\dot{B}H_3 \cdot NMe_3$
- (B) $B_2H_5 \cdot NMe_3$
- (C) $B_4H_9 \cdot NMe_3$
- (D) $BH_3 \cdot 2NMe_3$

END OF SECTION - H

http://www.questionpapers.net.in/

I: BIOCHEMISTRY





20. Which of following would be considered the longest feedback loop?

- (A) Reproductive steroid hormones...... aminergic neurons
- (B) Reproductive steroid hormones...... hypothalamic neurons
- (C) Reproductive steroid hormones...... pituitary gonadotrophs
- (D) Gonadotropin releasing hormone...... hypothalamic neurons

21. In myasthenia gravis, a neuromuscular disorder, the acetylcholine receptor becomes dysfunctional because http://www.questionpaper

- (A) the receptor is mutated.
- (B) antibodies to the receptor inhibit ligand binding.
- (C) of deficiency in acetylcholine transferase.
- (D) of excess of acctylcholine esterase.

END OF SECTION - I

S.

(A)

P, Q

(B)

R, S

J	: B	IOIE	CHNOLO	GY						
Q.	15	Q. 7 caı	rry one mark e	ach.						
1.	.net.		ethod used for p or more related			ensional st	ructure of a prot	ein from	known struct	ture(s)
	apers	(A) (C)	Multiple sequer Phylogeny	nce alig	nment	(B) (D)	Homology mod Docking	deling		
2.	dt		duce plants that		nozygous for a					
	Stior	(A) (C)	Protoplast cultue Anther and pole		ure	(B) (D)	Cell suspension Apical merister			
3.	dne	Restriction endonucleases from two different organisms that recognize the DNA sequence for cleavage are called								
	WW.	(A)	Isoschizomers	(B)	Isozymes	(C)	Concatamers	(D)	Palindromes	>
4.	X	(A)	es are involved i DNA replication		rocess of	(B)	Mutation and r	ecombir	nation	
	tp://	(C)	Antibody stnthe			(D)	Apootosis	COOMBI	idilon	
5.	ht		virus expression	system	is used to exp			l		
		(A) (C)	Mammals Batch bioreacto	nr.		(B) (D)	Plants Incubator			
		(0)	Dater bioreacte	Л		50)	incubator			
6.			re vessel in which				physiological co	nditions	, as well as ce	ell
		concen (A)	tration, are kept Cell concentrat		nt is known as	(B)	Biostat			
		(C)	Batch bioreacto			(D)	Incubator			
7.			esistant transger			.				
		(A) (C)	Cowpea trypsir Defective move			(B) (D)	Crystalline toxi Snowdrop lecti		n	
Q.	8 to	Q.21 c	arry two mark	s each		http				
8.		x are c P.	ommonly used a NPT gene	s repor	ter genes?					2
		Q.	Luciferase gene	Э						+
		R. S.	CFTR gene GFP gene							5
		3. (A)	Q, S	(B)	R, S	(C)	P, R	(D)	P, Q	7
9.			of the following				osate tolerant tra	ansgenic	plants?	
		P. Q	Transgenic plan				hat is not affacts	nd by als	mbosata	
	Q Transgenic plants produce an altered enzyme that is not affected by glyphosate. R. Transgenic plants sequester glyphosate in vacuoles.							a by gly	priosate.	• ‡

