- 1. In the experiment to determine the speed of sound using a resonance column,
  - (A) prongs of the tuning fork are kept in a vertical plane
  - (B) prongs of the tuning fork are kept in a horizontal plane
  - (C) in one of the two resonances observed, the length of the resonating air column is close to the wavelength of sound in air
  - (D) in one of the two resonances observed, the length of the resonating air column is close to half of the wavelength of sound in air

**Answer** 





(A)



(D)

- 2. A student performs an experiment to determine the Young's modulus of a wire, exactly 2 m long, by Searle's method. In a particular reading, the student measures the extension in the length of the wire to be 0.8 mm with an uncertainty of ±0.05 mm at a load of exactly 1.0 kg. The student also measures the diameter of the wire to be 0.4 mm with an uncertainty of  $\pm 0.01$  mm. Take g = 9.8 m/s<sup>2</sup> (exact). The Young's modulus obtained from the reading is
  - (A)  $(2.0\pm0.3)\times10^{11} \text{ N/m}^2$
- (B)  $(2.0 \pm 0.2) \times 10^{11} \text{ N/m}^2$
- (C)  $(2.0\pm0.1)\times10^{11} \text{ N/m}^2$
- (D)  $(2.0 \pm 0.05) \times 10^{11} \text{ N/m}^2$

**Answer** 









(A)

(B)

(D)

OR









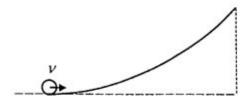
(B)

- 3. A particle moves in the X-Y plane under the influence of a force such that its linear momentum is  $\vec{p}(t) = A[\hat{i}\cos(kt) - \hat{j}\sin(kt)]$ , where A and k are constants. The angle between the force and the momentum is
  - (A) 0°
- (B) 30°
- (C) 45°
- (D) 90°

Answer



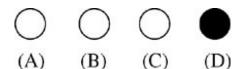
- (B) (D) (A)
- 4. A small object of uniform density rolls up a curved surface with an initial velocity  $\nu$ . It reaches up to a maximum height of  $\frac{3\nu^2}{4g}$  with respect to the initial position. The object is



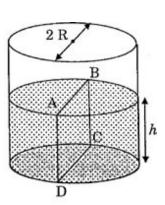
(A) ring

- (B) solid sphere
- (C) hollow sphere
- (D) disc

**Answer** 



5. Water is filled up to a height h in a beaker of radius R as shown in the figure. The density of water is  $\rho$ , the surface tension of water is T and the atmospheric pressure is  $P_0$ . Consider a vertical section ABCD of the water column through a diameter of the beaker. The force on water on one side of this section by water on the other side of this section has magnitude



(A) 
$$2P_0Rh + \pi R^2 \rho g h - 2RT$$

(A) 
$$|2P_0Rh + \pi R^2 \rho g h - 2RT|$$
 (B)  $|2P_0Rh + R\rho g h^2 - 2RT|$ 

(C) 
$$P_0 \pi R^2 + R \rho g h^2 - 2RT$$

(C) 
$$\left| P_0 \pi R^2 + R \rho g h^2 - 2RT \right|$$
 (D)  $\left| P_0 \pi R^2 + R \rho g h^2 + 2RT \right|$ 



- (A)
- (B)
- (D)

 A spherical portion has been removed from a solid sphere having a charge distributed uniformly in its volume as shown in the figure. The electric field inside the emptied space is



(A) zero everywhere

(B) non-zero and uniform

(C) non-uniform

(D) zero only at its center

**Answer** 









- (A)
- (B)
- (C)
- (D)
- 7. Positive and negative point charges of equal magnitude are kept at  $\left(0, 0, \frac{a}{2}\right)$  and  $\left(0, 0, \frac{-a}{2}\right)$ , respectively. The work done by the electric field when another positive point charge is moved from (-a, 0, 0) to (0, a, 0) is
  - (A) positive
  - (B) negative
  - (C) zero
  - (D) depends on the path connecting the initial and final positions



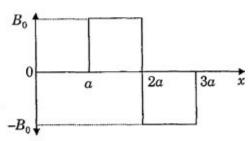


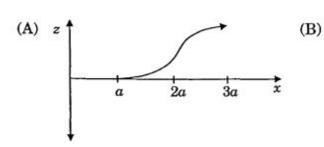


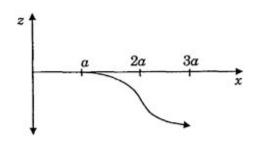


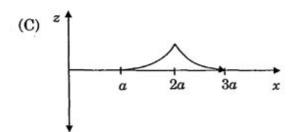
- (A)
- (B)
- (C)
- (D)

