1.	Asprin is an acetylation product of		
	(1) o-hydroxybenzoic acid	(2)	o-dihydroxybenzene
	(3) m-hydroxybenzoic acid	(4)	p-dihydroxybenzoic
2.	The number of molecules of ATP produced in the	e lipid	metabolism of a molecule of palmitic acid is
	(1) 130 (2) 36	(3)	
2			
3.	In DNA, the complementary bases are		
	(1) uracil and adenine; cytosine and guanine		
	(2) adenine and thymine; guanine and cytosine		
	(3) adenine and thymine; guanine and uracil		*
	(4) adenine and guanine; thymine and cytosine		
4.	Which of the following elements constitutes a ma	ajor in	npurity in pig iron ?
	(1) Silicon	(2)	Oxygen
	(3) Sulphur	(4)	Graphite
_	D 4 1 6 1:1 6:1 6:1 : 6 ::		
5.	Repeated use of which of the following fertilizers		
	(1) Urea	(2)	Potassium Nitrate
	(3) Ammonium Sulphate	(4)	Superphosphate of lime
6.	Phenylmethanol can be prepared by reducing the	e benz	zaldehyde with
	(1) CH <sub>3</sub> Br (2) Zn and HCL	(3)	
7.	In the commercial gasoline, the type of hydrocar	bons v	which are more desirable is
	(1) branched hydrocarbon	(2)	straight-chain hydrocarbon
	(3) linear unsaturated hydrocarbon	(4)	toluene
8.	The compound obtained by heating a mixture	re of	ethylamine and chloroform with ethanolic
	potassium hydroxide (KOH) is		And Annahis and a service service service and a service service and a service service service and a service service service and a service service and a service service and a service service service and a service service service service and a service service service and a service serv
	(1) an ethyl isocyanide	(2)	an alkyl halide
	(3) an amine	(4)	an amide and nitro compound
0	In graphite, electrons are		
9.		(0)	
	(1) localized on every third C-atom	(2)	present in anti-bonding orbital
	(3) localized on each C-atom	(4)	spread out between the structure
10.	Alkyl halide is converted into an alcohol by		
	(1) addition	(2)	substitution
	(3) elimination	(4)	dehydrohalogenation
		0.00	
11.	The electronic configuration of transition elemen		
	(1) $ns^2(n-1) d^{1-10}$	(2)	$ns^2(n-1)d^{10}$
	$(3)$ $ns^1$	(4)	$ns^2np^5$
12.	The given reaction 2 $\operatorname{FeCl}_3 + \operatorname{SnCl}_2 \rightarrow 2 \operatorname{FeCl}_2 +$	SnCl	is an example of
	(1) first order reaction		· · · · · · · · · · · · · · · · · · ·
	(3) third order reaction	(2)	second order reaction
1/12/1/		(4)	None of these
IVIS 1/F	V2K13/05 (2)		

13.	Which of the following is not a Lewis acid?													
	(1) $BF_3$ (2)	FeCl <sub>3</sub>	(3)	$SiF_4$	(4)	$\mathrm{C_2H_4}$								
14.	Aldol condensation will not take	e place in		- ' y	, K.									
	(1) HCHO		(2)	CH <sub>3</sub> CH <sub>2</sub> C	НО									
	(3) CH <sub>3</sub> CHO		(4)	CH <sub>3</sub> COCH	$I_3$	8								
15.	Which of the following compoun	d gives benzoic a	acid on	hvdrolysis	?									
	(1) Chlorophenol		(2)	Chlorotolu										
	(3) Chlorobenzene		(4)	Benzoyl Cl	nloride									
16.	Which of the following has zero	dipole moment?	K III											
	(1) cis 2-butene		(2)	trans 2-but	tene									
	(3) I-butene		(4)	2-methyl,	I-prope	ene								
17.	An aniline on nitration gives													
	$^{ m CH_3}$	NH <sub>2</sub>		NH <sub>2</sub>										
	(1) (2)	$NO_2$	(	3)		(4) B	oth (2) an	d (3)						
	$NO_2$	$\bigcirc$												
	$NO_2$			$NO_2$										
18.	The number of moles of oxygen in one litre of air containing 21% oxygen by volume, in standard													
	conditions, is			2000 mg 1000										
	(1) 0.186 mole		(2)	0.21 mole										
	(3) 0.0093 mole		(4)	2.10 moles										
19.	The concentration units, indepe	ndent of tempera	ature,	would be										
	(1) molality		(2)	molarity										
	(3) normality		(4)	weight vol	ume pe	ercent								
20.	Half-life of radioactive <sup>14</sup> C is 5760 years. In how many years 200 mg of <sup>14</sup> C will be reduced to 25 mg?													
	(1) 5760 years		(2)	11520 year	'S									
	(3) 17280 years		(4)	23040 year	'S									
21.	The coordination number and or	xidation state nu	mber	of Cr in K <sub>3</sub> C	${ m cr(C_2O_4)}$	) <sub>3</sub> are res	spectively							
8	(1) 6 and +3 (2)	4 and +2	(3)	3 and +3	(4)	3 and 0								
22.	The alkene $R - CH = CH_2$ real hydrogen peroxides produces	cts readily with	$B_2H_6$	and the pro	oduct o	n oxidati	on with a	alkaline						
	(1) $R - CH_2 - CHO$		(2)	$R - CH_2 -$	CH <sub>o</sub> =	ОН								
	(3)  R - C = O			=	_	011								
	÷ 1		(1)	R-CH-C										
	$\mathrm{CH}_3$			OH C	θH									
23.	Which of the following ions can	cause coagulatio	n of pr	roteins?										
	$(1) Ag^{\dagger} \qquad (2)$	K++	(3)	Mg <sup>++</sup>	(4)	Ni <sup>++</sup>								
M31/A	V2K13/05	(3)				2								