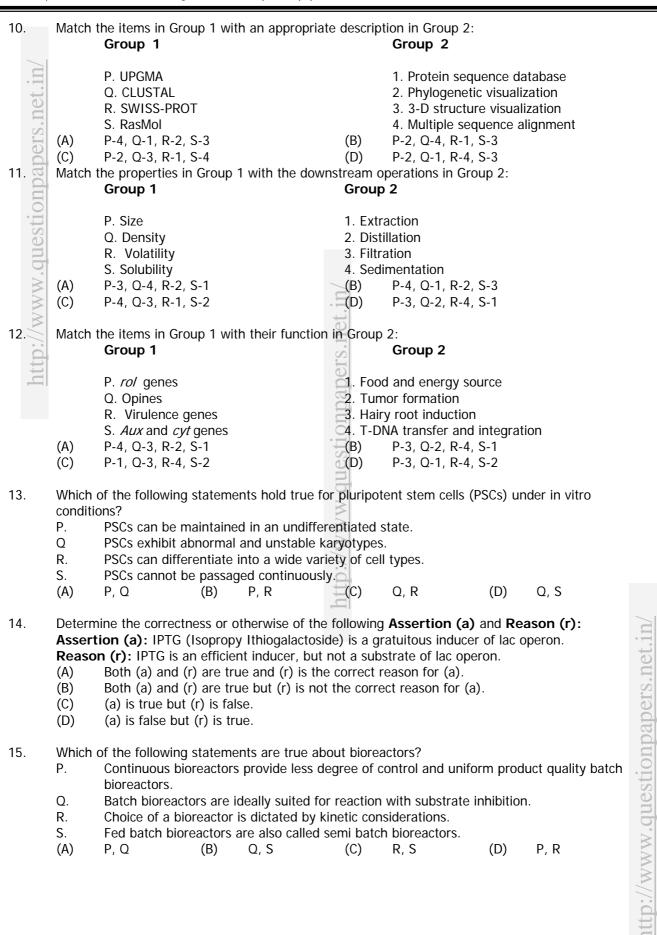
P, R

(D)

Transgenic plants overcome the inhibition of aromatic amino acid biosunthesis.

(C)

Q, S



16.	· · · · · · · · · · · · · · · · · · ·								
		Group 1				Group 2			
net.in/		P. DNA foctpri Q. Yeast two-l R. DNA finger	nybrid sy	stem		 Protein-prot VNTR DNA binding 			
		S. SAGE				4. Transcriptor			
apera	(A) (C)	P-1, Q-2, R-4, P-3, Q-4, R-1,			(B) (D)	P-3, Q-1, R-2, P-4, Q-2, R-1,			
http://www.questionpapers.net.in.	Assert the bac	ion (a): Bacteı terial cell cycle	rial growt	h is called syn	chronous \	Assertion (a) when majority o	f the cel	lls are in sam	
dne.	(A) (B)	Both (a) and (Both (a) and (r) are tru r) are tru	ue and (r) is th ue but (r) is no	e correct r	growing bacteria eason for (a). ct reason for (a		nriched med	ium
	(C)	(a) is true but	(r) is fals	se.	T				
18.	Match t	he products in	Group 1	with their poss	sible applic	ations in Group	2:		
		Group 1			.ne	Group 2			
ottto		P. Erythropoie			oapers	1. Blood clot			
		Q. Anti-fibrin			OE C	2. Binding and	l transpo	ort of iron	
		R. Collagenase S. Ttansferrin	2		Q	 Anaemia Animal cell 	senarati	on	
	(A)	P-3, Q-1, R-4,	S-2		C(B)	P-3, Q-4, R-1,		OH	
	(C)	P-2, Q-3, R-1,			(D)	P-2, Q-1, R-4,			
19.	Match t	he production Group 1	in Group	1 with their pr	oducer orç	ganisms in Grou Group 2	p 2:		
		P. Ethanol from Q. Probiotics R. Citric acid	m glucos	е	//ww/	 Aspergillus Leuconosto Saccharomy 	c mesen		
		S. Sauerkraut			0	4. Bifidobacter			
	(A) (C)	P-1, Q-3, R-2, P-3, Q-4, R-2,			(B) (D)	P-3, Q-2, R-4, P-1, Q-4, R-3,			
20.	activity trichlor	of 500 μCi / μr pacetic acid-ins cintillation cour	nol (1 μ0 oluble ra	$\dot{c}i = 2.2 \times 10^6 c$ dioactivity was	counts per s found to	TP as the labelle min). After 10 i be 692521 coun ³ H. The amoun	min incu its per m	bation, the nin as detern	nined in a
	(A)	15 nmol	(B)	105 nmol	(C)	150 nmol	(D)	50 nmol	ap
21.	μmol of mixture	f glucose per m with 1 ml of tl 0.6 μmol of glu	in in a 4º ne crude	% solution of L enzyme prepa	intner star ration cont	the amount of e ch at pH 4.5 an caining 8 mg pro ill be the specifi	id 60°C. otein and	If in a reacti d 9 ml of 4.4	ion 4%
	(A)	1 unit/mg pro	tein		(B)	1.5 units/mg p	rotein		9:
	(C)	0.25 units/mg			(D)	0.75 units/mg			WWW
				END OF SEC	CTION - J				

