Physics Paper III

Time Allowed: 2½ Hours]

[Maximum Marks: 150

Note: This Paper contains Seventy Five (75) multiple choice questions, each question carrying Two (2) marks. Attempt *All* questions.

1. The eigenvalues of (2×2) matrix

$$\begin{bmatrix} 2 & 3 \\ 1 & 0 \end{bmatrix}$$

are:

- (A) 0, 2
- (B) 1, 3
- (C) 3, -1
- (D) 1 + i, 1 i
- 2. The asymptotic behaviour of the curve $y = e^{-x^2} \sin x$ is :
 - (A) oscillatory
 - (B) goes to zero
 - (C) not defined
 - (D) goes to unity
- 3. Five boys and three girls are to stand in a line for a photograph, in such a way that no two girls are adjacent. Number of ways in which this can be done is:
 - (A) 720
 - (B) 360
 - (C) 14400
 - (D) 11600

4. The room temperature measured by different instruments gave the following results:

37°C, 70°F, 311°K, 35°C

The average temperature would be:

- (A) 113°K
- (B) 37°C
- (C) 35°C
- (D) 70.5°F
- 5. $\lim_{x\to 0} x^x \text{ is :}$
 - (A) 0
 - (B) 1
 - (C) i
 - (D) ∞
- 6. The particular integral of the inhomogeneous linear differential equation:

$$\frac{dy}{dx} - y = x$$

is given by:

- (A) e^x
- (B) $x e^x$
- (C) -x 1
- (D) x + 1

- 7. Masses m and 3m are attached to the two ends of a spring of spring constant k. What is the period of oscillation ?
 - (A) $4\pi\sqrt{\frac{m}{k}}$
 - (B) $2\pi\sqrt{\frac{k}{m}}$
 - (C) $2\pi\sqrt{\frac{m}{2k}}$
 - (D) $\pi\sqrt{\frac{3m}{k}}$
- 8. The mutual potential energy v of two particles depends on their mutual distance r as follows:

$$v = \frac{a}{r^2} - \frac{b}{r}$$

where a and b are positive constants. For what separation r are the particles in static equilibrium ?

- (A) r = a + b
- (B) r = ab/a + b
- (C) r = 2a/b
- (D) $r^2 = a^2 + b^2$

- 9. The captain of a boat becalmed in the equatorial doldrum moves the anchor vertically upwards with velocity v. Will the boat start moving? Which way?
 - (A) Yes, northwards
 - (B) Yes, eastwards
 - (C) Yes, westwards
 - (D) no.
- 10. An empty cylindrical can and a can tightly packed with wheat flour, both start rolling down on inclined plane at the same instant. Which can reaches the bottom first?
 - (A) Empty can
 - (B) Filled can
 - (C) Both arrive at the same time
 - (D) Sizes of the cans must be known to answer this question

- 11. For attractive inverse square forces the shape of the orbit will be:
 - (A) Elliptic
 - (B) Parabolic
 - (C) Hyperbolic
 - (D) All of the above
- 12. A cork is submerged in a pail of water by a spring attached to the bottom of the pail. The pail is held by a child in an elevator. During the initial acceleration as the elevator travels to the next lower floor, the length of the spring will:
 - (A) increase
 - (B) decrease
 - (C) remain the same
 - (D) any one of the above depending on the spring constant

- - $(A) \overline{0}$
 - (B) $\frac{1}{4\pi \in_0} \frac{q^2}{d^2} \hat{z}$
 - (C) $\frac{1}{16\pi \in_0} \frac{q^2}{d^2} \hat{z}$
 - (D) $\frac{-1}{16\pi \in_0} \frac{q^2}{d^2} \hat{z}$
- 14. The dimensions of a quantity $\in_0 E^2$ are
 - (A) MLT^{-2}
 - (B) ML^2T^{-2}
 - (C) $ML^{-1}T^{-2}$
 - $(\mathrm{D})\ ML^{-2}T^2$