24.	The explosive nitro-glycerine is		
	(1) a complex compound	(2)	a nitro hydrocarbon
	(3) an ester	(4)	an ether
25.	The manufacture of nylon-66 involves condensat  (1) hexa methylene diamine and adipic acid  (2) ethylene glycol and phthalic acid  (3) urea and formaldehyde  (4) None of these	tion or	co-polymerisation of
26.	Which of the following sets of elements has the s	trong	est tendency to form anions ?
-0.	(1) N, O, F	(2)	
	(3) Cu, Ag, Au		Pb, Bi, Po
12000			20, 20, 20
27.	In the formation of BeF <sub>2</sub> , Beryllium makes use of	of	
	(1) sp <sup>3</sup> orbitals	(2)	2s and 2p orbitals
	(3) dsp <sup>2</sup> hybrid orbital	(4)	sp hybrid orbitals
28.	The Cannizzaro reaction is not given by		
20.	(1) trimethyl acetaldehyde	(2)	acetaldehyde
	(3) formaldehyde	(4)	benzaldehyde
	(o) iormalacity ac	(4)	benzaidenyde
29.	The lowest ionization potential in a periodic tabl (1) representative elements (3) noble gases	(2) (4)	lanthanides alkali metals
30.	Hydration of butyne-1 in presence of $\mathrm{H_2SO_4}$ and		AND THE RESERVE TO THE PROPERTY OF THE PROPERT
	(1) CH <sub>3</sub> – CHO		$CH_3 - COC_2H_5$
	(3) $C_2H_5 - CO - C_2H_5$	(4)	$C_2H_5 - CO - CH_3$
31.	The atomic number of an element is related to		*
	(1) the number of electrons in the nucleus of th		
	(2) units of positive charge on the nucleus of th	e aton	n.
	<ul><li>(3) the number of nucleons.</li><li>(4) its atomic weight.</li></ul>		2
	Annual Maries and Annual Marie		
32.	The IUPAC name of the compound represented	by the	formula $(CH_3)_2$ $CHCH_2 - CH_2 = CH_2$ is
	(1) isopentene	(2)	4-methyl pentene-1
	(3) 2-methyl pentene-4	(4)	4, 4-dimethyl butane-1
33.	C - C bond length is shortest in		
	(1) methane (2) benzene	(3)	acetylene (4) ethylene
34.	Order of relative spidio strongths of others athe	no or	
04.	Order of relative acidic strengths of ethane, ethe		
	(1) ethane > ethyne > ethene	(2)	ethyne > ethane > ethene
	(3) ethene > ethane > ethyne	(4)	ethyne > ethene > ethane
M31/A	/2K13/05 (4)		

35.	The	reaction $C_6H_5N_2Cl \xrightarrow{CU_2Br_2/HBr} C_6H_5$	Br +	N <sub>2</sub> + HCl is called
	(1)	Sandmeyer's reaction		
	(2)	Wurtz Fittig reaction		
	(3)	Perkin's reaction		
	(4)	Ethard reaction		
				HBr KCN H <sub>o</sub> O
36.	The	product 'C' in the following series CH3 - CH	= CI	$12 \xrightarrow{\text{REIV}} A \xrightarrow{\text{REIV}} B \xrightarrow{2} C \text{ is}$
	(1)	$\mathrm{CH_3.CH_2CONH_2}$		
	(2)	$\mathrm{CH_3}$ . $\mathrm{CH_2CH_2COOH}$		
	(3)	(CH <sub>3</sub> ) <sub>2</sub> CHCOOH		
	(4)	$\mathrm{CH_{3}CH_{2}CH_{2}CONH_{2}}$		
37.	Gas	es cannot be liquefied above their		
	(1)	inversion temperature	(2)	critical temperature
	(3)	boiling point	(4)	None of these
38.	Ent	halpy of compound under standard condition	is equ	ual to its
	(1)	enthalpy of formation	(2)	enthalpy of combustion
	(3)	enthalpy of sublimation	(4)	enthalpy of solution
39.	The	most important source of helium is		
	(1)	ozone layer (2) natural gas	(3)	minerals (4) radon
40.	The	best method to establish the purity of an orga	anic s	solid is to determine its
	(1)	mixed melting point	(2)	melting point
	(3)	colour	(4)	None of these
41.	Ace	taldehyde when treated with ethyl alcohol in	the p	resence of dilute HCl gives
17 8	(1)	formaldehyde	(2)	crotonaldehyde
	(3)	pronanal	(4)	acetal
42.	Feh	ling's solution is an alkaline solution of		
	(1)	Cu <sup>+</sup> ions in the presence of tartrate ions		
	(2)	Cu <sup>+</sup> ions in the presence of citrate ions		
	(3)	Cu <sup>++</sup> ions in the presence of citrate ions		
	(4)	Cu <sup>++</sup> ions in the presence of tartrate ions		
	800	entre destruction contributor. • 12 destruction per 1900 (19		

(5)

M31/A/2K13/05

43.	Heisenberg test can be used to distinguish between												
	) primary amines, secondary amines and tertiary amines												
	(2) primary amines and secondary amines												
	(3) primary amines and tertiary amines												
	(4) All these												
44.	Grignard reagent does not react with												
	(1) tertiary amine (2) secondary amine												
	(3) primary amine (4) All these												
45.	The amount of gas dissolving in a given volume of the liquid decreases when												
	(1) temperature and pressure both are decreased												
	(2) temperature and pressure both are increased												
	(3) temperature is increased												
	(4) pressure is decreased												
46.	The osmotic pressure of a solution of benzoic acid in benzene is found to be half the expected variable is due to	lue.											
	(1) association of benzoic acid molecules forming dimers												
	(2) wrong method of measurement of osmotic pressure												
	(3) limitation of the theory of osmotic pressure												
	(4) ionization of benzoic acid												
47.	The protective power of lyophilic colloid is												
	(1) determined by the size of the colloidal particles												
	(2) dependent on quantity of charge passed												
	(3) measured by nature of colloidal sol												
	(4) expressed by gold number												
48.	2-butene on heating with alkaline KMnO <sub>4</sub> gives												
	(1) acetone (2) acetic acid												
	(3) acetaldehyde (4) propanol												
49.	The rate constant of a reaction depends on the												
10.	(1) temperature (2) pressure												
	(3) concentration (4) time												
<b>F</b> 0													
50.	The quality of gasoline sample is determined by its												
	(1) iodine value (2) cetane number												
M31/4	(3) octane number (4) mass density (2K13/05 (6)												