K: BOTANY

Q. 1 - Q. 7 carry one mark each.

(Å) Acid rain

(B) Photochemical smog

(C) Ozone hole

(D) Global warming

- 2. Phagotrophs are
 - (A) Organisms that feed in dead organic matter
 - (B) Organisms that absorb dissolved organic matter
 - (C) Organisms that ingest other organisms or particulate organic matter
 - (D) Organisms that manufacture food from simple inorganic substances
- 3. Identify the INCORRECT statement:
 - (A) 2, 4 dichloriohenoxyacetic acid (2, 4 -D) is the most commonly used chemical analogue of indole-3-acetic acid.
 - (B) In somatic embryogenesis, embryo initiation needs a high concentration of 2, 4 -D.
 - (C) Crown-gall disease depends on the presence of Ti-plasmid in Agrobacterium tumefaciens.
 - (D) Agrobacterium tumefaciens is responsible for hairy root formation in plants.
- 4. Choose the correct relation between Angstrom (OA) and nanometer (nm)
 - (A) $1^{\circ}A = 10^{1} \text{ nm}$ (B) $1^{\circ}A = 10^{-1} \text{ nm}$ (C) $1^{\circ}A = 10^{-2} \text{ nm}$ (D) $1^{\circ}A = 10^{2} \text{ nm}$
- 5. In hypogynous flower
 - (A) Ovary occupies the highest position on the thallamus
 - (B) Ovary may be partially sunken in the thallamus
 - (C) Ovary is completely sunken in the thallamus
 - (D) Ovary is naked
- 6. Fill up the blanks with appropriate matches.

The main axis of the inflorescence is known as —— and the stalk of the individual flower is called

- (A) Pedicel and Panicle
- (C) Pedicel and Peduncle
- (B) Panicle and Pedicel
- (D) Peduncle and Pedicel
- 7. Microorganisms responsible for nitrification
 - (A) Nitrosomonas and Nitrobacter
- (B) (D)
- Nastoc and Anabaena Clostridium and Pseudomonas

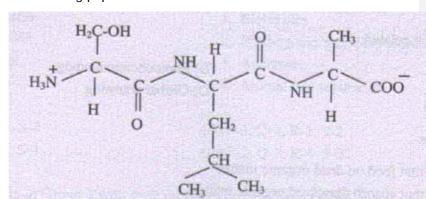
- Q. 8 to Q.21 carry two marks each.
- **8.** Identify the amino acids in the following peptide chain:

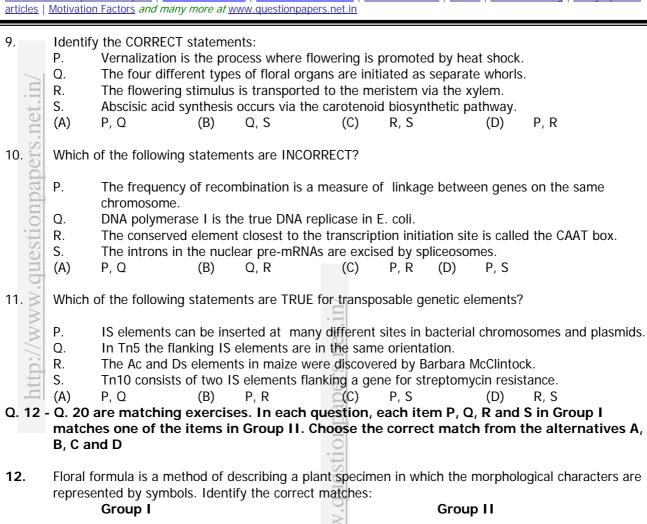
Rhizobium and Azotobacter

- P. Iso-Leu-Ala
- Q. Leu-Iso-Asn
- R. Ser-Leu-Ala
- S. Ser-Pro-Gln
- (A) P

(C)

