(A) (B) (C) (T) (T) (A) (C) (C) (C) (C) (C) (C) (C) (C) (C) (C	8 bits 12 bits 16 bits 16 bits 17 bits 18 bits 19 16 bits 19 10 1 M 10 10 M 10 10 M 11 bits 19 10 M 10 10 M 10 Dipole-dipole forces only exist between non-polar molecules 19 Dipole-dipole forces occur between polar molecules 10 Dipole-dipole forces are caused by the interaction of partial charges on both molecules
(C) (D) 6. Mc (A) (B) (C) (T) (T) (A) (D) (D) (D) (D) (D) (D) (D) (D) (D) (D	12 bits 16 bits blarity of 4% solution of sodium hydroxide solution is: 0.1 M 10 0.5 M 20 0.01 M 21 1.0 M 22 Dipole-dipole forces only exist between non-polar molecules 23 Dipole-dipole forces occur between polar molecules 24 Dipole-dipole forces are caused by the interaction of partial charges on both molecules
(D) 6. Mc (A) (B) (C) (T) (T) (() (()	plarity of 4% solution of sodium hydroxide solution is: 0.1 M 0.5 M 0.01 M 1.0 M The difference between dipole-dipole forces and hydrogen bonds are that: A) Dipole-dipole forces only exist between non-polar molecules B) Dipole-dipole forces occur between polar molecules C) Dipole-dipole forces are caused by the interaction of partial charges on both molecules
6. Mc (A) (B) (C) (T) (T) (A) (C) (C) (C) (C) (C) (C) (C) (C) (C) (C	colarity of 4% solution of sodium hydroxide solution is: 0.1 M 0.5 M 0.01 M 1.0 M The difference between dipole-dipole forces and hydrogen bonds are that: A) Dipole-dipole forces only exist between non-polar molecules B) Dipole-dipole forces occur between polar molecules C) Dipole-dipole forces are caused by the interaction of partial charges on both molecules
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(B) (C) (T) (T) (()	 0.5 M 0.01 M 1.0 M he difference between dipole-dipole forces and hydrogen bonds are that: A) Dipole-dipole forces only exist between non-polar molecules B) Dipole-dipole forces occur between polar molecules C) Dipole-dipole forces are caused by the interaction of partial charges on both molecules
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. (a	 A) Dipole-dipole forces only exist between non-polar molecules B) Dipole-dipole forces occur between polar molecules C) Dipole-dipole forces are caused by the interaction of partial charges on both molecules
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((C) Dipole-dipole forces are caused by the interaction of partial charges on both molecules
્રી	both molecules
	DOCH INOTOCHIO
	(D) None of the above are able to distinguish between dipole-dipole forces
	and hydrogen bonds Which of the following bonds would show the strongest absorption in the Infra
	Which of the following bonds would show and state Red ?
	(D) Sulfur-hydrogen Chem. 2

- Which of the following compounds is the strongest Brönsted base?
 (A) H₂PO⁴⁻
 (B) HSO⁴⁻
 - (C) NO⁸⁻
 - (D) CH₃COO-
 - 10. A homozygous, Rh-positive man (RR) marries an Rh-negative (rr) woman. Their first child is normal, but their second child has hemolytic disease (Rh disease). The first child did not have hemolytic disease because:
 - (A) The child was heterozygous (Rr)
 - (B) The child lacked Rh antigens
 - (C) Anti-Rh antibodies were induced only after the birth of the first child
 - (D) Anti-Rh antibodies present in the mother were destroyed by the immune system of the first child
 - 11. Mendel's law of segregation, as applied to the behavior of chromosomes in meiosis, means that:
 - (A) Pairing of homologs will convert one allele into the other, leading to separation of the types
 - (B) Alleles of a gene separate from each other when homologs separate in meiosis I, or in meiosis II if there is a single crossover between the gene and the centromere
 - (C) Genes on the same chromosome will show 50% recombination
 - (D) Alleles of a gene will be linked and passed on together through meiosis
 - 12. With respect to human height, the production of short individuals by two average-sized parents is best explained by :
 - (A) Mutation
 - (B) Sex linkage
 - (C) Polygenic inheritance
 - (D) Discontinuous variation