51.	The cause of the potential (1) depletion of positive c (2) concentration of positive c (3) depletion of negative c (4) concentration of positive c	harges n ive charg charges i	ear the junction ges near the junction near the junction	n nction on	1 × × ×		
52.	Light of wavelength 5000 The kinetic energy of the p		[1] [[[[[[] [[] [[] [] [] [] [] [] [] [] [			ctric	work function of 1.9 eV.
	(1) 0·58 eV	(2) 2	2·48 eV	(3)	1·24 eV	(4)	1·16 eV
53.	A ball of mass 0.25 kg att circle. The string will brea which the ball can be move	ak, if the			10 2011 회원이라 참 하다. (15 H.		없이 있다면서 그리스 CONTROL (CONTROL CONTROL C
	(1) 3 m/s	(2) 5	5 m/s	(3)	9·8 m/s	(4)	14 m/s
54.	The time of flight of a projection (3) air resistance	ectile on	an upward incl	ined (2) (4)	plane depends angle of inclin Both (1) and (	ation	
55.	If a ladder is not in equilibly (1) increasing the angle of			verti			n be made in equilibrium
	(3) increasing the length	of the la	dder	(4)	decreasing the	e len	gth of the ladder
56.	In an isothermal change of	an idea	l gas, $\Delta U = 0$ . T	he cl	nange in heat e	nerg	y $\Delta Q$ is equal to
	(1) 0·5 W	(2) 2	2 W	(3)	1.5 W	(4)	None of these
57.	For protecting sensitive eq (1) placed inside an alum (3) wrapped with insulat	inium ca		(2)	placed inside	an ir	
58.	A 5°C rise in temperature doubled, the rise in temper	rature w	ill be approxim	ately		\$ 11 \$2.50	
	(1) 5°C	(2) 1	l0°C	(3)	20°C	(4)	40°C
59.	A hollow insulated conduction field at the centre of the sp	here, if	ts radius is 2 n	etre	s?		
	(1) Zero	(2) 5	бµ Ст <sup>-2</sup>	(3)	$20\mu~\mathrm{Cm}^{-2}$	(4)	32μ Cm <sup>-2</sup>
60.	A transverse wave is repre	esented b	y the equation	: y =	$y_0 \sin \frac{2\pi}{\lambda}$ (vt	- x). ]	For what value of $\lambda$ is the
	particle velocity equal to to						
	(1) $\lambda = 2\pi y_0$	(2) λ	$x = \frac{\pi y_0}{3}$	(3)	$\lambda = \frac{\pi y_0}{2}$	(4)	$\lambda = \pi y_0$
61.	The physical quantity who (1) weight	se dimer			fferent from ot thrust		is electromotive force
62.	If two bulbs, whose resista in them has the ratio of						
	(1) 1:1	(2)	1:2	(3)	2:1	(4)	1:4
M31/A	/2K13/05		(7)				a a
	P <sub>2</sub>						

						# P
						1 1
63.	Which of the following star	A STATE OF THE PARTY OF THE PAR	50° 00'00'00' <b>8</b> 0'00'00 o			
	No. Programme and the second	cell increases with incre				
		proportional to the appl		A 270		
		reases with intensity of				8,0
	(4) The stopping potentia	al increases with increa	se or	incident light		
64.	The centre of mass of syste		depe	nds on		
	(1) masses of the particle					
	(2) forces on the particles					
	(3) position of the particl					
	(4) relative distances bet					
65.	A ball whose kinetic energy highest point of its flight w		ang	le of 45° with	the h	norizontal. Its K.E. at the
	(1) E	$(2)  \frac{\mathrm{E}}{2}$	(3)	$\frac{\mathrm{E}}{\sqrt{2}}$	(4)	0
		2	(0)	$\sqrt{2}$	(-)	v
66.	Due to earth's magnetic fie	ald the changed accomic		nautialas		
00.	Due to earth's magnetic fie (1) require greater kineti	ic energy to reach the e	- 5	7		
		nergy to reach the equa	-			
	(3) can never reach the p		.001 01	an pole		
	(4) can never reach the e			and p		
CT.			m.			1
67.	A sample of gas expands to	from volume $v_1$ to $v_2$ .	The	amount of wor	K do	ne by gas is the greatest,
	when the expansion is (1) isothermal	(2) isobaric	(3)	adiabatic	(4)	ogual in all agges
					(4)	equal in all cases
68.	Which of the following dim	nensions will be the san	ne as	that of time?		
	(1) LC	$(2)$ $\frac{R}{r}$	(3)	$\frac{L}{R}$	(4)	$\frac{\mathbf{C}}{\mathbf{c}}$
	pr .	L		K		L
69.	An equation is given $(P)$	$+\frac{a}{V^2}$ = $b\frac{\theta}{V}$ , where	e P	= Pressure, V	= V	olume and $\theta$ = Absolute
	temperature. If a and b are	e constants, the dimens	sions	of 'a' will be		
	(1) $M L^5 T^{-2}$	(2) $M^{-1} L^5 T^2$	(3)	${ m M} \; { m L}^{-5} \; { m T}^{-1}$	(4)	$\mathrm{M}\ \mathrm{L}^{5}\ \mathrm{T}^{1}$
70.	The dimensions of impulse	are equal to that of				
	(1) force	dro equal to that of	(2)	angular mom	entiii	m
	(3) pressure		(4)	linear momen		
71		1.1	223400			1 1100 - 77 1
71.	An electron of mass m an vacuum. Its final velocity		ited fi	rom rest thoug	gh a	potential difference V in
	$(1)  \frac{\text{eV}}{2\text{m}}$	(2) $\frac{\text{eV}}{\text{m}}$	(3)	$\frac{2eV}{}$	(4)	eV
	2m	m		V m		V m
72.	The refractive index of wat	ter is 1.33 N. What will	be th	ne speed of ligh	t in v	water?
	(1) $3 \times 10^8 \text{m/s}$	(2) $2.25 \times 10^8 \text{ m/s}$		1204		
<b>#</b> C				1 10 111/3	(1)	100 / 10 111/3
73.	Which of the following is a	dimensional constant				19
	(1) Refractive index		(2)	Poisons ratio		
1/124/4/	(3) Relative density	(0)	(4)	Gravitational	cons	tant
IVIS I/A/	2K13/05	(8)				

74.	A 4 μF capacitor is charged produced in the resistan		0 V. If its pla	tes are j	oined throug	gh a resi	stance of 2 k $\Omega$ ,	then heat
	(1) 0·16 J	(2)	0·32 J	(3)	0.64 J	(4)	12·8 J	
75.	In a simple harmonic m the total energy is kineti		hen the disp	lacement	t is one-half	f the am	plitude, what f	raction of
	(1) Zero	(2)	1/4	(3)	1/2	(4)	3/4	
76.	An oscillator is nothing (1) voltage gain (3) positive feedback	out an ar	mplifier with	(2) (4)	no feedbac			
77.	For measuring temperat (1) barometer (2) pyrometer (3) gas thermometer	ures in t	he range of 2				ploy	8
	(4) platinum-rhodium	thermom	eter					
78.	If we consider electrons (1) energy (3) momentum			vaveleng (2) (4)	th, then the velocity angular m			
79.	In good conductors of electronic (1) ionic	etricity,	the type of bo covalent		nat exist is metallic	(4)	Van der Waals	
80.	One watt hour equals							
	(1) one horse power			(2)	$3.6 \times 10^3$ jo	oules		
	(3) $3.6 \times 10^3$ calories			(4)	4.2 joules			
81.	If a particle moves in a c (1) changes in direction (2) remains always con (3) changes both in ma (4) neither changes in a	n stant gnitude :	and direction		in equal tim	es, its v	elocity vector	
82.	If the distance between t	wo mass	es is doubled	, the gra	vitational a	ttraction	between them	
	(1) becomes double			(2)	is reduced	to 1/4		
	(3) is reduced to half			(4)	gets four-t	imes		
83.	Ball pen functions on the	e princip	le of					
	(1) Stoke's law			(2)	Pascal's la	w		
	(3) Capillary action			(4)	Gravitatio	nal force		
84.	Which of the following is	a vector	quantity?					
	(1) Power	. , , , , , ,	quinterly .	(2)	Electrostat	tic poten	tial	
	(3) Work			(4)	Momentun	372		
QE.		no io co	al to	7-2				
85.	One atmospheric pressu	_	90.0		6			
1404	$(1) 10^5  \text{dynes/cm}^2$	(2)	One pascal	(3)	10 <sup>6</sup> N/m	(4)	One bar	
W31/A	V2K13/05		( 9	)			2 10	