- (B) Q
- (C) R
- (D) S





1 chi co	represented by symbols, identity the correct matches.									
	Group I	>	Group II							
P.	I		Zygomorphic							
Q.	\oplus	2 .	Male flower							
R.	CA	3.	Epipetalous							
S.	0	<u></u> 24.	Actinomorphic							
		5.	Superior ovary							
		6.	Inferior ovary							
(A)	P-2, Q-4, R-3, S-5	(B)	P-1, Q-6, R-5, S-2							

P-2, Q-4, R-3, S-6

	Identify the pathways where the following reactions occur:	
•	Group I P. Fatty acid + GTP + Co A Acyl-CoA + GDP + Pi	Group 11 1. Fatty acid synthesis
	Q. NH_3 + Glutamate + ATP \Longrightarrow Glutamine + AdP + Pi	2. Fatty acid oxidation
	R. Succinate + E - FAD \Longrightarrow Fumarate + E - FADH ₂	3. Oxidative phosphorylation
	S. Malonyl-S-CoA + ACP - SH \Longrightarrow Malonyl-S-ACP + CoA-SH	4. Citric acid cycle
		5. Gluconeogenesis
		6. Amino acid biosynthesis

(D)

(C)

P-6, Q-4, R-1, S-3

1.

14. Group I

- P. Ricinus communis
- Q. Jatropha curcasR. Pongamia pinnata
- S. Madhuca indica
- (A) P-4, Q-6, R-5, S-2
- (C) P-3, Q-6, R-5, S-2

R. F. S. M. (A) F. (C) F. Q. F. R. F. S. (A) F. (C) F. (C)

- P. Förster mechanism
- Q. Ping-Pong reaction
- R. Feed-back inhibition
- S. DNA recombination
- (A) P-5, Q-4, R-2, S-6
- (C) P-3, Q-4, R-2, S-6

16. **Group I**

- P. AG CT
- Q. GTPy PuAC CAPu PyTG
- R. C^VCGG GG C∧C
- S. AVAGOCTT TTC GAMA
- (A) P-1, Q-5, R-6, S-3
- (C) P-1, Q-6, R-4, S-2

17. **Group I**

- P. Linnaeus
- Q. William Roxburgh
- R. Bentham-Hooker
- S. Engler
- (A) P-2, Q-4, R-3, S-5
- (C) P-3, Q-1, R-5, S-6

Group II

- Rice bran oil
- 2. Mahua oil
- 3. Sun-flower oil
- 4. Castor oil
- 5. Karanja oil
- 6. Jatropha oil
- (B) P-1, Q-6, R-4, S-5
- (D) P-1, Q-6, R-5, S-2

Group II

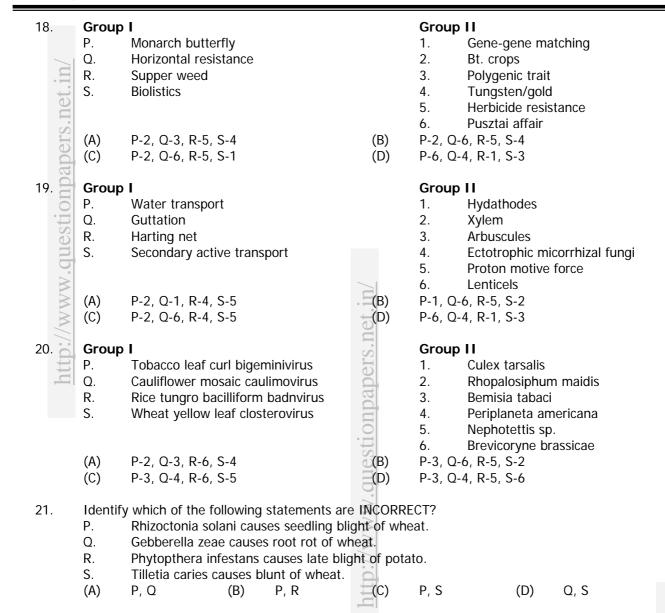
- 1. The initial product inhibit the initial reaction
- 2. The accumulation of end product inhibit the initial reaction
- 3. Process of exciton transfer in chromophore molecule
- 4. Double displacement reaction
- 5. Michaelis-Menten enzyme kinetics
- 6. Holliday model
- (B) P-2, Q-3, R-4, S-6
- (D) P-6, Q-5, R-4, S-1

