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SCIENCE (Theory) - Paper I (Physics and Chemistry)

Time Allowed : $2\frac{1}{2}$ Hours]

[Maximum Marks: 100

Instructions to the Candidates:

c)

d)

- Use of logarithm table is permitted.
- Answer all the questions in Section A. · ii)
- Answer any five questions in Section B. iii)
- Answer any five questions in Section C choosing at least one from each Part.
- Use diagrams, expressions and equations, wherever necessary. v)

(PHYSICS)

(Marks: 50)

SECTION - A

			Answe	er an the	quest	10115.	
I.	Cho	ose tl	ne correct answers :			and a remotivation to the	10 × 1 = 10
	1.	The	resultant of two forces	6N and 8	N act	ting at a point in the sa	ame direction
		is					
		a)	2N		b)	14N	
		c)	48N		d)	on.	
	2.	The	energy possessed by a	body by	virtue	e of its position is	
		a)	potential energy	th.	b)	kinetic energy	
		c)	electrical energy	in andres	d)	heat energy.	
	3.	A fl	oating ship has stabilit			or man to spring similari o	
		a)	metacentre is below t	he centre	of gr	avity of the ship	
		b)	metacentre is below t	he centre	of bu	ioyancy	

metacentre is above the centre of gravity of the ship

metracentre and centre of gravity coincide at the same point.

· II.-

4	. Di	During change of state, the temperature of a substance					
	a)	increases	b)	decreases			
	c)	remains constant	d)	increases and decreases.			
5.	In	a freezing mixture, the ratio of	salt a	nd ice is			
	a)	3:1	b)	2:3			
	c)	3:5	d)	1:3.			
6.	Ast	tigmatism can be corrected by t	using				
	a)	spherical lens	b)	concave lens			
	c)	cylindrical lens	d)	convex lens.			
7.	The	e frequency of a stretched string	g can l	be determined by			
	a)	bolometer	b)	sonometer			
	c),	lactometer	d)	galvanometer.			
8.	The call	e vertical plane passing through led	n the a	axis of a freely suspended magnet is			
	a)	magnetic equator	b)	magnetic meridian			
	c)	magnetic induction	d)	magnetic flux.			
9.	Mut	tual inductance is measured in					
	a)	hertz	b)	henry			
	c)	amperes	d)	ohms.			
10.	X-ra	ys were discovered by					
	a)	Regnault	b)	Roentgen			
	c)	J.J. Thompson					
Con	nplete	the following using appropriate	e word	/ words / expressions : $5 \times 1 = 5$			
11.		boiling point of sea water is		idaia ann quie, patreon a			
12.	Artifi	icial teeth appear	. unde	er ultraviolet light.			
13.							
14.		oractical unit of electrical energ					
15.		ogen nuclei fuse to form					

SECTION - B

Answer any five of the following in one or two sentences each:

 $5 \times 2 = 10$

- 16. What is meant by centre of buoyancy?
- 17. Calculate the work done in lifting a mass of 10 kg through 8 m.
- 18. Define fundamental frequency.
- 19. What are the magnetic elements of earth's magnetism?

 Give reasons for the following:
- 20. Nichrome is embedded in mica in electric iron.
- 21. Sun dose not get cooled.Give any two practical applications of the following :
- 22. Latent heat of vaporisation.
- 23. Total internal reflection.

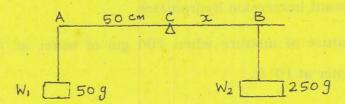
SECTION - C

Answer any five of the following, choosing at least one question from each Part:

 $5 \times 5 = 25$

PART - I

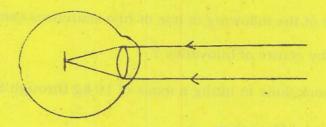
24. Study the diagram and answer the following questions:



- a) Name the principle involved.
- b) State the principle.
- c) Calculate x.

2

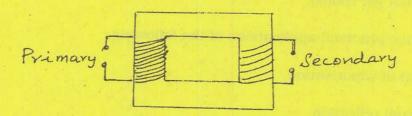
25. Study the diagram and answer the following questions:



- a) Name the defect.

 b) State the causes for this defect.

 c) Draw how can this defect be rectified?
- 26. Study the diagram and answer the questions given below:



a) What is the name of this device?

b) State the principle on which it works.

c) What is called turns ratio?

d) Write one use of this device.

PART - II

- 27. Describe an experiment to determine the relative density of a liquid using the test tube float as constant immersion hydrometer. 5
- 28. Find the temperature of mixture when 100 gm of water at 70° C is added to 200 gm of cold water at 10° C.
- 29. State the laws of transverse vibrations of a stretched string and derive an expression for the frequency of vibration of a stretched string.
- 30. Describe dip circle and explain how it is used to find dip at a place.
- 31. Write the properties of X-rays.

