M.Sc. DEGREE I SEMESTER EXAMINATION IN ENVIRONMENTAL TECHNOLOGY, DECEMBER 2006

ENV/ENB 2102 CHEMISTRY OF THE ENVIRONMENT

Time: 3 Hrs. Maximum marks: 50

PART - A (Answer ANY FIVE questions)

 $(5 \times 2 = 10)$

- I. 1. Define oxidizing smog.
 - 2. What are the organic and inorganic components of soil?
 - 3. What is Henry's law on the solubility of any gas?
 - 4. What are the sources and sinks of oxygen?
 - 5. What are the radioactive elements?
 - 6. Mention 3 different methods of degradation of organic compounds

PART - B (Answer ANY FIVE questions)

 $(5 \times 3 = 15)$

- II. Write the chemical equations for the depletion of ozone in the stratosphere.
- III. Write a method of mineralisation of a pesticide in soil.
- IV. Explain with chemical equations why water exposed to air shows a pH less than 7.0.
- V. What is role of speciation in the mobility of metals in water?
- VI. Give the mathematical equation for the decay of a radioactive substance.
- VII. How synthetic polymers are degraded in the environment?

$\frac{PART - C}{(Answer \underline{ANY FIVE}}$ questions)

 $(5 \times 5 = 25)$

- VIII. Describe the different layers of atmosphere.
- IX. Explain the ion-exchange and adsorption reactions in the soil.
- X. Describe the carbonate equilibrium in water.
- XI. Describe the sulphur cycle.
- XII. Describe the radioactive tracer technique in environmental applications.
- XIII. Explain the transformation of refractory organic compounds with 5 examples.