15.	The direction of propagation	17.	An electron enters a uniform
	, , , ,		magnetic field region with its
	of electromagnetic wave is given		velocity perpendicular to the
	by		direction of the field. In the field
			region, the trajectory of the electron
	(A) \overline{E}		is
	(P) =		(A) linear
	(B) <u>B</u>		(B) circular
	(C) $\overline{E} \times \overline{B}$		(C) parabolic
	(D) $\overline{B} \times \overline{E}$		(D) hyperbolic
16.	Electric field at large distance r , from	18.	An infinitely long straight wire
	Electric field at large distance 7, from		is carrying current I. The
	the electric quadrupole is		magnetic field at a distance r from
	proportional to		it is
			(A) $\frac{\mu_0 I}{r}$
	(A) r^{-2}		Γ
	(D) -3		(B) $\frac{\mu_0 I}{\pi r}$
	(B) r^{-3}		
	(C) r^{-4}		(C) $\frac{\mu_0}{2\pi r}$
	(B) r^{-3} (C) r^{-4} (D) r^3		(C) $\frac{\mu_0}{2\pi r}$ (D) $\frac{\mu_0}{\pi r^2}$
	(D) r^3		(D) $\frac{}{\pi r^2}$

- 19. The parity of wave function ψ is associated with which of the following transformation ?
 - (A) Space inversion
 - (B) Space rotation
 - (C) Space translation
 - (D) Space exchange of two particles
- 20. The energy levels of the deuterium atom are given in terms of the principal quantum number n and a positive constant A by the expression:
 - (A) $A(n + \frac{1}{2})$
 - (B) $A\left(-\frac{1}{4} + \frac{1}{n^2}\right)$
 - (C) $-A/n^2$
 - (D) An^2

- 21. A one-dimensional harmonic oscillator is perturbed by a perturbation potential αx^3 . The ground state energy of the oscillator to a first order perturbation is:
 - (A) $\frac{\hbar\omega}{2}$
 - (B) $\frac{\hbar\omega}{2} + \alpha$
 - (C) $\frac{3}{2}\hbar\omega + \alpha$
 - (D) $\frac{\hbar\omega}{2} + \alpha^3$
- 22. The configuration of three electrons 1s2p3p has which of the following as the value of its maximum possible total angular momentum quantum number ?
 - (A) $\frac{7}{2}$
 - (B) $\frac{5}{9}$
 - (C) 3
 - (D) $\frac{3}{9}$

23. A freely moving electron is localized in space to within Δx at x_0 , its wave function can be described by a wave packet :

$$\psi(x, t) = \int_{-\infty}^{\infty} e^{i(kx - wt)} f(k) dk,$$

where f(k) is peaked around a central value k_0 . Which of the following is most nearly the width of the peak in k?

(A)
$$\Delta k = \frac{1}{x_0}$$

(B)
$$\Delta k = \frac{1}{\Delta x}$$

(C)
$$\Delta k = \frac{\Delta x}{x_0^2}$$

(D)
$$\Delta k = (\Delta x) \frac{k_0}{x_0}$$

24. ψ_1 and ψ_2 are two orthonormal wave functions for an electron. Which of the following can describe a state of two electrons ?

(A)
$$\psi_1(x_1) \ \psi_2(x_2)$$

(B)
$$\frac{\left[\psi_1(x_1) + \psi_2(x_2)\right]}{\sqrt{2}}$$

(C)
$$\frac{1}{\sqrt{2}} \{ \psi_1(x_1) \psi_2(x_2) + \psi_1 \}$$

(D)
$$\frac{1}{\sqrt{2}} \Big\{ \Psi_1 \Big(x_1 \Big) \, \Psi_2 \Big(x_2 \Big) - \Psi_1 \\ \Big(x_2 \Big) \Psi_2 \Big(x_1 \Big) \Big\}$$

- 25. Consider distributing 2 identical particles over 3 energy levels.