(CHEMISTRY)

(Marks: 50)

SECTION - A

Answer all the questions.

	Cho	ose t	$10 \times 1 = 10$				
	1.	The	e law of definite proportions was exp	laine	d by		
		a)	Dalton Manual III	b)	Lavoisier		
*		c)	Charles	d)	Proust.		
	2.	The	e ideal gas equation is				
		a)	PV = nRT	b)	PT = nRT		
		c)	nR/PV = T	d)	PV ∝ nT.		
	3.	If a	n atom has 11 electrons, the numb	er of e	electrons present in K shell is		
		a)	O. Somming being	b)	1 consister on to to dW 00		
		c)	2	d)	3.		
	4.	The	e compound which is soluble in orga	anic s	olvents is		
		a)	NaCl	b)	CH ₄		
		c)	MgO	d)	CaF ₂ .		
	5.	Wh	ich of the following is a weak electro	olyte '	? The second as books and the second		
		a)	NaOH	b)	HCl		
		c)	CuSO ₄	d)	NH ₄ OH.		
	6.	The effect of emission of α and β particles on mass and atomic number is expressed in terms of					
		a)	Dalton's law	b)	Group displacement law		
		c)	Graham's law	d)	Gay Lussac's law.		

		The oxidising agent present in the matchstick is				
		a)	Potassium hydroxide	b) Potassium chloride	
		c)	Potassium chlorate	d	Potassium sulphate.	
	8.	The	e process of removal of gang	ue from po	wdered ore is known as	
		a)	smelting	b)	refining	
		c)	poling	d)	one dressing.	
	9.	The	e general formula for alcohol	is		
		a)	$C_nH_{2n+1}OH$	b)	$C_nH_{2n-1}OH$	
		c)	$C_nH_{2n+2}OH$	d)	$C_nH_{2n}OH$.	
	10.	Whi	ich of the following is a natur	ral polymer	?	
		a)	Cellulose	b)	PVC	
		c)	Polythene	d)	Benzene.	
1.	Con	Complete the following, using appropriate word/words/expressions: $5 \times 1 = 5$				
	11.	The	Avogadro number is			
	12.		powder mixed	with pota	assium chlorate is used in flash	
		light	photography.			
	13.	Sulp	huric acid is manufactured l	by	process.	
	14.	Amm	noniacal silver nitrate is calle	d as	CASE CONTRIBUTE OF	
	15.		is a phosphatic i	insecticide.		

SECTION - B

Answer any five questions in one or two sentences each:

 $5 \times 2 = 10$

- 16. Define Absolute zero.
- 17. What is an alloy?

Give reasons:

- 18. Aluminium is used to make overhead electric cables.
- 19. Acetylene shows acidic properties. Complete and balance the following equations:
- 20. $P_2O_5 + C \rightarrow +$
- 21. $FeSO_4 + NaOH \rightarrow \dots + \dots$ Give two practical applications of the following:
- 22. Sodium bicarbonate.
- 23. Ethylene.

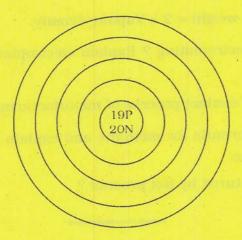
SECTION - C

Answer any five of the following, choosing at least one question from each Part:

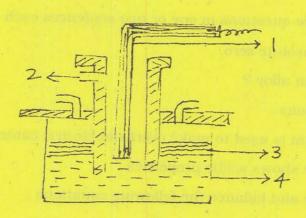
 $5 \times 5 = 25$

PART - I

24. Study the following diagram and answer:



- What is the atomic number of the element? a)
- Write the number of electrons present in this atom. b)
- Find its mass number. c)
- Complete the diagram. d)



	a) Nume the metal extracted.	1
	b) Label the parts marked.	2
	c) Why is the metal extracted in an atmosphere of coal gas?	1
	d) Name the ore taken.	1
26.	You are provided with a conical flask, dropping funnel, delivery tube, b shelf, gas jar, trough and two holed rubber corks:	ee hive
	a) How will you set up the apparatus for preparing acetylene?	3
	b) Write the balanced chemical equation.	2
	PART - II	
27.	Prove that molecular weight = 2 × Vapour density.	5
28.	What is meant by electroplating? Explain electroplating of an aluminium with copper.	spoon 1 + 4
29.	Explain the theory of contact process of manufacturing sulphuric acid.	5
30.		
31.	How is soap manufactured by hot process?	1 + 4
	and a soup method by Hot process?	5