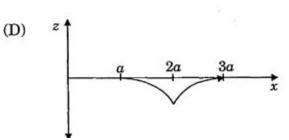
A magnetic field  $\vec{B} = B_0 \hat{j}$  exists in the region  $B_0$ 8. a < x < 2a and  $\vec{B} = -B_0\hat{j}$ , in the region 2a < x < 3a, where  $B_0$  is a positive constant. A positive point charge moving with a velocity  $\vec{v} = v_0 \hat{i}$ , where  $v_0$  is a positive constant, enters the magnetic field at x = a. The trajectory of the charge in this region can be like,



















9. Electrons with de-Broglie wavelength  $\lambda$  fall on the target in an X-ray tube. The cut-off wavelength of the emitted X-rays is

(A) 
$$\lambda_0 = \frac{2mc \lambda^2}{h}$$

(B) 
$$\lambda_0 = \frac{2h}{mc}$$

(C) 
$$\lambda_0 = \frac{2m^2c^2\lambda^3}{h^2}$$

(D) 
$$\lambda_0 = \lambda$$

Answer









(A)



(C)

(D)

#### 10. STATEMENT-1

If there is no external torque on a body about its center of mass, then the velocity of the center of mass remains constant.

#### because

### STATEMENT-2

The linear momentum of an isolated system remains constant.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True











### 11. STATEMENT-1

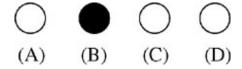
A cloth covers a table. Some dishes are kept on it. The cloth can be pulled out without dislodging the dishes from the table.

### because

### STATEMENT-2

For every action there is an equal and opposite reaction.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True



### STATEMENT-1

A vertical iron rod has a coil of wire wound over it at the bottom end. An alternating current flows in the coil. The rod goes through a conducting ring as shown in the figure. The ring can float at a certain height above the coil.

### because

#### STATEMENT-2

In the above situation, a current is induced in the ring which interacts with the horizontal component of the magnetic field to produce an average force in the upward direction.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True

#### Answer









(A)

(B)

(C)

### 13. STATEMENT-1

The total translational kinetic energy of all the molecules of a given mass of an ideal gas is 1.5 times the product of its pressure and its volume.

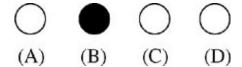
#### because

### STATEMENT-2

The molecules of a gas collide with each other and the velocities of the molecules change due to the collision.

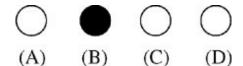
- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True

#### **Answer**

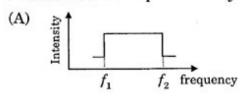


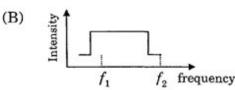
## The speed of sound of the whistle is

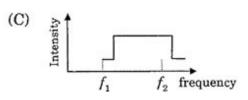
- (A) 340 m/s for passengers in A and 310 m/s for passengers in B
- (B) 360 m/s for passengers in A and 310 m/s for passengers in B
- (C) 310 m/s for passengers in A and 360 m/s for passengers in B
- (D) 340 m/s for passengers in both the trains



15. The distribution of the sound intensity of the whistle as observed by the passengers in train A is best represented by

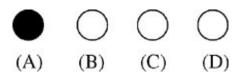








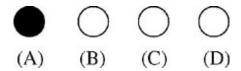
Answer



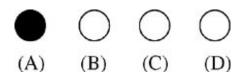
16. The spread of frequency as observed by the passengers in train B is

- (A) 310 Hz
- (B) 330 Hz
- (C) 350 Hz
- (D) 290 Hz

**Answer** 



- 17. Light travels as a
  - (A) parallel beam in each medium
  - (B) convergent beam in each medium
  - (C) divergent beam in each medium
  - (D) divergent beam in one medium and convergent beam in the other medium



18. The phases of the light wave at c, d, e and f are  $\phi_c$ ,  $\phi_d$ ,  $\phi_e$  and  $\phi_f$  respectively. It is given that  $\phi_c \neq \phi_f$ .