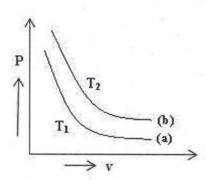
- 86. On heating a liquid of coefficient of cubical expansion  $\gamma$  in a container having a coefficient of linear expansion  $\frac{\gamma}{3}$ , the level of liquid in the container will
  - (1) fall

(2) rise

(3) it is difficult to say

- (4) remain almost stationary
- 87. The slopes of isothermal and adiabatic curves are related as
  - (1) Isothermal curve slope =  $y \times Adiabatic curve slope$
  - (2) Isothermal curve slope = Adiabatic curve slope.
  - (3) Adiabatic curve slope =  $\gamma \times$  Isothermal curve slope
  - (4) Adiabatic curve slope =  $\frac{1}{2}$  × Isothermal curve slope
- 88. Coefficient of thermal conductivity depends upon
  - (1) temperature difference between the two sides
  - (2) thickness of the metal plate
  - (3) area of the plate
  - (4) All these
- 89. Two bodies have their thermal capacities in the ratio 1:4. It is found that rates of temperature are the same. Then their rates of loss of heat must be in the ratio
  - (1) 1:1

- (2) 1:4
- (3) 4:1
- (4) 1:3
- 90. For a certain mass of gas the isothermal relations between p and v are shown by graphs (a) and (b) at the kelvin temperatures  $T_1$  and  $T_2$  respectively. Then



(1)  $T_1 < T_2$ 

(2)  $T_1 > T_2$ 

(3)  $T_1 = T_2$ 

- (4) None of these
- 91. In Carnot's heat engine, the total work done during the adiabatic processes is
  - (1) zero
  - (2) infinite
  - (3) dependent on sink temperature
  - (4) more than that in case of isothermal
- 92. The main feature of the photoelectric effect is that
  - (1) electrons are emitted by the heating of metals with ultraviolet rays
  - (2) photographic images are formed under action of light
  - (3) electrons are ejected when light falls on some metals
  - (4) metals become conducting under the action of light

93.	A liquid will present a c	oncave surface if the ad	lhesion with the con	tainer is									
	(1) less than cohesion among its molecules												
	(2) equal to cohesion a	mong its molecules											
	(3) greater than cohes	on among its molecules	3										
	(4) None of these												
94.	Glycerin is more viscous	s than water and this ca	an be ascertained fro	m the fact that									
	(1) glycerin more read	ily sticks to the body th	an water.										
	(2) glycerin is heavier	than water.											
	(3) a body falls throug	(3) a body falls through glycerin is slower than in water.											
	(4) a body loses more v	veight in glycerin than	water.										
95.	Two charged spheres of they have the same	radii 10 cm and 15 cm	are connected by a t	chin wire. No current will flow if									
	(1) field on their surfa	ces.											
	(2) charge on each.												
	(3) potential.												
	(4) force.												
96.	In a charged capacitor,												
	<ul><li>(1) the positive charges</li><li>(2) both the positive and negative charges</li></ul>												
	(3) around the edges of												
	(4) the field between t												
	(1) the held between t	are praces											
97.	Kirchhoff's first law is related to												
	(1) Faraday's second law of electrolysis												
	(2) Law of conservation of momentum												
	(3) Law of conservatio	(3) Law of conservation of energy											
	(4) Lenz's law												
98.	In a sinusoidal wave, the time required for a particular point to move from maximum displacement to zero displacement is 0·17 sec. The frequency of the wave is												
	(1) 1·47 Hz	(2) 0·36 Hz	(3) 0·73 Hz	(4) 2·94 Hz									
99.		gths in the ratio 2:1.	If the temperature	diameters of the rods are in the difference between the ends is									
	(1) 1:1	(2) 1:8	(3) 2:1	(4) 8:1									
100.	Intensity of magnetizat	ion is equal to											
	(1) 4 times the permea	ability of the magnetic r	naterial										
	(2) the magnetic mom	ent per unit volume		x . x									
	(3) the pole strength p	R we illustrate											
	(4) the magnetic induc	etion											
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101.	From the top of a light house 60 metres high w of a boat is 150°. The distance of the boat from	rith its base at the sea-level, the angle of depression the foot of the light house is
	(1) $\left(\frac{\sqrt{3-1}}{\sqrt{3+1}}\right)$ 60 m	$(2)  \left(\frac{\sqrt{3-1}}{\sqrt{3+1}}\right) m$
	$(3)  \left(\frac{\sqrt{3+1}}{\sqrt{3-1}}\right) m$	(4) $\left(\frac{\sqrt{3+1}}{\sqrt{3-1}}\right)$ 60 m

- 102. If  $\tan^{-1} x + \tan^{-1} y + \tan^{-1} z = \pi$ , then the value of x + y + z is

  (1) 2xyz (2) 3xyz(3) 4xyz (4) None of these
- 103. If  $\begin{vmatrix} 6i 3i & 1 \\ 4 & 3i & -1 \\ 20 & 3 & i \end{vmatrix} = x + iy$ , then

  (1) x = 3, y = 1(2) x = 1, y = 3(3) x = 0, y = 3(4) x = 0, y = 0
- 104. The function  $f(x) = (x^2 1) |x^2 3x + 2| + \cos (|x|)$  is not differentiable at

  (1) -1 (2) 0 (3) 1 (4) 2
- 105. The shortest distance of (0, c) from the parabola  $y = x^2$ ,  $0 \le c \le 5$  is (1)  $\sqrt{4c+1}$  (2)  $\sqrt{\frac{4c+1}{2}}$  (3)  $\sqrt{\frac{4c-1}{2}}$  (4) None of these
- 106. The area of the figure bounded by the curves y = |x-1| and y = 3 |x| is

  (1) 2 (2) 3 (3) 4 (4) 1
- 107. The number of vectors of unit length perpendicular to vectors  $\overrightarrow{a}(1, 1, 0)$  and  $\overrightarrow{b}(0, 1, 1)$  is