Group II

- 1. EcoRI
- 2. Alul
 - 3. Hpall
 - 4. HindIII
- 5. Pstl
- 6. HincII
- (B) P-2, Q-6, R-3, S-4
- (D) P-1, Q-2, R-6, S-4

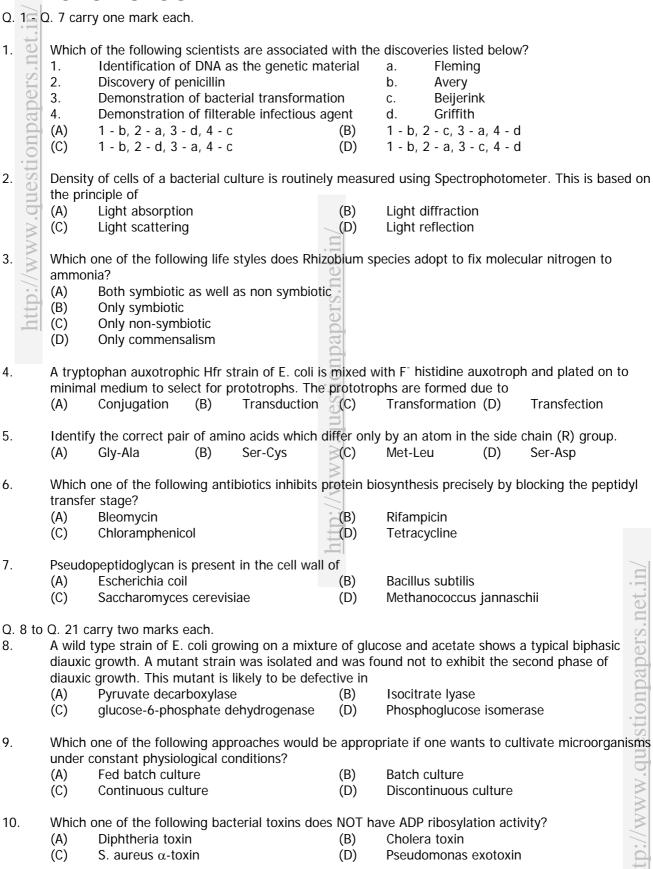
Group II

- 1. Flora Indica
- 2. Genera plantarum
- 3. Artificial-sexual system
- 4. Sero diagnostec system
- 5. Phylogenetic system
- 6. The families of flowering plants
- (B) P-1, Q-6, R-5, S-2
- (D) P-3, Q-1, R-2, S-5



END OF SECTION - K

L: MICROBIOLOGY



11.	 The high frequency of recombinants obtained from a cross between Hfr and F⁻ strains is because (A) the F factor integrated into the chromosome mediates transfer of large regions of chromosome into the recipient. (B) the strain bears high number of F pilli on its surface. (C) there are high number of copies of the F plasmid per cell. 									
.net	(C) (D)	it induces recor		-	-	d per cell.				
//www.questionpapers.net.ii	A silent (A) (B) (C) (D)	mutation is one results in a trur replaces an am does not chang changes the rea	ncated po ino acid e the am	with an equivalence	uence of th	e polypeptide		de		
13. Max.	The rotatory motion of bacterial flagellum is driven by (A) energy obtained through hydrolysis of ATP (B) trans-membrane electrochemical potential linked to proton pumping (C) direct uptake of extra-cellular nutrients (D) concentration gradient of nutrient sin the environment									
14.\\.\day		one of the follow osine in bacteria Chorismate α-ketobutyrate		abolic interme	ediates is in (B) (D)	volved in the Pantothenate Indole-3-pho	e	sis of Phenyl	alanin	ie
15.	allowed	A mixture conta to reanneal. Hove mixture? 0								ion:
16.	A tempo (A) (B) (C) (D)	erate phage differ can exhibit only can choose beth can exhibit only engages in lyso	/ lytic cyd ween lyti / lysogen	cle ic and lysoge ic cycle	nic cycle		nage			
17.	RecA is (A) (C)	a protein involv Recombinationa Nucleotide exci	al repair	air	(B) (D)	Mismatch rep				in/
18.	energy? (A) (C)	Vibrio fischeri Escherichia coli			(B) (D)	Bacillus subti Pseudomona	ilis s fluoresce	ence		pers.n
19.	Initial d 5 hours (A)	ensity of a cultu of incubation, v 10 ³	re of bac vhat seria (B)	cteria with a q al dilution wil 10 ⁴	generation I you have (C)	time of 30 mir to plate out to 10 ⁵	nutes was get ~ 100 (D)	1 × 10 ⁵ cells 0 colonies p 10 ⁶	s/ml. <i>F</i> er ml.'	? 0
20.	Effective because (A) (B) (C) (D)	e chemotherape Fungi have cell Fungi have beti Fungi are eucar Fungla pathoge	wall. ter mech ryotic cel	anisms to ina Is and their c	ctivate dru ellular mac	gs. hinery is simila	ar to that o	of the host.	ons	:://www.questi