- (A)  $\phi_c$  cannot be equal to  $\phi_d$ 
  - (B)  $\phi_d$  can be equal to  $\phi_e$
- (C)  $(\phi_d \phi_f)$  is equal to  $(\phi_c \phi_e)$  (D)  $(\phi_d \phi_c)$  is not equal to  $(\phi_f \phi_e)$

**Answer** 







(A)

- (B)
- (D) (C)

19. Speed of light is

- (A) the same in medium-1 and medium-2
- (B) larger in medium-1 than in medium-2
- (C) larger in medium-2 than in medium-1
- (D) different at b and d

**Answer** 









(D)

(A)

(B)

(C)

20. Column I describes some situations in which a small object moves. Column II describes some characteristics of these motions. Match the situations in Column I with the characteristics in Column II and indicate your answer by darkening appropriate bubbles in the 4 × 4 matrix given in the ORS.

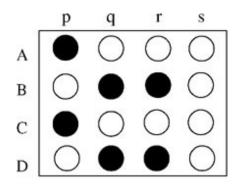
Column I

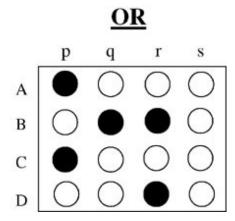
- (A) The object moves on the x-axis under a (p) The conservative force in such a way that its simpl "speed" and "position" satisfy  $v = c_1 \sqrt{c_2 x^2}$ , where  $c_1$  and  $c_2$  are positive constants.
- (B) The object moves on the x-axis in such a way that its velocity and its displacement from the origin satisfy v = -kx, where k is a positive constant.
- (C) The object is attached to one end of a mass-less spring of a given spring constant. The other end of the spring is attached to the ceiling of an elevator. Initially everything is at rest. The elevator starts going upwards with a constant acceleration a. The motion of the object is observed from the elevator during the period it maintains this acceleration.
- (D) The object is projected from the earth's surface vertically upwards with a speed  $2\sqrt{GM_e/R_e}$ , where,  $M_e$  is the mass of the earth and  $R_e$  is the radius of the earth. Neglect forces from objects other than the earth.

Column II

- (p) The object executes a simple harmonic motion.
- (q) The object does not change its direction.
- (r) The kinetic energy of the object keeps on decreasing.

(s) The object can change its direction only once.





21. Two wires each carrying a steady current *I* are shown in four configurations in **Column I**. Some of the resulting effects are described in **Column II**. Match the statements in **Column I** with the statements in **Column II** and indicate your answer by darkening appropriate bubbles in the 4×4 matrix given in the ORS.

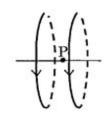
### Column I

- (A) Point P is situated --midway between the --wires.
- p) The magnetic fields (B) at P due to the currents

Column II

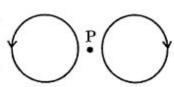
in the wires are in the same direction.

(B) Point P is situated at the mid-point of the line joining the centers of the circular wires, which have same radii.



(q) The magnetic fields (B) at P due to the currents in the wires are in opposite directions.

(C) Point P is situated at the mid-point of the line joining the centers of the circular wires, which have same radii.

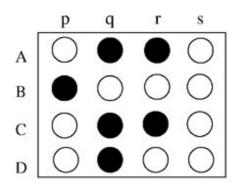


(r) There is no magnetic field at P.

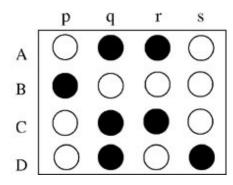
(D) Point P is situated at the common center of the wires.



(s) The wires repel each other.



## <u>OR</u>



22. Column I gives some devices and Column II gives some processes on which the functioning of these devices depend. Match the devices in Column I with the processes in Column II and indicate your answer by darkening appropriate bubbles in the 4×4 matrix given in the ORS.