  (1) 1

  (2) 2

  (3) 3

  (4) infinite
- 108.  $f(x) = \begin{cases} x^2 \sin \frac{1}{x}; & x \neq 0 \\ 0; & x = 0 \end{cases}$ , then
  - (1) f and f' are continuous at x = 0
  - (2) f is derivable at x = 0
  - (3) f is derivable at x = 0 and f' is not continuous at x = 0
  - (4) f' is derivable at x = 0
- 109. A circle of radius unity is inscribed in an isosceles triangle. The least perimeter of the triangle is
- (1)  $2\sqrt{3}$  (2)  $3\sqrt{3}$  (3)  $6\sqrt{3}$  (4) 9 M31/A/2K13/05 (12)

110.	The ball	number of wa s is	ys of selec	ting	10 ba	lls out of	an unl	imited	numbe	r of w	vhite, red,	blue and g	reen
	(1)	270		(2)	84		(3)	286		(4)	86		
111.		wo unequal par ween them at th					ocus aı	nd axis	in oppo	site o	directions,	then the a	ngle
	(1)	$\frac{\pi}{4}$	- Si	(2)	$\frac{\pi}{6}$		(3)	$\frac{\pi}{3}$		(4)	$\frac{\pi}{2}$		
112.	If f(	$\mathbf{x}) = \left(\frac{\mathbf{x}}{2+\mathbf{x}}\right)^{2\mathbf{x}}$	, then lim	<sup>l</sup> x→∞	f (x)	equals			0.				
	(1)	$e^{-6}$		(2)	2		(3)	$e^{-4}$	*	(4)	$\frac{1}{9}$		
113.	The	curve y-e <sup>xy</sup>	+x=0 ha	ıs a v	ertica	l tangent	at the	point					
		(1, 1)				point		(0, 1)		(4)	(1, 0)		
114.		CD is a square that from B is					of the t	op of a	pole sta	andin	g at D from	m A or C is	s 30°
	(1)	$\sqrt{6}$		(2)	$\frac{1}{\sqrt{6}}$		(3)	$\frac{\sqrt{3}}{2}$		(4)	$\frac{\sqrt{2}}{3}$		
115.	Ifα	$+\beta=3$ and $\alpha$	$^{3} + 6^{3} = 2'$	7 th	en a f	3 are the	roots o	f					
		$3x^2 + 9x + 7 =$			, ,	71			27x + 20	0 = 0			
		$2x^2 - 6x + 15$							of these				
116.		Q(1, 0), Q(-1, 0) tion $SQ^2 + SI$			are t	hree give	en poir	nts, the	en the l	ocus	of point S	satisfying	g the
	(1)	a straight line	parallel	to x-a	axis								
	(2)	a circle through	gh origin										
	(3)	a circle with o	entre at t	he or	igin								
	(4)	a straight line	e parallel	to y-a	axis								
117.	If d	$\overrightarrow{b} = \overrightarrow{a} \times (\overrightarrow{b} \times \overrightarrow{b})$	$\overrightarrow{c}$ ) + $\overrightarrow{b}$ >	$\langle (\stackrel{\rightarrow}{c})$	$\times \stackrel{\rightarrow}{a}$	$\rightarrow$ + $\stackrel{\rightarrow}{c}$ ×(	$\frac{\rightarrow}{a} \times \frac{1}{b}$	), the	n				
	(1)	→ d is a unit ve	ector				(2)	$\overrightarrow{d} =$	$\overrightarrow{a} + \overrightarrow{b}$	+ c			
	(3)	$\overrightarrow{d} = \overrightarrow{0}$					(4)	$\stackrel{\rightarrow}{a}$ , $\stackrel{\rightarrow}{b}$	$\overrightarrow{c}$ , $\overrightarrow{c}$ ,	→ d ar	e coplanar		
118.	If th	ne sum of the d	istances o	f a po	oint fr	om two p	erpend	icular l	lines in	a pla	ne is 1, the	en its locus	s is
	(1)	square					(2)	circle					
	(3)	straight line					(4)	two i	ntersect	ing li	ines		
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119.	Value of $\int_{1/e}^{\tan x} \frac{t}{1+t^2} dt + \int_{1/e}^{\cot x} \frac{dt}{t(1+t^2)}$ is
	(1) $\frac{1}{2}$ (2) 1
120.	The vector $\overrightarrow{a} = x \hat{i} - 2 \hat{j} + 5 \hat{k}$ and $\overrightarrow{b} = \hat{i} + y$
	(1) $x = 1, y = -2, z = -5$

120. The vector 
$$\vec{a} = x \hat{i} - 2 \hat{j} + 5 \hat{k}$$
 and  $\vec{b} = \hat{i} + y \hat{j} - z \hat{k}$  are collinear if

(2) 
$$x = \frac{1}{2}$$
,  $y = -4$ ,  $z = -10$ 

(3) 
$$x = -\frac{1}{2}$$
,  $y = 4$ ,  $z = 10$ 

(4) 
$$x = -1$$
,  $y = 2$ ,  $z = 5$ 

121. The image of the interval 
$$[-1, 3]$$
 under the mapping  $f(x) = 4x^3 - 12x$  is

$$(1)$$
  $[-2, 0]$ 

$$(2)$$
  $[-8,72]$ 

$$(3) [-8, 0]$$

(4) None of these

122. If 
$$\int f(x) dx = f(x)$$
, then  $\int [f(x)]^2 dx$  is equal to

$$(1) \quad \frac{1}{2} \left[ f(x) \right]^2$$

(2) 
$$[f(x)]^3$$

(3) 
$$\frac{[f(x)]^3}{3}$$
 (4)  $[f(x)]^2$ 

(4) 
$$[f(x)]^2$$

123. Let 
$$f: R \to R$$
 be a differentiable function and  $f(1) = 4$ . Then the value of  $\lim_{x \to 1} \int_4^{f(x)} \frac{2t}{x-1} dt$  is

124. If 
$$\cos 5A = a \cos A + b \cos^3 A + c \cos^5 A + d$$
, then

(1) 
$$a = 20$$

(2) 
$$b = -20$$

(3) 
$$c = 18$$

(4) 
$$b = 15$$

$$(1) \quad 2^8 - 1$$

(2) 
$$3^8 - 1$$

(3) 
$$4^8 - 1$$

126. If 
$$\lim x_{x\to 0} \phi(x) = a^3$$
,  $a \ne 0$ , then  $\lim x_{x\to 0} \phi\left(\frac{x}{a}\right)$  is

(1) 
$$a^2$$

(2) 
$$\frac{1}{a^3}$$
 (3)  $\frac{1}{a^2}$ 

(3) 
$$\frac{1}{a^2}$$

(4) 
$$a^3$$

127. Let 
$$f: \mathbb{R} \to \mathbb{R}$$
 be any function. Define  $g: \mathbb{R} \to \mathbb{R}$  by  $g(x) = |f(x)|$  for all x. Then g is