13.	A bal	anced polymorphism may be maintained by all the following, except:
	(A)	Natural selection
	(B)	Directional selection
	(C)	Heterozygote advantage
	(D)	Frequency dependent selection
14.	Mem ATP	bers of which of the following groups cannot generate their own
	(A)	Lichens
	(B)	Bacteria
	(C)	Viruses
	(D)	Protozoa
15.	In v	vascular plants DNA is contained in which of the following?
	I.	Nucleus
	II.	Chloroplast
	III.	Mitochondrion
	(A)	I only
	(B)	I and II only
	(C)	I and III only
	(D)	I, II and III
16.	A i	retroviral genome possesses complete information for the synthesis of the lowing components, except:
	(A)) Viral matrix
	(B)) Viral capsid
	(C) Viral envelope
	(D	Receptor binding machinery
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. 17.	How	do virus-infected cells help other cells resist viruses?
	(A)	By producing antimicrobial proteins called complement
	(B)	By producing proteins called interferon
	(C)	By producing proteins called viricide
	(D)	By producing histamine
18.	Antib	iotic penicillin acts by :
	(A)	Acting on plasma membrane of prokaryotic cell
	(B)	Inhibiting the synthesis of NAM and NAG units
	(C)	Inhibiting the cross linking of peptidoglycan strands
	(D)	All of the above
19.		er-soluble globular protein is most likely to have the highest proportio ich of the following amino acid residues buried in its core?
	(A)	Serine
	(B)	Glycine
	(C)	Glutamate
	(D)	Isoleucine
20.	Which	of the following would yield more energy when catabolized tate?
	(A)	Glucose
	(B)	Glucose 1-phosphate
	(C)	Fructose
	(D)	Phospho-enol pyruvate
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- 21. Which of the following does not contribute to tertiary structure?
 - (A) The 'hydrophobic effect', driving non-polar residues to the interior
 - (B) The ability of water to solubilize uncharged, polar side groups
 - (C) The ability of water to solubilize charged side groups
 - (D) The presence at the extreme ends of the protein chain of an ionizable carboxylic acid (C-terminus) and an ionizable amino group (N-terminus)
- 22. Which of the following types of information cannot be determined from the traditional northern blotting technique?
 - (A) The size of an m-RNA species
 - (B) Relative abundance of the m-RNA species
 - (C) The half life of an m-RNA species
 - (D) None of the above
- 23. A protein in an SDS PAGE gel moves slower than the expected molecular weight. If the protein is not post-translationally modified then the behaviour is most likely due to:
 - (A) Denaturation
 - (B) Excessive charge
 - (C) Fatty acylation
 - (D) Multimerization
- 24. Beta Adrenergic receptors are located in :
 - (A) Heart muscle
 - (B) Parasympathetic nervous system
 - (C) Postganglionic neurones of the autonomic nervous system
 - (D) Autonomic ganglia

or.	A	eximately, how much blood flows directly through the atria into	the
25.	E CONTRACTOR STORY	cles even before the atria contract?	0110
	(A)	40%—50%	
	(B)	20%—30%	
	(C)	70%—80%	
	(D)	The atria must contract for blood to flow	
26.	The	exchange of gases between the lungs and lung capillaries is calle	ed :
	(A)	Internal respiration	
	(B)	External respiration	
	(C)	Ventilation	
	(D)	Breathing	
27.	When	n the osmolality of the blood increases:	
	(A)	ADH secretion is decreased in response	
	(B)	Blood volume tends to increase in response	
	(C)	Both occur	
	(D)	Neither occur	
28.		nucleotide sequence at the 3' end of a t -RNA molecule specific to c would be :	robox
	(A)	CUC	
	(B)	CTC	
	(C)	GAG	
	(D)	ACC	
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	29.	Whic	th of the following is not a post-translational modification?
		(A)	Adenylation
		(B)	Glycosylation
		(C)	Phosphorylation
26		(D)	Palmitoylation
	3 0.	Whic	h of the following is not a cis element?
		(A)	Promoter
		(B)	Operator
	86	(C)	Repressor
		(D)	Enhancer
	31.		t product of the immune system attaches to bacteria, making them easier eaten by white blood cells?
		(A)	Hemoglobin
		(B)	Antibody
		(C)	Antigen
		(D)	None of the above
	32.	Plasi	nid vectors for cloning:
		(A)	can generally accommodate larger inserts than phage vectors can
		(B)	grow within bacteria and are present in bacterial colonies on an again
		(C)	include centromeres to allow propagation in yeast
		(D)	burst bacteria and form plaques on a 'lawn' of bacteria
	Ch.	Chem.	8