### Column I

- (A) Bimetallic strip
- (B) Steam engine
- (C) Incandescent lamp
- (D) Electric fuse

### Column II

- (p) Radiation from a hot body
- (q) Energy conversion
- (r) Melting
- (s) Thermal expansion of solids

23.Consider a titration of potassium dichromate solution with acidified Mohr's salt solution using diphenylamine as indicator. The number of moles of Mohr's salt required per mole of dichromate is

- (A) 3
- (B) 4
- (C) 5
- (D) 6

**Answer** 

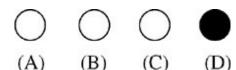


- (A)
- (B)
- (C)
- (D)

24. Among the following metal carbonyls, the C-O bond order is lowest in

- (A)  $[Mn(CO)_6]^+$  (B)  $[Fe(CO)_5]$  (C)  $[Cr(CO)_6]$
- (D) [V(CO)<sub>6</sub>]

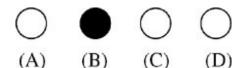
**Answer** 



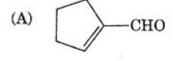
A solution of a metal ion when treated with KI gives a red precipitate which dissolves 25.in excess KI to give a colourless solution. Moreover, the solution of metal ion on treatment with a solution of cobalt(II) thiocyanate gives rise to a deep blue crystalline precipitate. The metal ion is

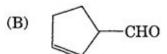
- (A) Pb2+
- (B) Hg<sup>2+</sup> (C) Cu<sup>2+</sup>
- (D) Co<sup>2+</sup>

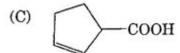
**Answer** 



Cyclohexene on ozonolysis followed by reaction with zinc dust and water gives 26.compound E. Compound E on further treatment with aqueous KOH yields compound F. Compound F is







**Answer** 







- (A)
- (B)
- (C)
- 27. The number of stereoisomers obtained by bromination of trans-2-butene is
  - (A) 1
- (B) 2

- (C) 3.
- (D) 4

Answer





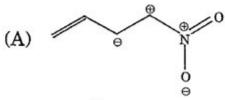


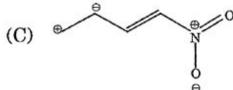


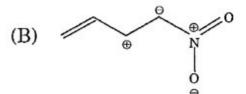
- (A)
- (B)
- (C)
- (D)

(D)

28. Among the following, the least stable resonance structure is







(D) 
$$\oplus$$
  $\delta$ 

Answer









- (A)
- (B)
- (C)
- (D)
- 29. A positron is emitted from <sup>23</sup><sub>11</sub>Na. The ratio of the atomic mass and atomic number of the resulting nuclide is
  - (A) 22/10

(B) 22/11

(C) 23/10

(D) 23/12

Answer









(D)

(A)

- (B)
- (C)
- 30. For the process  $H_2O(l)$  (1 bar, 373 K)  $\rightarrow$   $H_2O(g)$  (1 bar, 373 K), the correct set of thermodynamic parameters is
  - (A)  $\Delta G = 0$ ,  $\Delta S = +ve$

(B)  $\Delta G = 0$ ,  $\Delta S = -ve$ 

(C)  $\Delta G = +ve$ ,  $\Delta S = 0$ 

(D)  $\Delta G = -ve$ ,  $\Delta S = +ve$ 

Answer









(A)

- (B)
- (C)
- (D)
- 31. Consider a reaction aG + bH → Products. When concentration of both the reactants G and H is doubled, the rate increases by eight times. However, when concentration of G is doubled keeping the concentration of H fixed, the rate is doubled. The overall order of the reaction is
  - (A) 0
- (B) 1
- (C) 2
- (D) 3









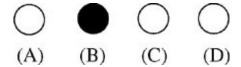
- (A)
- (B)
- (C)
- (D)

32. STATEMENT-1: Alkali metals dissolve in liquid ammonia to give blue solutions.

### because

- STATEMENT-2: Alkali metals in liquid ammonia give solvated species of the type  $[M(NH_3)_n]^+$  (M = alkali metals).
- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True; Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True

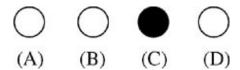
#### **Answer**



STATEMENT-1: Glucose gives a reddish-brown precipitate with Fehling's solution.

#### because

- STATEMENT-2: Reaction of glucose with Fehling's solution gives CuO and gluconic acid.
- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True



 STATEMENT-1: Molecules that are not superimposable on their mirror images are chiral.

#### because

STATEMENT-2: All chiral molecules have chiral centres.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True

#### **Answer**

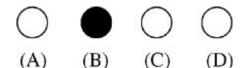
$\bigcirc$	$\bigcirc$		$\bigcirc$
(A)	(B)	(C)	(D)

STATEMENT-1: Band gap in germanium is small.