 Treating the particles as indistinguishable Fermi-Dirac particles, the number of ways of distributing is:
 - (A) 9
 - (B) 6
 - (C) 3
 - (D) 2^3

- 26. The energy density of the photon gas maintained at a temperature "T" is proportional to T^n , where 'n' is:
 - (A) 1
 - (B) 2
 - (C) 3
 - (D) 4
- 27. The average value \overline{v} (non-relativistic) of the velocity of a gas of molecules maintained at a temperature 'T' is given by :
 - (A) $\frac{1}{2}k_{\mathrm{B}}\mathrm{T}$
 - (B) zero
 - (C) $\frac{3}{2}k_{\rm B}T$
 - (D) $\frac{k_{\rm B}T}{m}$

- 28. In the canonical ensembles the system:
 - (A) is not maintained at a fixed temperature
 - (B) can exchange energy with the surroundings
 - (C) can exchange the number of particles with the surroundings
 - (D) is completely isolated from the surroundings
- 29. The entropy of the universe in a reversible process is:
 - (A) is constant
 - (B) is increasing
 - (C) is decreasing
 - (D) becomes infinite

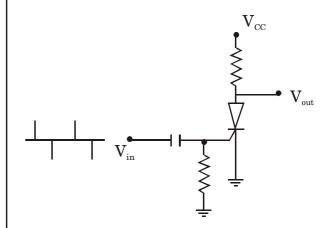
- 30. For a system in thermodynamic equilibrium the following must be necessarily constant throughout the system:
 - (A) Temperature and pressure
 - (B) Temperature and not pressure
 - (C) Pressure and chemical potential
 - (D) Temperature, pressure, chemical potential
- 31. Noise voltage varies in a resistor as:
 - (A) $4kTR_B$
 - (B) $\sqrt{4kTR_{\rm B}}$
 - (C) $\sqrt{2eI_PB}$
 - (D) $\sqrt{2eI_PB.R}$

- 32. G.M. counter is used for the detection of:
 - (A) only alpha particles
 - (B) only beta particles
 - (C) only gamma radiation
 - (D) alpha, beta and gamma radiations
- 33. In a photomultiplier if each stage emits 4 secondaries per primary and there are ten stages the gain is of the order of:
 - (A) 4^{10}
 - (B) 10^4
 - (C) 40
 - (D) $\frac{10!}{4!}$

- 34. A thin and hollow electrode is tapered and open at the end. The electrode is filled with an electrolyte.

 The electrode is put in an electrolyte bath. The tip will have:
 - (A) A very high resistance only
 - (B) A very low resistance only
 - (C) A high resistance with capacitance
 - $\begin{array}{cccc} \text{(D)} & A & low & resistance & with \\ & & capacitance & & & \\ \end{array}$
- 35. Best vacuum that can be obtained with a rotary pump is:
 - (A) 10^{-3} Torr
 - (B) 10 Torr
 - (C) 10^{-6} Torr
 - (D) 10^{-10} Torr

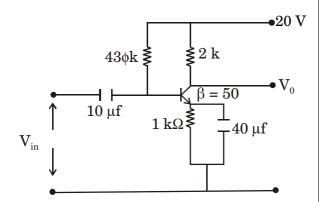
- 36. Precise value of e/\hbar can be obtained by :
 - (A) Millikan's oil drop method
 - (B) Compton effect
 - (C) Michelson interferrometer
 - (D) Superconducting quantum interference device
- 37. What is the output for the following circuit?