(1) onto if f is one-one

(2) one-one if f is one-one

(3) continuous if f is continuous

(4) differentiable if f is differentiable

(1) 
$$\frac{\cos A}{a} = \frac{\cos B}{b} = \frac{\cos C}{c}$$

(2) 
$$\frac{\cos A}{a} + \frac{\cos B}{b} + \frac{\cos C}{c} = \frac{a^2 + b^2 + c^2}{2abc}$$

$$(3) \quad \frac{\sin A}{a} + \frac{\sin B}{b} + \frac{\sin C}{c} = \frac{3}{R}$$

(4) 
$$\sin 2A/a^2 = \frac{\sin 2B}{b^2} + \frac{\sin 2C}{c^2}$$

129.	Area of the region bounded k						
	(1) $\frac{2}{3}$	(2) $\frac{4}{3}$	13	(3)	$\frac{8}{3}$	(4)	None of these
130.	The line which is parallel to	x-axis a	and crosses the	curv	$e y = \sqrt{x}$ at an	n ang	le of 45° is
	$(1)  \mathbf{x} = \frac{1}{4}$	(2) y	$=\frac{1}{4}$	(3)	$y = \frac{1}{2}$	(4)	y = 1
131.	In a certain primary school, and 50 of age 15 years. The						
	(1) 13.50	(2) 13	3	(3)	13.45	(4)	14
132.	A bag contains coins of denote the coins of the three denomare 480 what is total amount	ninatior t of mor	ns are in the pr ney in rupees?	ropor	tion of 2 : 3 : 4	1. If t	he total number of coins
	(1) 180	(2) 1:	20	(3)	190	(4)	170
133.	A boy goes to school at a spe 5 h in all, what is the distan	ce betw	een the village	and	the school?		
	(1) 5 km	(2) 1			8 km	Carriero.	6 km
134.	In a stream flowing at 2 km point in 55 minutes. What is	the spe	eed of the moto	rboa	t in still water	?	
	(1) 22 km/h	(2) 1	5 km/h	(3)	20 km/h	(4)	None of these
135.	8 children and 12 men comp by a man to finish the work.	In how	many days wil	ll 12	men finish the	same	e work ?
	(1) 8	(2) 9		(3)	12	(4)	15
136.	The rate of interest for the fi for the period beyond 5 year how much money did he dep	s 10% p	(1985년) (1984년) - 1984년 (1984년) (1984년				565의 : : () [1] [1] [1] [1] [1] [1] [2] [2] [2] [2] [2] [2] [2] [3] [3] [3] [4] [4] [4] [4] [4] [4] [4] [4] [4
	(1) ₹ 3800	(2) ₹	1500	(3)	₹ 1356	(4)	₹ 1920
137.	A car travels a distance of 1 What is the distance travelled	ed at a s	speed of 100 km	n/h ?			
	(1) 140 km	(2) 1:			230 km		620 km
138.	The ratio between the spee Shyam to cover a certain dis	tance, t	then what is th	e act	ual time taken	by S	hyam ?
	(1) 3 hours	(2) 5	hours	(3)	4 hours	(4)	7 hours
139.	A plot of land is in the s $50\sqrt{2}$ m. The cost of fencing (1) less than $\neq 300$	100	100000				
	(2) more than ₹ 300, but le						
	(3) more than ₹ 500, but le	ess thar	n ₹ 600				
	(4) more than ₹ 600						
140.	Perimeter of a square an $12\sqrt{2}$ cm, then what is the a					he d	iagonal of the square is
	(1) $64\sqrt{3} \text{ sq cm}$			(2)	$54\sqrt{3}$ sq cm		
	(3) $65\sqrt{3} \text{ sq cm}$		10	(4)	$41\sqrt{3}$ sq cm		
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141.		oom 15 m long requires adth of the room?	7500	tiles, each 15 c	m by	12 cm, to cove	r the	e entire floor.	What is the	
	(1)	10 m	(2)	14 m	(3)	6 m	(4)	9 m		
142.		nual income of A and B i h of them saves ₹ 500 at							o of 4:3. If	1000 M
	(1)	₹ 2500 and ₹ 2000			(2)	₹ 1000 and ₹	F 150	00		
	(3)	₹ 1500 and ₹ 2000			(4)	None of these	9	-		
143.		mixture contains water he mixture?	and	alcohol in the ra	tio 2	: 3, what is the	e per	centage quant	ity of water	E
	(1)	25%	(2)	40%	(3)	32%	(4)	50%		
144.		began a business with ₹ profits at the end of the					th ₹	5000. When d	id 'B' join if	
	(1)	After 5 months			(2)	After 3 month	ns			
	(3)	After 7 months			(4)	None of these	)			
145.		erson covers a distance ning at 12 km per hour.						km per hour	and rest by	Œ
	(1)	28 km, 72 km			(2)	32 km, 82 km	ı			
	(3)	24 km, 68 km			(4)	26 km, 70 km	l			
146.		a mixture of 60 litres, the 2, then the amount of wa						of milk and wa	ter is to be	
	(1)	20	(2)	30	(3)	40	(4)	None of these	Э	
147.		in can do a piece of work e to finish the work work			work	s twice as fast	as N	itin, how long	would they	£
	(1)	6 days	(2)	9 days	(3)	8 days	(4)	5 days		
148.		trains of length 110 me n a speed of 35 km/h and								
	(1)	144 sec	(2)	177 sec	(3)	155 sec	(4)	123 sec		
149.		rain passes a railway br			seco	nds. If the tr	ain is	s running at	a speed of	
	(1)	160 m	(2)	150 m	(3)	180 m	(4)	170 m		
150.		iece of wire 132 cm long ular hexagon and a circle						teral triangle,	a square, a	12
	(1)	Equilateral triangle			(2)	Square		0		
	(3)	Circle			(4)	Regular hexa	gon			
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	tions (C		: Ch	oose the word u	hich	best expresses	the n	neaning of the underlined
151.	He wa	s drawn into the <u>vort</u>	ex of	politics at a very	y early	y age.		
	(1) w	hirlpool	(2)	field	(3)	arena	(4)	hell
152.	He ha	s the <u>propensity</u> for ge	etting	into debt.		e e		
	(1) cl	haracteristic	(2)	quality	(3)	tendency	(4)	aptitude
153.		eed of the hour is to					scienc	ce and halt the rampant
	(1) ri	ise	(2)	introduction	(3)	revival	(4)	significance
	tions (6 sentence		: Ch	oose the word w	hich i	s opposite in n	ıeanii	ng of the underlined word
154.	She is	beautiful as well as i	rivol	ous.				
	(1) ir	ndecent	(2)	serious	(3)	insane	(4)	rude
155.	It is ol	oligatory for a commo	n citi:	zen to follow the	rules	i.		
	(1) o	ptional	(2)	superfluous	(3)	necessary	(4)	advisable
156.	You m	ust <u>quote</u> examples to	sup	port your staten	nent.			
	(1) re	eveal	(2)	restrain	(3)	contradict	(4)	adduce
	<b>tions (C</b> e given u		: In 6	each of these qu	estion	s, choose the o	ption	which can be substituted
157.	Stealin	ng from the writings o	of oth	ers				
	(1) C	opying			(2)	Reframing		
	(3) R	eproducing			(4)	Plagiarism		
158.	A pers	on with full discretion	nary j	powers to act on	behal	lf of a country		31
	(1) A	mbassador			(2)	Emissary		*
	(3) P	lenipotentiary			(4)	Envoy		
159.	Design	n made by putting tog	ether	colored pieces of	of glas	s or stones		
	(1) O	leograph	(2)	Mosaic	(3)	Tracery	(4)	Relief
160.	A tem		f and	l often surprisi	ng ch	ange from w	hat i	s normal or accepted as
	(1) A	bhorrence	(2)	Abet	(3)	Abeyance		(4) Aberration
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	tions (Questions 161–164): Choose the option which best expresses the meaning of the underlined phrase in the sentence.						
161.	I just paid him a left-handed compliment.						
	(1) an honest (2) a well deserved						
	(3) an insincere (4) a flattering						
162.	The train was late and we had to kick our heels.						
	(1) run fast (2) wait eagerly						
	(3) waste time (4) play some game						
163.	A movement for the world unity is in the offing.						
	(1) at the end (2) about to start (3) on decline (4) in the air						
164.	It is evident from the minister's statement that heads will roll.  (1) government will change  (2) transfers will take place  (3) dismissals will occur  (4) heads of department will have to repent						
	tions (Questions 165–167): A word has been written in four different ways out of which only one is ly spelt. Choose the correctly spelt word.						
165.	(1) colaboration (2) collaboration (3) collaboration (4) collaboration						
166.	(1) exagerate (2) exagarate (3) exaggerate (4) exeggerate						
167.	(1) acquiescence (2) aquicence (3) acquisence (4) acquissence						
Direc	tions (Questions 168 – 171): A sentence has been broken into four parts. Choose the part that has or.						
168.	<ol> <li>When dinner was done,</li> <li>my master went out to his labourers, and,</li> <li>as I could discover by his voice and gesture,</li> <li>gave his family strict charges to take care of me.</li> </ol>						
169.	<ol> <li>Most of them, and especially those</li> <li>who deal in the astronomical part,</li> <li>have great faith in judicial astrology,</li> <li>although he is ashamed to own it publicly.</li> </ol>						
170.	<ol> <li>From comparing notes afterwards it was but a hour and a quarter,</li> <li>yet it appeared to me that</li> <li>the night must have almost gone,</li> <li>and the dawn be breaking above us.</li> </ol>						
171.	<ol> <li>The supper party given by Mr. Julius Hershey</li> <li>to a few friends at the evening of the 30th</li> <li>will long be remembered</li> <li>in catering circles.</li> </ol>						