21. Which one of the following is NOT the criterion for using ribosomal RNAs as evolutionary

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- (A) The large size of Ribosomal RNAs, given large number of possible sequence combinations.
- (B) Ribosomal RNAs are functionally invariant.
- (C) Ribosomal RNAs are universally distributed.
- (D) Ribosomal RNAs have enzyme activity

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chronometers?

END OF SECTION - L

M: ZOOLOGY

1.	et.i	In an n	rry one mark each. formal animal cell, the extra-chromoson Single stranded circular Doubled stranded circular	nal DNA is (B) (D)	Single stranded linear Double stranded linear
2.	npaper	Which (A)	of the following is NOT a somite derivat Cardiac muscle Cartilage	tive? (B) (D)	Skeletal muscle Tendons
3.	http://www.questionpapers.n	During (A) (C)	hibernation in a hibernating mammal, i Lower than normal state Higher than normal state	ts body te (B) (D)	emperature would be Same as normal state Fluctuate between high and low points
4.	WWW.0	Mendel (A) (C)	's principle of segregation means that t One of the paired alleles One quarter of the genes	he germ (B) (D)	cells (egg or sperm) always receive One pair of alleles Any pair of alleles
5.	http://	Cholera (A) (C)	a toxin acts by Activating a G-protein Activating synaptic transmission	(B)	Blocking an ion channel Blocking glycolysis
6.			of the following statements does NOT of ialized countries? Relatively small family size Rapid reproduction rate	(B) (D)	ne characteristics of human population in Relatively even age structure Delayed reproduction
7.		Sex-linl (A) (B) (C) (D)	kage refers to Inheritance of genes linked to sex det The ratio of genes on the autosome a Inheritance of genes carried on the se Linkage of genes present on the sex of	nd sex chr ex chromo	romosomes somes
Q.	8 t	o Q. 21 (carry two marks each.	tp:/	
8.			se in the emission of fossil fuels is though ease in atmospheric CO ₂ level would Increase the amount of sun light ente Increase the amount of infrared radial Absorb the infrared radiation reflected Insulate earth from cold breeze comin	ring the eation enteri	ng the earth
9.				he	arth e space nance their physical strength, are synthetic
10		an adu		hree key h rain horm rete ecdys e juvenine horacic gl	tars before molting into pupal stage, and into normones namely, brain hormone, ecdysone, ones is to sone e hormone and