#### because

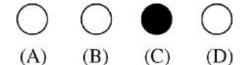
STATEMENT-2: The energy spread of each germanium atomic energy level is infinitesimally small.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True



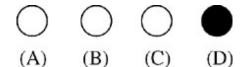
- 36. Among the following, identify the correct statement.
  - (A) Chloride ion is oxidised by O<sub>2</sub> (B) Fe<sup>2+</sup> is oxidised by iodine
  - (C) Iodide ion is oxidised by chlorine (D) Mn2+ is oxidised by chlorine

#### **Answer**



- 37. While Fe<sup>3+</sup> is stable, Mn<sup>3+</sup> is not stable in acid solution because
  - (A) O<sub>2</sub> oxidises Mn<sup>2+</sup> to Mn<sup>3+</sup>
  - (B) O2 oxidises both Mn2+ to Mn3+ and Fe2+ to Fe3+
  - (C) Fe3+ oxidises H2O to O2
  - (D) Mn3+ oxidises H2O to O2

### Answer

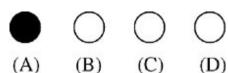


- 38. Sodium fusion extract, obtained from aniline, on treatment with iron(II) sulphate and H<sub>2</sub>SO<sub>4</sub> in presence of air gives a Prussian blue precipitate. The blue colour is due to the formation of
  - (A)  $\operatorname{Fe}_{4}[\operatorname{Fe}(\operatorname{CN})_{6}]_{3}$

(B)  $\operatorname{Fe}_{3}[\operatorname{Fe}(\operatorname{CN})_{6}]_{2}$ 

(C)  $\operatorname{Fe_4[Fe(CN)_6]_2}$ 

(D)  $\operatorname{Fe}_{3}[\operatorname{Fe}(\operatorname{CN})_{6}]_{3}$ 



39. Which one of the following reagents is used in the above reaction?

(A)  $aq.NaOH + CH_3Cl$ 

(B)  $aq. NaOH + CH_2Cl_2$ 

(C) aq. NaOH + CHCl<sub>3</sub>

(D) aq. NaOH + CCl<sub>4</sub>

Answer



(A) (B)

(C) (D)

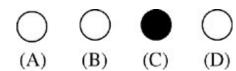
40. The electrophile in this reaction is

(A) :CHCl

(B) +CHCl<sub>2</sub>

(C) :CCl<sub>2</sub>

(D) ·CCl<sub>3</sub>



# 41. The structure of the intermediate I is

(A) 
$$O^{\ominus}$$
  $N_{a}^{\oplus}$   $CH_{2}Cl$   $CH_{3}$ 

(C) 
$$O$$
  $Na$   $CCl_3$   $CH_3$ 

(D) 
$$\overset{\circ}{\overset{\circ}{\overset{\circ}{\operatorname{Na}}}}$$
  $\overset{\circ}{\overset{\circ}{\overset{\circ}{\operatorname{CH_2OH}}}}$   $\overset{\circ}{\overset{\circ}{\overset{\circ}{\overset{\circ}{\operatorname{CH_2OH}}}}}$ 

Answer



(A) (B) (C) (D)

42. Match the reactions in Column I with nature of the reactions/type of the products in Column II. Indicate your answer by darkening the appropriate bubbles of the 4×4 matrix given in the ORS.

## Column I

(A) 
$$O_2^- \to O_2 + O_2^{2-}$$

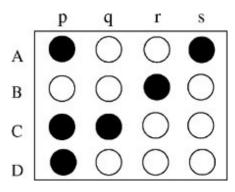
(B) 
$$\operatorname{CrO}_4^{2-} + \operatorname{H}^+ \rightarrow$$

(C) 
$$MnO_4^- + NO_2^- + H^+ \rightarrow$$

(D) 
$$NO_3^- + H_2SO_4 + Fe^{2+} \rightarrow$$

### Column II

- (p) redox reaction
- (q) one of the products has trigonal planar structure
- (r) dimeric bridged tetrahedral metal ion
- (s) disproportionation

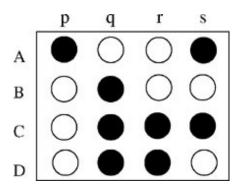


43. Match the compounds/ions in Column I with their properties/reactions in Column II.
Indicate your answer by darkening the appropriate bubbles of the 4×4 matrix given in the ORS.