33.	Whic	h of the following is required for the cell cycle pro	ogression?
	(A)	Cdk and cyclin	
	(B)	Cdk alone	
	(C)	Cyclin alone	
	(D)	None of the above	
34.	If the	e first number of an enzyme in classification is 4,	then it belongs to
	(A)	Ligases	
	(B)	Oxidoreductases	
	(C)	Lyases	
	(D)	Transferases	
35.		Line-Weaver Burk plot which of the following show increased inhibitor concentration?	ws increase in slope
	(A)	Competitive inhibition	
	(B)	Uncompetitive inhibition	
	(C)	Non-competitive inhibition	
	(D)	Both (A) and (C)	
36.	α-ke	etoglutarate + enzyme-NH $_2 \leftrightarrow$ Enzyme + glutam	ate is an example
	(A)	Transamination reaction	
	(B)	Oxidative deamination reaction	
	(C)	Both (A) and (B)	
	(D)	None of the above	
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- 37. Rho factor is required for :
 - (A) Transcription initiation
 - (B) Replication initiation
 - (C) Transcription termination
 - (D) Replication termination
 - 38. DNA solutions "A" absorbs 40% higher at all wave lengths than solution "B", it indicates :
 - (A) DNA in solution A is stable
 - (B) DNA in solution B is denatured
 - (C) DNA in solution A is denatured
 - (D) DNA in both solutions are denatured
 - 39. One explanation for the partial suppression of glucose-dependent insulin release seen in type II diabetes mellitus is that :
 - (A) Pancreatic cells lose their muscarinic receptors
 - (B) Insulin is not processed normally, remaining in the proinsulin form
 - (C) Type II diabetes is characterized by peripheral tissue resistance to insulin only with pancreatic insulin release being normal
 - (D) The GLUT-2 glucose transporter may be under expressed in pancreatic beta cells.

	40.	Which	of the following is not a feature of cancerous cell?	
		(A)	Aneuploidy	
		(B)	Change in cytoskeleton	
		(C)	Decrease in motility	
		(D)	None of the above	
	41.	Which	h of the following statements is true about nucleic acids?	
		(A)	DNA and RNA are isomers because they have the same elemental	
			composition	
		(B)	Uracil and thymine are pyrimidines with each containing two hexago-	138
			nal rings	
		(C)	The sugar phosphate backbone is held together with hydrogen bonds	3
		(D)	None of the above	
	42.	Whi	ch of the following is not true about SRP (signal recognition particle)	?
		(A)	It contains 7s RNA	
		(B)	It determines the destination of proteins	
		(C)	It causes a temporary halt on translation	
		(D)	None of the above	
	Cli.	Chem	n. 11	0

43. Electrons entering the mitochondria via the glycerol phosphate shuttle enter the electron transport chain at the level of :

4. .