- (A) Square wave
- (B) Triangular wave
- (C) Sawtooth
- (D) Sine wave

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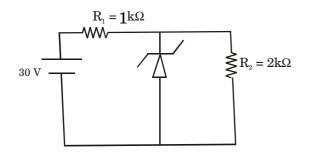
38. Find out the value of ${\rm I}_B$ and ${\rm V}_C$ in the following circuit :



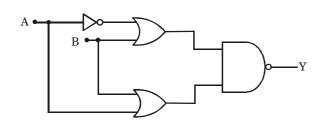
- (A) 43 μA and 11.4 V
- (B) 50 μA and 10 V
- (C) 40 $\,\mu A$ and 16 V
- (D) 45 $\,\mu A$ and 11 V
- 39. In which of the following ADC the speed of conversion is maximum:
 - $(A) \ \ Successive-approximation \ ADC$
 - (B) Counter-type ADC
 - (C) Dual-slope type ADC
 - (D) Parallel-comparator ADC

- 40. Which of the following interrupts is un-maskable interrupt ?
 - (A) RST 5.5
 - (B) RST 7.5
 - (C) TRAP
 - (D) INTR 1
- 41. To send digital data over telephone lines, the modulation technique used is:
 - (A) AM
 - (B) FM
 - (C) FSK
 - (D) Z-modulation

42. In the given circuit the current through resistor $R_2(2\ k\Omega)$ is :

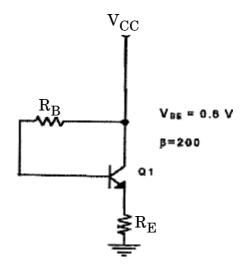


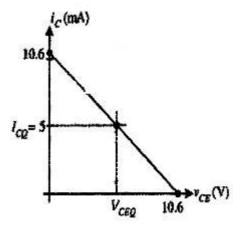
- (A) 2 mA
- (B) 15 mA
- (C) 10 mA
- (D) 6 mA
- 43. In the given digital logic circuit, A and B form the inputs. The output Y is:



- $(A) Y = \overline{A}$
- (B) $Y = A\overline{B}$
- (C) $Y = A \oplus B$
- (D) $Y = \overline{B}$

44. Find the values of V_{CC} , R_{B} and R_{E} in the following circuit :

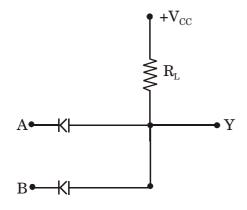




- (B) V_{CC} = 10.6 V, R_B = 200 Ω , R_E = 1 $k\Omega$
- (C) V_{CC} = 10.6 V, R_B = 200 kΩ, R_E = 1 kΩ

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- 45. In op-amp zero crossing detector :
 - (A) output of comparator is zero
 - (B) $-V_{CC} = 0$
 - (C) Reference voltage is zero
 - (D) $+V_{CC} = 0$
- 46. The following circuit acts as:



- (A) AND
- (B) OR
- (C) NOT
- (D) NAND

- 47. Consider a planar ion CO_3^{2-} . How many normal modes of vibration this ion will exhibit ?
 - (A) 6
 - (B) 7
 - (C) 4
 - (D) 12
- 48. For 1s and 2s states of the hydrogen atom, the all points is space at which the corresponding wave functions vanish is:

1s 2s

- (A) ∞ $r = 2a_0$ and ∞
- B) 0 $r = a_0, \infty$
- (C) $r = a_0 \quad \infty$
- (D) ∞ 0

- 49. Give functional form of the variation of the potential energy with the separation r between a neutral atom and an ion :
 - (A) $\frac{-1}{r}$
 - (B) $\frac{1}{r^{12}} \frac{1}{r^6}$
 - (C) $\frac{-1}{r^2}$
 - (D) $\frac{-1}{r^3}$
- 50. A molecule can be infrared active provided a mode of vibration produces:
 - (A) a change in magnetic dipole moment
 - (B) a change in polarizability
 - (C) a change in electric dipole moment
 - (D) a change in quadrupole moment

- 51. A light source of wavelength 4358 Å is used to irradiate an organic liquid giving 3 Raman lines at 4400, 4419 and 4447 Å. The 'Raman Shift' of 4447 Å line will be:
 - (A) 219 cm^{-1}
 - (B) 316.8 cm^{-1}
 - (C) 345.2 cm^{-1}
 - (D) 459.2 cm^{-1}
- 52. Distance between two protons in a hydrogen molecule is :
 - (A) 1 nm
 - (B) 0.1 nm
 - (C) 0.01 nm
 - (D) 0.001 nm
- 53. State the quantum numbers for the ground state of neutral boron (atomic number 5).
 - (A) ${}^{2}P_{1/2}$
 - (B) ${}^{1}S_{0}$
 - (C) $^{2}S_{0}$
 - (D) ${}^{3}D_{1/2}$