- 11. Homology in anatomical parts helps in determining evolutionary kinship because
 - (A) Homologous body parts invariably perform similar functions
 - (B) Display evolutionary adaptations
 - (C) Undergo similar genetic changes
 - (D) Have common embryological origin
- 12. Which of the following does NOT explain the term "survival of the fittest"?
 - (A) Fittest animals leave higher number of progeny than those which perish.
 - (B) Fittest group of animals out populates its competitors.
 - (C) Fittest animals are best predators.
 - (D) Fittest group is rich in genetic variations.
- In a coral island in Atlantic Ocean, a natural calamity killed most of its population. The island is repopulated with the surviving individuals and their progeny. The present day population in the island shows high incidence of a rare recessive genetic disorder. What could be the most probable evolutionary force responsible for this phenomenon?
 - (A) Geographical isolation
 - (B) Genetic drift
 - (C) Selective advantage of the recessive disorder
 - (D) Natural selection
- 14. The human immune system is able to mount a response when it encounters a novel microorganism for the first time because
 - (A) White blood cells are able to change their antigen specificity depending upon the microorganism they interact with.
 - (B) Our body contains millions of different kinds of white blood cells, each with a unique type of antigen receptor.
 - (C) Bone marrow cells make different antigen receptor s depending upon the kind of invading microorganism.
 - (D) Bone marrow cells are ale to change their antigen specificity upon physical interaction with the microorganism.
- 15. Acetylcholinesterase is an enzyme that the degrades acetylcholine. What would be the effect of administration of an inhibitor of acetylcholinesterase on nerve transmission?
 - (A) No effect.
 - (B) Synaptic transmission will be prevented.
 - (C) Extra excitatory postsynaptic potentials would occur in the postsynaptic neuron.
 - (D) The presynaptic neuron will be inactivated.
- 16. Which one of the following comparisons between oogenesis and spermatogenesis in human in NO correct?
 - (A) FSH promotes development of both eggs as well as sperms.
 - (B) LH triggers ovulation in ovary and androgen production in testis.
 - (C) Primary oocytes and primary spermatocytes follow similar pattern of development through meiosis.
 - (D) An ovum is not produced until it fuses with the sperm whereas sperm is produced even in the absence of the ovum.
- 17. Match the parasitic species with their correct hosts.
 - P. Schistosoma mansoni
- 1. snails
- Q. Trichinella spiralis
- 2. human3. mosquito

R. Plasmodium

- 4. pig
- S. Taenia solium(A) P 1, Q 2, R 3, S 4
- (B) P 4, Q 3, R 2, S 1
- (C) P 2, Q 1, R 3, S 4
- (D) P 3, Q 4, R 1, S 2

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18.	Match	the embryonic cleavage pattern	s with the	e corresp	onding organisn	ns	
	P.	Isolecithal	1.	Click	0 0		
	Q.	Mesolecithal	2.	Drosop	hila		
2.	R.	Telolecithal	3.	Humar	1		
+	S.	Centrolecithal	4.	Frog			
9	(A)	P - 1, Q - 2, R - 3, S - 4		(B)	P - 2, Q - 1, R	- 3, S - ·	4
	(C)	P - 3, Q - 4, R - 1, S - 2		(D)		- 1, S - :	3
www.questionpapers.net.in	Which	combination of the following sta	atements	with reg	ard to gene exp	ression i	is true?
2	1.	Heteronuclear RNA represents	contigue	us segm	ent of genomic	DNA	
	2.	UTRs are part of heteronuclea	r RNA bu	t not of	mRNA.		
0	3.	UTRs are part of exons.					
. 1	4.	Translation start codon MUST	be in the	first exc	n.		
0	(A)	1 - true, 2 - false, 3 - false, 4	- false				
\equiv	(B)	1 - true, 2 - false, 3 - true, 4 -	false				
0	(C)	1 - true, 2 - true, 3 - false, 4 -	true				
\geq	(D)	1 - false, 2 - true, 3 - true, 4 -	true	2			
20.	Which	of the following combination of	statemer	nts regar	ding <u>M</u> aturation	<u>P</u> romoti	ng Factor (MPF) is
	TRUE?			TI I			
	1.	MPF in amphibians regulates of					
http://	2.	MPF in amphibians is compose					
		MPF in amphibians undergoes periodic degradation.					
	4.	MPF in amphibians undergoes	periodic	5)	,		
	(A)	1, 2, and 3 only		(B)	1 and 4 only		
	(C)	2 and 3 only		(D)	1, 2, 3 and 4		
				S			
21.		of the following combination of				ı is TRU	E?
	1.	It is marked by coordinated m		novemer	ıt.		
	2.	It results in the formation of c	0				
	3.	It results in the formation of t		n layers.			
	4.	it marks the end of neurulatio					
	(A)	1 and 3 only (B) 1 and	2 only	(C)	2 and 3 only	(D)	1, 2, 3 and 4

END OF THE QUESTION PAPRE