### Column I

- (A) C<sub>6</sub>H<sub>5</sub>CHO
- (B)  $CH_3C \equiv CH$
- (C) CN-
- (D) I-

#### Answer



### Column II

- (p) gives precipitate with 2,4-dinitrophenylhydrazine
- (q) gives precipitate with AgNO3
- (r) is a nucleophile
- (s) is involved in cyanohydrin formation

44. Match the crystal system/unit cells mentioned in Column I with their characteristic features mentioned in Column II. Indicate your answer by darkening the appropriate bubbles of the 4 × 4 matrix given in the ORS.

Column I

- (A) simple cubic and face-centred cubic
- (B) cubic and rhombohedral
- (C) cubic and tetragonal
- (D) hexagonal and monoclinic

### Column II

- (p) have these cell parameters a=b=c and  $\alpha=\beta=\gamma$
- (q) are two crystal systems
- (r) have only two crystallographic angles of 90°
- (s) belong to same crystal system

#### **Answer**

	p	q	r	S
A		0	0	
В			0	$\bigcirc$
С	0		$\bigcirc$	$\bigcirc$
D	0			$\bigcirc$

- 45. Let O(0, 0), P(3, 4), Q(6, 0) be the vertices of the triangle OPQ. The point R inside the triangle OPQ is such that the triangles OPR, PQR, OQR are of equal area. The coordinates of R are
  - (A)  $\left(\frac{4}{3}, 3\right)$

(B)  $\left(3, \frac{2}{3}\right)$ 

(C)  $\left(3, \frac{4}{3}\right)$ 

(D)  $\left(\frac{4}{3}, \frac{2}{3}\right)$ 









- (A)
- (B)
- (C)
- (D)

46. If |z| = 1 and  $z \neq \pm 1$ , then all the values of  $\frac{z}{1-z^2}$  lie on

(A) a line not passing through the origin

(B) 
$$|z| = \sqrt{2}$$

- (C) the x-axis
- (D) the y-axis

Answer

 $\circ \circ \circ \bullet$ 

(A) (B) (C) (D)

47. Let  $E^c$  denote the complement of an event E. Let E, F, G be pairwise independent events with P(G) > 0 and  $P(E \cap F \cap G) = 0$ . Then  $P(E^c \cap F^c \mid G)$  equals

(A)  $P(E^c) + P(F^c)$ 

(B)  $P(E^c) - P(F^c)$ 

(C)  $P(E^c)-P(F)$ 

(D)  $P(E) - P(F^c)$ 

Answer

 $\bigcirc$   $\bigcirc$   $\bigcirc$   $\bigcirc$ 

(A) (B) (C) (D)

48.  $\frac{d^2x}{dy^2}$  equals

(A)  $\left(\frac{d^2y}{dx^2}\right)^{-1}$ 

(B)  $-\left(\frac{d^2y}{dx^2}\right)^{-1} \left(\frac{dy}{dx}\right)^{-3}$ 

(C)  $\left(\frac{d^2y}{dx^2}\right)\left(\frac{dy}{dx}\right)^{-2}$ 

(D)  $-\left(\frac{d^2y}{dx^2}\right)\left(\frac{dy}{dx}\right)^{-3}$ 

**Answer** 









- (A
- (B
- (C)
- (D)
- 49. The differential equation  $\frac{dy}{dx} = \frac{\sqrt{1-y^2}}{y}$  determines a family of circles with
  - (A) variable radii and a fixed centre at (0,1)
  - (B) variable radii and a fixed centre at (0, -1)
  - (C) fixed radius 1 and variable centres along the x-axis
  - (D) fixed radius 1 and variable centres along the y-axis

Answer









(A

- (B)
- (C)
- (D)
- 50. Let  $\vec{a}$ ,  $\vec{b}$ ,  $\vec{c}$  be unit vectors such that  $\vec{a} + \vec{b} + \vec{c} = \vec{0}$ . Which one of the following is correct?
  - (A)  $\overrightarrow{a} \times \overrightarrow{b} = \overrightarrow{b} \times \overrightarrow{c} = \overrightarrow{c} \times \overrightarrow{a} = \overrightarrow{0}$
  - (B)  $\overrightarrow{a} \times \overrightarrow{b} = \overrightarrow{b} \times \overrightarrow{c} = \overrightarrow{c} \times \overrightarrow{a} \neq \overrightarrow{0}$
  - (C)  $\overrightarrow{a} \times \overrightarrow{b} = \overrightarrow{b} \times \overrightarrow{c} = \overrightarrow{a} \times \overrightarrow{c} \neq \overrightarrow{0}$
  - (D)  $\vec{a} \times \vec{b}$ ,  $\vec{b} \times \vec{c}$ ,  $\vec{c} \times \vec{a}$  are mutually perpendicular