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- 54. Give the formula and the numerical value in meters, for the radius of the first Bohr orbit in hydrogen atom:
 - (A) $\frac{\hbar}{mc} = 3.8 \times 10^{-13}$ meters
 - (B) $\frac{\hbar}{mc} = 1.4 \times 10^{-15} \text{ meters}$
 - (C) $\frac{\hbar^2}{me^2} = 5.3 \times 10^{-11}$ meters
 - (D) $\frac{\hbar}{p} = 2 \times 10^{-17}$ meters
- 55. Alpha particles and protons of the same kinetic energy are passed through a gold foil. What is the ratio of their Coulomb scattering intensity?
 - (A) 2
 - (B) 3
 - (C) 3/2
 - (D) 4
- 56. What is the ground state binding energy of an atom consisting of an electron and a positron bound to each other by their Coulomb interaction?
 - (A) 13.6 eV
 - (B) 6.8 eV
 - (C) 8 eV
 - (D) 20.2 eV

- 57. If the first reflection from an f_{cc} crystal has a Bragg angle θ of 21.5°, the second reflection will have an angle θ of :
 - (A) 18.5°
 - (B) 25.0°
 - (C) 31.0°
 - (D) 47.13°
- 58. One of the following is a non-polar molecule:
 - (A) HCl
 - (B) CO_2
 - (C) H₂O
 - (D) HF
- 59. A rare earth ion has a ground state described by a basic level ⁵I₈. The maximum value of the orbital angular momentum L according to Hund rules is:
 - (A) 3
 - (B) 4
 - (C) 5
 - (D) 6

- 60. For Al, critical magnetic field at absolute zero is 105 gauss.

 The stabilization energy density of the superconducting Al wire in erg/cm³ is:
 - (A) 530.00
 - (B) 320.00
 - (C) 439.00
 - (D) 100.00
- 61. In a semiconductor the effective mass of a hole near the top of valence band is:
 - (A) positive
 - (B) negative
 - (C) zero
 - (D) pure imaginary
- 62. The plot of the energy momentum relation for free electrons has the shape of :
 - (A) A rectangular hyperbola
 - (B) An ellipse
 - (C) A circle
 - (D) A parabola

- 63. The magnetisation of inside a type I superconductor at temperatures T below the critical temperature T_c behaves :
 - (A) linearly as a function of the applied field $H_{\rm C}$
 - (B) is a non-zero constant as a function of H, upto the critical field $H_{\rm C}$
 - (C) is identically zero upto the critical field $H_{\rm C}$
 - (D) shows parabolic behaviour upto H_{C} and shows a gap above H_{C}
- 64. In an experiment of photoelectric effect, monochromatic X-rays are incident on the surface of a metal. Some of the resulting photoemitted electrons do not lose energy in escaping the surface and they leave the metal surface with a certain kinetic energy. If φ is the workfunction of the metal, hυ the photon energy and E is the binding energy (depth below the Fermi level from which the electron originates), the kinetic energy of the electron will be:
 - (A) K.E. = $hv + E \phi$
 - (B) K.E. = $hv E + \phi$
 - (C) K.E. = $hv E \phi$
 - (D) K.E. = $hv + E + \phi$

- 65. When a beam of electron is reflected from a solid Aluminum surface, it is found that the electrons lose energy in discrete amounts of 15, 31 and 46 eV. Assuming the ideal electron gas model to be valid, the losses must be due to:
 - (A) plasma oscillations
 - (B) valence electron excitations
 - (C) core electron excitations
 - (D) generation of photoelectrons
- 66. Which of the following is *not* true in connection with the nuclear force ?
 - (A) Nuclear force is short-ranged
 - (B) Nuclear force is charge independent
 - (C) Nuclear force is spin-dependent
 - (D) The proton-proton force is stronger than the neutron-neutron force