(18)

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Directions (Questions 172 - 180): Study the passages below and answer the questions that follow each passage.

## Passage I

Nature is like business. Business sense dictates that we guard our capital and live from the interest. Nature's capital is the enormous diversity of living things. Without it, we cannot feed ourselves, cure ourselves of illness or provide industry with the raw materials of wealth creation. Professor Edward Wilson, of Harvard University says, "The folly our descendants are least likely to forgive us is the ongoing loss of genetic and species diversity. This will take millions of years to correct." Only 150 plant species have ever been widely cultivated. Yet over 75,000 edible plants are known in the wild. In a hungry world, with a population growing by 90 million each year, so much wasted potential is tragic. Medicines from the wild are worth around 40 billion dollars a year. Over 5000 species are known to yield chemical with cancer fighting potential. Scientists currently estimate that the total number of species in the world is between 10-30 million with only around 1.4 million identified. The web of life is torn when mankind exploits natural resources in short-sighted ways. The trade in tropical hardwoods can destroy whole forests to extract just a few commercially attractive specimens. Bad agricultural practice triggers 24 billion tonnes of top soil erosion a year losing the equivalent of 9 million tonnes of grain output. Cutting this kind of unsuitable exploitation and instituting "sustainable utilisation" will help turn the environmental crisis around.

- 172. Why does the author compare 'nature' to business?
  - (1) Because of the capital depletion in nature and business.
  - (2) Because of the similarity with which one should use both.
  - (3) Because of the same interest level yield.
  - (4) Because of the diversity of the various capital inputs.
- 173. "The folly our descendants are least likely to forgive us." What is the business equivalent of the folly the author is referring to?
  - (1) Reducing the profit margin.
  - (2) Not pumping some money out of profits into the business.
  - (3) Eroding the capital lease of the business.
  - (4) Putting interest on capital back into the business.
- 174. Which of the following statements is false in context of the given passage?
  - (1) The diversity of plant life is essential for human existence.
  - (2) Scientists know the usefulness of most plant species.
  - (3) Chemicals for cancer treatment are available from plants.
  - (4) There are around ten times the plant species undiscovered as compared to the discovered ones.

- 175. Which of the following correctly reflects the opinion of the author to take care of hunger in the world?
  - (1) Increase the number of edible plants being cultivated.
  - (2) Increase cultivation of the 150 species presently under cultivation.
  - (3) Increase the cultivation of medical plants.
  - (4) Increase the potential of the uncultivated edible plants.
- 176. Which of the following is mentioned as the immediate cause for the destruction of plant species?
  - (1) Soil Erosion

(2) Destruction of habitat

(3) Cultivation

(4) Agricultural practices

## Passage II

Motivations for ruralism in under developed countries are understandably different from those in developed countries. There, it is a sheer physical necessity for the very act of man's survival. In the Third World countries, which are predominantly rural, the only lever that can lift human life above its present subhuman level, is rural development. Rural life in such countries has been stagnating for countries on end. Nothing worthwhile has been done to ameliorate the conditions of the rural population which is only slightly different from that of their quadruped counterparts. Ignorance, ill health and poverty have become synonyms of rural life in the undeveloped and underdeveloped countries. But the worst tragedy is that the concerned human populations have taken this state of affairs for granted, as something unalterable, something for which there is no remedy. Every ray of hope has gone out of their lives. In such countries, Rural Development is the inevitable condition of any material or non-material advancement. As such, enlightened sections of all such countries have been taking ever growing interest in the question of Rural Development. This was also part of the legacy of their freedom struggle. In countries like India, it is wellknown that attempts at Rural Development were an inseparable part of the Independence movement. Leaders like Gandhiji realised quite well that Real India lived in her stagnating villages. Cities, which were mostly the products of Western colonialism, were just artificial showpieces. Even there, there were two worlds. The posh areas, where the affluent few, mostly, the products and custodians of imperial interest lived, were little islands engulfed by the vast ocean of dirt, represented by the vast majority of people. Cities were by no means unknown to India, but in ancient India, they were integral parts, organically related to the rest of the country and society. But, modern cities are exotic centres of commercial and industrial exploitation. Cities in ancient India were the flowers of cultural and artistic excellence of the nation, modern cities are just parasites, preying on and debilitating the country. Hence, Gandhiji started the 'Go to Village Movement' which alone, according to him, could bring freedom to India and sustain it. Rural Development had the pride of place in his strategy for the nation's freedom. Thus, it had its origin in the freedom struggle.