- (A)
- (B)
- (C)
- (D)

Let ABCD be a quadrilateral with area 18, with side AB parallel to the side CD and 51. AB = 2 CD. Let AD be perpendicular to AB and CD. If a circle is drawn inside the quadrilateral ABCD touching all the sides, then its radius is

- (A) 3
- (B) 2
- (C)  $\frac{3}{2}$
- (D) 1

**Answer** 









- (A)
- (B)
- (D)

Let  $f(x) = \frac{x}{(1+x^n)^{1/n}}$  for  $n \ge 2$  and  $g(x) = \underbrace{(f \circ f \circ \cdots \circ f)}_{f \text{ occurs } n \text{ times}} (x)$ . Then  $\int x^{n-2}g(x) dx$ equals

- (A)  $\frac{1}{n(n-1)} (1+nx^n)^{1-\frac{1}{n}} + K$  (B)  $\frac{1}{n-1} (1+nx^n)^{1-\frac{1}{n}} + K$ 

  - (C)  $\frac{1}{n(n+1)} (1+nx^n)^{1+\frac{1}{n}} + K$  (D)  $\frac{1}{n+1} (1+nx^n)^{1+\frac{1}{n}} + K$

**Answer** 









- (A)
- (B)
- (C)
- (D)

The letters of the word COCHIN are permuted and all the permutations are 53. arranged in an alphabetical order as in an English dictionary. The number of words that appear before the word COCHIN is

- (A) 360
- (B) 192
- (C) 96
- (D) 48







- (A)
- (B)
- (C)
- (D)

54. Consider the planes 3x - 6y - 2z = 15 and 2x + y - 2z = 5.

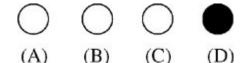
STATEMENT-1: The parametric equations of the line of intersection of the given planes are x = 3 + 14t, y = 1 + 2t, z = 15t.

### because

STATEMENT-2: The vector  $14\hat{i} + 2\hat{j} + 15\hat{k}$  is parallel to the line of intersection of given planes.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True

#### Answer

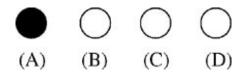


55. STATEMENT-1: The curve  $y = \frac{-x^2}{2} + x + 1$  is symmetric with respect to the line x = 1.

#### because

STATEMENT-2: A parabola is symmetric about its axis.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True



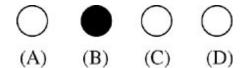
56. Let  $f(x) = 2 + \cos x$  for all real x.

STATEMENT-1: For each real t, there exists a point c in  $[t, t + \pi]$  such that f'(c) = 0.

### because

STATEMENT-2:  $f(t) = f(t + 2\pi)$  for each real t.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True



57. Lines  $L_1: y-x=0$  and  $L_2: 2x+y=0$  intersect the line  $L_3: y+2=0$  at P and Q, respectively. The bisector of the acute angle between  $L_1$  and  $L_2$  intersects  $L_3$  at R.

STATEMENT-1: The ratio PR:RQ equals  $2\sqrt{2}:\sqrt{5}$ .

### because

STATEMENT-2: In any triangle, bisector of an angle divides the triangle into two similar triangles.

- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- (B) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1
- (C) Statement-1 is True, Statement-2 is False
- (D) Statement-1 is False, Statement-2 is True

**Answer** 



(A) (B) (C) (D)

58. Which one of the following statements is correct?

(A) 
$$G_1 > G_2 > G_3 > \cdots$$

(B) 
$$G_1 < G_2 < G_3 < \cdots$$

(C) 
$$G_1 = G_2 = G_3 = \cdots$$

(D) 
$$G_1 < G_3 < G_5 < \cdots$$
 and  $G_2 > G_4 > G_6 > \cdots$ 

Answer



(A) (B) (C) (D)