- 67. The ratio of sizes (radii) of $^{208}_{82}$ Pb and $^{26}_{12}$ Mg is approximately.
 - (A) 2
 - (B) 4
 - (C) 8
 - (D) 7
- 68. Which of the following is *not* a doubly magic nucleus ?
 - (A) $^{16}_{8}$ O
 - (B) $^{40}_{20}$ Ca
 - (C) $^{208}_{82}$ Pb
 - (D) $^{238}_{92}$ U
- 69. A nucleus with mass number 204 decays by α -emission. The Q-value of the reaction is 5.26 MeV. The kinetic energy of the α -particle is :
 - $(A) \ 5.26 \ MeV$
 - (B) 2.63 MeV
 - (C) $5.26 \times \frac{200}{204} \text{ MeV}$
 - (D) $5.26 \times \frac{4}{204} \text{ MeV}$

- 70. Which of the following is violated by a β -decay phenomenon ?
 - (A) Energy conservation
 - (B) Momentum conservation
 - (C) Angular momentum conservation
 - (D) Parity conservation
- 71. The nuclear fission due to an absorption of a slow neutron is an example of:
 - (A) Compound nuclear reaction
 - (B) Direct reaction
 - (C) Neutron scattering reaction
 - (D) Quantum mechanical tunneling
- 72. Which of the following reactions is allowed one?

(A)
$$P + P \rightarrow k^+ + \Sigma^+$$

(B)
$$P + P \rightarrow k^+ + P + \wedge^0$$

(C)
$$\pi^- + P \rightarrow \Sigma^+ + K^0$$

(D)
$$\pi^{-} + P \rightarrow \pi^{0} + \wedge^{0}$$

- 73. The quark structure of π^+ is :
 - (A) uu
 - (B) $\bar{u}d$
 - (C) *dd*
 - (D)
- 74. The symmetry term in the semiempirical mass formula is due to:
 - (A) non-equality of proton number Z and neutron number N
 - (B) non-spherical shape of nuclei
 - (C) charge-independence of nuclear force
 - (D) non-zero value of quadrupole moment of nuclei
- 75. α-particles of energy 5 MeV passing through an ionisation chamber completely lose their energy. Assuming that 35 eV of energy is required to produce one ion pair, find the height of the output pulse, if the capacity of the system is 10 μμf.
 - (A) $2.85 \times 10^{-3} \text{ volts}$
 - (B) $3.2 \times 10^{-3} \text{ volts}$
 - (C) 2.24×10^{-3} volts
 - (D) $3.4 \times 10^{-3} \text{ volts}$

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ROUGH WORK

Test Booklet No.				
प्रश्नपत्रिका क्र.				