177.	Peo	People are taking growing interest in Rural Development because									
	(1)	(1) nothing worthwhile can be done in the near future.									
	(2)	<ol><li>they have now become optimistic about it.</li></ol>									
	(3)	3) they have realised the indispensability of it.									
	(4)	they have been suffering	from seve	re health	probl	lems.					
178.	Wh	ich of the following is the	worst trag	edy accor	ding t	to the author?					
	(1)	Lack of realisation of the	importan	ce of rura	l deve	elopment.					
	(2)	Exploitation of the rural	people by	the city-c	lwelle	ers.					
	(3)	The subhuman condition	of the peo	ple.							
	(4)	The pessimism of the run	ral people	about the	ir ow	n conditions.					
179.	Wh	ich of the following statem	ents is <i>no</i>	t true in	the co	ontext of the passage?					
	(1										
	(2)	Rural development is a p	re-requisi	te of any	other	advancement and progress.					
	(3)	The rural folk in the Timproved.	hird Worl	d countri	es fee	el that their subhuman condition cannot be					
	(4)		t can rais	se the st	anda	rd of living of people in the Third World					
180.	Which of the following best describes the two divergent worlds of the modern cities?										
100.	(1) Commercial and industrial exploitation.										
	(2)										
	(3)										
	(4)	Posh area an	any poor p	eopie.							
	(4)	rosn area an									
181.	Wh	o won the Laureus World	Snortemar	Award i	n the	veer 2013 2					
101.	(1)	Andy Murry	opor tsmar	1 maru 1	(2)	Usain Bolt					
	25 C.S. III	Michael Phelps			50.60	Lionel Messi					
	(3)	wichael Flielps			(4)	Lioner Wessi					
182.	Wh 201		after defea	iting Aus	tralia	4-0 in cricket test series played in India in					
	(1)	Bradman-Pataudi			(2)	Warne-Kumble					
	(3)	Waugh-Dravid			(4)	Border-Gavaskar					
183.		-		93		a Bharat Petroleum Corporation Ltd (BPCL) Cr to set up a LNG terminal at					
	(1)	Tezpur			(2)	Mangalore					
	(3)	Rajahmundry			(4)	Jamnagar					
M31/A	/2K1	3/05		(21)							

184.	Who heads the companies	Grasim Ind. Ltd, Hind	dalco I	nd. Ltd and Idea Cellular Ltd.?
	(1) Anand G. Mahindra		(2)	Pawan Munjal
	(3) Laxmi Mittal		(4)	Kumar Mangalam Birla
185.	Who among the following Officer (CEO) of National	할머니는 어머니는 이 이 이 그는 그리에서 어머니 맛이 아니다 하는 때 그 그 그 그리다 이어 없는데 얼마를 먹었다		anaging Director (MD) and Chief Executive 13?
	(1) Ravi Narain		(2)	Chitra Ramkrishna
	(3) Naina Lal Kidwai		(4)	Chandra Kochhar
186.	Name the asteroid (a 150	ft rock) passed within 2	27,600	km of earth in 2013.
	(1) 2013 AS15		(2)	2012 DA14
	(3) AS14 2010		(4)	AT2011 13
187.	As per Knights Frank's W in the world?	ealth Report 2013, wh	ich co	untry has the highest number of billionaires
	(1) United Kingdom	(2) China	(3)	USA (4) India
188.	Who became the first n International'?	on-American world e	ditor	in more than 80 years history of 'TIME
	(1) Bobby Ghosh		(2)	Jug Suraiya
	(3) Ravindra Kumar		(4)	None of these
189.	Who among the following messaging application?	co-founded the recen	tly la	unched popular 'MessageMe', a multimedia
	(1) Sanjay Mehrotra	(2) Arjun Sethi	(3)	Doug Purdy (4) Sabeer Bhatia
190.	Which pharmaceutical Mi Court of India to restrain			ia with the rejection of its plea by Supreme sfacturing generic drugs?
	(1) GlaxoSmithKline	(2) Pfizer	(3)	Ranbaxy (4) Novartis AG
191.	Which mobile manufactur for an alleged tax violation		with a	a ₹ 2000 Cr notice by Government of India
	(1) Samsung	(2) Nokia	(3)	BlackBarry (4) Apple
192.	Pakistan People's Party s complete its full			e first democratically elected government to n's existence.
	(1) five years	(2) three year	(3)	four years (4) six years
M31/A	/2K13/05	(22)		

193.	Chi	China exports about 55% of its arms/weapons to which country?								
	(1)	North Korea		(2)	Iran					
	(3)	Pakistan		(4)	Myanmar					
194.	UK	froze the 'Bank Mellat' mon	ey of around € 15	2 mil	lion in 2009 belongs to which country?					
	(1)	Iran		(2)	Iraq					
	(3)	Egypt		(4)	Libya					
195.		ently, India won the legal arding the construction of w			Arbitration in The Hague against Pakistan pjects in Kashmir?					
	(1)	Raavi		(2)	Kishenganga					
	(3)	Chenab		(4)	Indus Suru					
196.					ki' to avert collapse of the banking system in on and International Monitory Fund ?					
	(1)	Greece		(2)	Turkey					
	(3)	Cyprus		(4)	Serbia					
197.		o won the 'Saraswati Samm rary work ?	an' in 2013, insti	tuted	by KK Birla Foundation for an outstanding					
	(1)	Sugathakumari		(2)	Hariram Meena					
	(3)	Sonika Agarwal		(4)	None of these					
198.		ich film won the 'Best Featu nformation and Broadcastin		ation	al film award for 2012 organised by Ministry					
	(1)	Chittagong		(2)	Vicky Donar					
	(3)	Dhag		(4)	Paan Singh Tomar					
199.		and Government of India o	decided to launch		plastic notes in five cities of India on					
	(1)	₹ 5 (2	) ₹ 10	(3)	₹ 20 (4) ₹ 50					
200.	Mr Ind		who died recently	, was	the president of which group of companies of					
	(1)	Sahara Group		(2)	JK Group					
6	(3)	Mahindra Group	N 2 1	(4)	Hero Group					
M31/A	/2K13	3/05	(23)							