- (A) Coenzyme Q
- (B) NADH dehydrogenase at the beginning of Complex I
- (C) Cytochrome b at the beginning of Complex III
- (D) Cytochrome c
- 44. Recoverin acts to 'reset' the visual cycle after a light burst by :
 - (A) Promoting conversion of GTP into cGMP via guanylyl cyclase
 - (B) Closing a calcium channel in the cell membrane
 - (C) Converting all-trans retinal to 11-cis retinal
 - (D) Phosphorylating metarhodopsin
- 45. Which of the following statements about the plasmalemma (cell surface membrane) is true?
 - (A) It allows free and unlimited movement of essential molecules into and out of the cytoplasm
 - (B) Glycolipids and glycoproteins are biological markers which act as antibodies to destory foreign antigens
 - (C) It sometimes contains cholesterol which is thought to affect the fluidity of membrane
 - (D) All of the above

46.	Whi	ch of the following statements about photosynthesis is correct?
	(A)	The first stable product of the light-independent reaction is glycerate 3-phosphate
	(B)	Photolysis take place in the light-dependent stage
	(C)	Water supplies electrons for non-cyclic photophosphorylation
	(D)	All of the above
47.	Whi	ch enzyme is responsible for the production of uric acid?
	(A)	Xanthine oxidase
	(B)	Nucleoside triphosphate pyrophosphohydrolase
	(C)	Hypoxanthine-guanine phosphoribosyltransferase
	(D)	PRPP synthetase
48.	Whi	ch of the following is not a cardiac marker?
	(A)	CPK
	(B)	LDH
	(C)	Troponin T
	(D)	None of the above
49.	The exce	following are all associated with the transport of cabron dioxide by blood pt :
	(A)	Carbaminohaemoglobin
	(B)	Carboxyhaemoglobin
	(C)	Carbonic anhydrase
	(D)	Chloride shift

50.		in response to an intramuscular injection can lead to the elevation ich of the following in the blood?
	(A)	Phosphocreatine kinase
	(B)	Myosin light chain kinase
	(C)	Alkaline phosphatase
	(D)	None of the above
51.	Whic	h of the following statements is true about BMR (Basal Metabolic
	Rate)	?
	(A)	Male and female have equal BMR
	(B)	Children have higher BMR
	(C)	BMR is higher in malnutrition
	(D)	All of the above
52.	The	intake of which foodstuff results in greatest SDA (Specific Dynamic
	Actio	on) ?
	(A)	Carbohydrates
	(B)	Fats
	(C)	Proteins
	(D)	Vitamins
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53.	Smoo	th endoplasmic reticulum is not involved in:	
	(A)	Sequestering of Ca ²⁺	
	(B)	Detoxification of various organic compounds	
	(C)	Release of glucose from glucose-6-phosphate in liver	
	(D)	None of the above	
54.	Shine	e Delgarno sequence is :	
	(A)	Present on r-RNA and rich in purine nucleotides	
	(B)	Present on m-RNA and rich in pyrimidine nucleotides	
	(C)	Present on t-RNA and rich in purine nucleotides	
	(D)	Present on m-RNA and rich in purine nucleotides	
55.	Whic	th of the following is an autoimmune disorder?	
	(A)	Rheumatoid arthritis	
	(B)	Gout	
	(C)	Jaundice	
	(D)	All of the above	
56.		e plotted between formation of double-stranded DNA against timestion and DNA denaturation is called:	ne o
	(A)	Tm curve	
	(B)	Cot curve	
	(C)	Hyperchromic curve	
	(D)	None of the above	
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57.	Which	of the following is not a genetic disorder?
	(A)	Gaucher disease
	(B)	Nieman-Pick disease
	(C)	Burkit lymphoma
	(D)	Goiter
58.		oserine despite being an amino acid is <i>not</i> preferred substrate for protein ation because :
	(A)	It would form serine-homoserine adducts
	(B)	It would lead to cleavage of a peptide bond
	(C)	It is highly hydrophobic
	(D)	It is highly susceptible to proteolytic cleavage
59.		tion between carbohydrates and phenyl hydrazine leads to the formation azone, this is a:
	(A)	Nucleophilic addition
	(B)	Nucleophilic substitution
	(C)	Electrophilic addition
	(D)	None of the above
60.		ompound containing ceramide and phosphocholine attached to terminal OH is called:
	(A)	Cerebroside
	(B)	Ganglioside
	(C)	Cholesterol
	(D)	Sphingomyelin
Cli.	Chem	. 16