59. Which one of the following statements is correct?

(A) 
$$A_1 > A_2 > A_3 > \cdots$$

(B) 
$$A_1 < A_2 < A_3 < \cdots$$

(C) 
$$A_1 > A_3 > A_5 > \cdots$$
 and  $A_2 < A_4 < A_6 < \cdots$ 

(D) 
$$A_1 < A_3 < A_5 < \cdots$$
 and  $A_2 > A_4 > A_6 > \cdots$ 

Answer



(A) (B) (C) (D)

60. Which one of the following statements is correct?

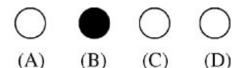
- (A)  $H_1 > H_2 > H_3 > \cdots$
- (B)  $H_1 < H_2 < H_3 < \cdots$
- (C)  $H_1 > H_3 > H_5 > \cdots$  and  $H_2 < H_4 < H_6 < \cdots$
- (D)  $H_1 < H_3 < H_5 < \cdots$  and  $H_2 > H_4 > H_6 > \cdots$

M<sub>61-63</sub>: Paragraph for Question Nos. 61 to 63

If a continuous function f defined on the real line  $\mathbf{R}$ , assumes positive and negative values in  $\mathbf{R}$  then the equation f(x) = 0 has a root in  $\mathbf{R}$ . For example, if it is known that a continuous function f on  $\mathbf{R}$  is positive at some point and its minimum value is negative then the equation f(x) = 0 has a root in  $\mathbf{R}$ .

Consider  $f(x) = ke^x - x$  for all real x where k is a real constant.

Answer



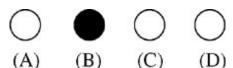
61. The line y = x meets  $y = ke^x$  for  $k \le 0$  at

(A) no point

(B) one point

(C) two points

(D) more than two points



The positive value of k for which  $ke^x - x = 0$  has only one root is 62.

- (B) 1

(C) e

(D) log 2

**Answer** 



- (A) (B)
- (C) (D)

For k > 0, the set of all values of k for which  $ke^x - x = 0$  has two distinct roots is 63.

- (A)  $\left(0, \frac{1}{e}\right)$  (B)  $\left(\frac{1}{e}, 1\right)$  (C)  $\left(\frac{1}{e}, \infty\right)$
- (D) (0, 1)

**Answer** 



- (A)
- (B)
- (D)

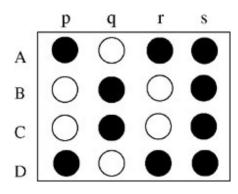
Let  $f(x) = \frac{x^2 - 6x + 5}{x^2 - 5x + 6}$ .

Match the expressions/statements in Column I with expressions/statements in Column II and indicate your answer by darkening the appropriate bubbles in the  $4 \times 4$  matrix given in the ORS.

### Column I

### Column II

- (A) If -1 < x < 1, then f(x) satisfies
- 0 < f(x) < 1
- (B) If 1 < x < 2, then f(x) satisfies
- (q) f(x) < 0
- (C) If 3 < x < 5, then f(x) satisfies
- f(x) > 0
- (D) If x > 5, then f(x) satisfies
- (s) f(x) < 1



65. Let (x, y) be such that

$$\sin^{-1}(ax) + \cos^{-1}(y) + \cos^{-1}(bxy) = \frac{\pi}{2}$$
.

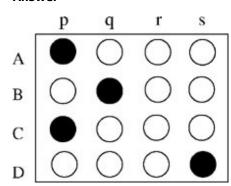
Match the statements in Column I with statements in Column II and indicate your answer by darkening the appropriate bubbles in the  $4 \times 4$  matrix given in the ORS.

### Column I

- (A) If a = 1 and b = 0, then (x, y)
- (B) If a = 1 and b = 1, then (x, y)
- (C) If a = 1 and b = 2, then (x, y)
- (D) If a = 2 and b = 2, then (x, y)

### Column II

- (p) lies on the circle  $x^2 + y^2 = 1$
- (q) lies on  $(x^2-1)(y^2-1)=0$
- (r) lies on y = x
- (s) lies on  $(4x^2-1)(y^2-1)=0$



66. Match the statements in **Column I** with the properties in **Column II** and indicate your answer by darkening the appropriate bubbles in the 4 × 4 matrix given in the ORS.

### Column I

- (A) Two intersecting circles
- (B) Two mutually external circles
- (C) Two circles, one strictly inside the other
- (D) Two branches of a hyperbola

### Column II

- (p) have a common tangent
- (q) have a common normal
- (r) do not have a common tangent
- (s) do not have a common normal