Paper-III PHYSICS

Sign	nature and Name of Invigilator	Seat No.
1. (S	ignature)	(In figures as in Admit Card)
(N	Vame)	Seat No.
2. (Signature)		(In words)
(N	Jame)	OMR Sheet No.
FEB - 32313		(To be filled by the Candidate)
Tim	e Allowed : 2½ Hours]	[Maximum Marks : 150
Nun	nber of Pages in this Booklet : 20	Number of Questions in this Booklet: 75
1. 2. 3.	Instructions for the Candidates Write your Seat No. and OMR Sheet No. in the space provided on the top of this page. This paper consists of 75 objective type questions. Each question will carry two marks. All questions of Paper-III will be compulsory, covering entire syllabus (including all electives, without options). At the commencement of examination, the question booklet will be given to the student. In the first 5 minutes, you are requested to open the booklet and compulsorily examine it as follows: (i) To have access to the Question Booklet, tear off the paper seal on the edge of this cover page. Do not accept a booklet without sticker-seal or open booklet. (ii) Tally the number of pages and number of questions in the booklet with the information printed on the cover page. Faulty booklets due to missing pages/ questions or questions repeated or not in serial order or any other discrepancy should not be accepted and correct booklet should be obtained from the invigilator within the period of 5 minutes. Afterwards, neither the Question Booklet will be replaced nor any extra time will be given. The same may please be noted. (iii) After this verification is over, the OMR Sheet Number should be entered on this Test Booklet. Each question has four alternative responses marked (A), (B), (C) and (D). You have to darken the circle as indicated below on the correct response against each item. Example: where (C) is the correct response.	विद्यार्थ्यांसाठी महत्त्वाच्या सूचना 1. परिक्षार्थींनी आपला आसन क्रमांक या पृष्ठावरील वरच्या कोप-यात लिहावा. तसेच आपणांस दिलेल्या उत्तरपत्रिकेचा क्रमांक त्याखाली लिहावा. 2. सदर प्रश्नपत्रिकेत 75 बहुपर्यायी प्रश्न आहेत. प्रत्येक प्रश्नास दोन गुण आहेत. या प्रश्नपत्रिकेतील सर्व प्रश्न सोडविणे अनिवार्य आहे. सदरचे प्रश्न हे या विषयाच्या संपूर्ण अभ्यासक्रमावर आधारित आहेत. 3. परीक्षा सुरू झाल्यावर विद्यार्थ्याला प्रश्नपत्रिका दिली जाईल. सुरुवातीच्या 5 मिनीटांमध्ये आपण सदर प्रश्नपत्रिका उघडून खालील बाबी आवश्य तपासून पहाव्यात. (i) प्रश्नपत्रिका उघडण्यासाठी प्रश्नपत्रिकेवर लावलेले सील उघडावे. सील नसलेली किंवा सील उघडलेली प्रश्नपत्रिकची एकूण पृष्ठे तसेच प्रश्नपत्रिकेतील एकूण प्रश्नांची संख्या पडताळून पहार्वा. पृष्ठे कमी असलेली/कमी प्रश्न असलेली/प्रश्नांचा चूकीचा कम असलेली किंवा इतर त्रुटी असलेली सदोष प्रश्नपत्रिका सुरुवातीच्या 5 मिनिटातच पर्यवेक्षकाला परत देऊन दुसरी प्रश्नपत्रिका मागवून घ्यावी. त्यानंतर प्रश्नपत्रिका बदलून मिळणार नाही तसेच वेळही वाढवून मिळणार नाही याची कृपया विद्यार्थांनी नोंद घ्यावी. (iii) वरीलप्रमाणे सर्व पडताळून पहिल्यानंतरच प्रश्नपत्रिकेवर ओ.एम.आर. उत्तरपत्रिकेचा नबर लिहावा. 4. प्रत्येक प्रश्नासाठी (A), (B), (C) आणि (D) अशी चार विकल्प उत्तरे दिली आहेत. त्यातील योग्य उत्तराचा रकाना खाली दर्शविल्याप्रमाणे ठळकपणे काळ/निळा करावा. उदा. : जर (C) हे योग्य उत्तर असेल तर.
5.6.7.8.	Your responses to the items are to be indicated in the OMR Sheet given inside the Booklet only. If you mark at any place other than in the circle in the OMR Sheet, it will not be evaluated. Read instructions given inside carefully. Rough Work is to be done at the end of this booklet. If you write your Name, Seat Number, Phone Number or put any mark on any part of the OMR Sheet, except for the space allotted for the relevant entries, which may disclose your identity, or use abusive language or employ any other unfair means, you will render yourself liable to disqualification. You have to return original OMR Sheet to the invigilator at the end of the examination compulsorily and must not carry it with you outside the Examination Hall. You are, however, allowed to carry the Test Booklet and duplicate copy of OMR Sheet on conclusion of examination. Use only Blue/Black Ball point pen. Use of any calculator or log table, etc., is prohibited. There is no negative marking for incorrect answers.	