

Note - There are five questions in this paper.

All questions are compulsory.

1. Draw neat and labelled diagrams wherever necessary.

2. Figures to the right indicate full marks of a question.

Q1A Answer the following to the point (Any two) 6

(1) Describe, with diagram the working of a Steam Engine.

(2) Describe, with diagram, the experiment of destructive distillation of a Coal.

(3) Describe the factors responsible for the occurrence of deficiency ~~dis~~ disorders.

B. Answer the following in brief: - [Any five] [10]

(1) Describe the use of wind energy.

(2) Distinguish between petrol and diesel.

(3) State the advantages of a biogas plant.

(4) State the four ~~sem~~ causes of energy crisis.

(5) Give four characteristic symptoms of ~~paper~~ myasthenia.

(6) State the characteristic of Ideal fuel.

(C) Answer the following in one sentence [4]

(1) Which semi metal is used in constructing a solar cell.

(2) By which process is ammonia manufactured on large scale.

③ what is biomass?

④ what is the unit of power in British system?

Q 2 A Answer the following to the point [any two]

① describe the conditions leading to goitre and its characteristic symptoms

② State the pre-harvesting measures

③ Describe one different method of food preservation.

(B) Answer the following (any five) [10]

① Under what circumstances does an individual need extra nutrition?

② Give the symptoms of fluorosis.

③ Mention the biotic factors that spoil stored food grains.

④ State the various phases of agriculture farming.

⑤ Give two points of difference between deficiency diseases and communicable diseases.

⑥ Give protective measures against chemical pollutants.

(C) Answer the following to the point in one sentence each [4]

① What are weeds?

② What are enzymes chiefly made up of?

③ Increase of which mineral element in the blood increases blood pressure.

④ Which mineral element is present in haemoglobin?

Answer the following (Any two)

Q3A (1) Describe the carbon cycle in nature. [56]

(2) Describe in detail, the effects of pollution due to noise.

(3) Explain biodegradation, and abiotic degradation, and explain its recycling.

B Answer the following (any five) [10]

(1) Give scientific explanation: It is our moral duty to provide protection to wild life.

(2) What should be done to protect natural wealth?

(3) Distinguish between colloidal solution and crystalline solution.

(4) Describe the present methods of disposing off nuclear wastes.

(5) State the advantages of development of an ideal balanced ecosystem.

(6) What is biogeochemical cycle?

C Answer the following in one sentence each [4]

(1) State the unit of measuring radioactivity.

(2) When did WHO organise a world meet on environmental conservation?

(3) What percentage nitrogen is in the atmosphere?

(4) What is silviculture?

Answer the following (any two)

[56]

Q4A (1) Explain the preparation of ~~ethane~~ methane gas in laboratory.

(2) Write short note on ~~Si~~ silicon.

(3) How is pure alumina obtained from bauxite by Bayer's process?

B Answer the following (any five) [10]

(1) Distinguish between metal and nonmetal according to ~~physical~~ ~~properties~~ chemical property.

- Q4 B (2) Distinguish between saturated hydrocarbon and unsaturated hydrocarbon.
- (3) State your observations with chemical equation.  
~~Phosphorus~~ Phosphorus reacts with chlorine.
- (4) State your observation with equation, sodium <sup>hydro</sup>carbonate is heated.
- (5) Give four uses of sulphur.
- (6) What is ester. Give name of two esters and write their formulae.

C Answer in one sentence only. [4]

(1) Which substance is added to natural rubber to make it hard?

(2) How methanol is prepared?

(3) What is colour of Iron sulphide.

(4) Which compound makes the sea rock hygroscopic.

Q5A

Answer the following (Any two) [5]

(1) State the uses of artificial satellite.

(2) Explain how star is formed from protostar.

(3) Write short note on nylon fibres-

B Answer the following (Any five)

(1) Write short note on meteor.

(2) Distinguish between solid fuel rocket and liquid fuel rocket.

(3) Distinguish between Mars and Mercury.

Q5B(1) Explain why, life is not possible on moon.

(5) Explain Concept of the violet shift.

(6) Distinguish between <sup>thermo</sup> plastic and thermo setting plastic.

Q C Answer in one sentence only [4]

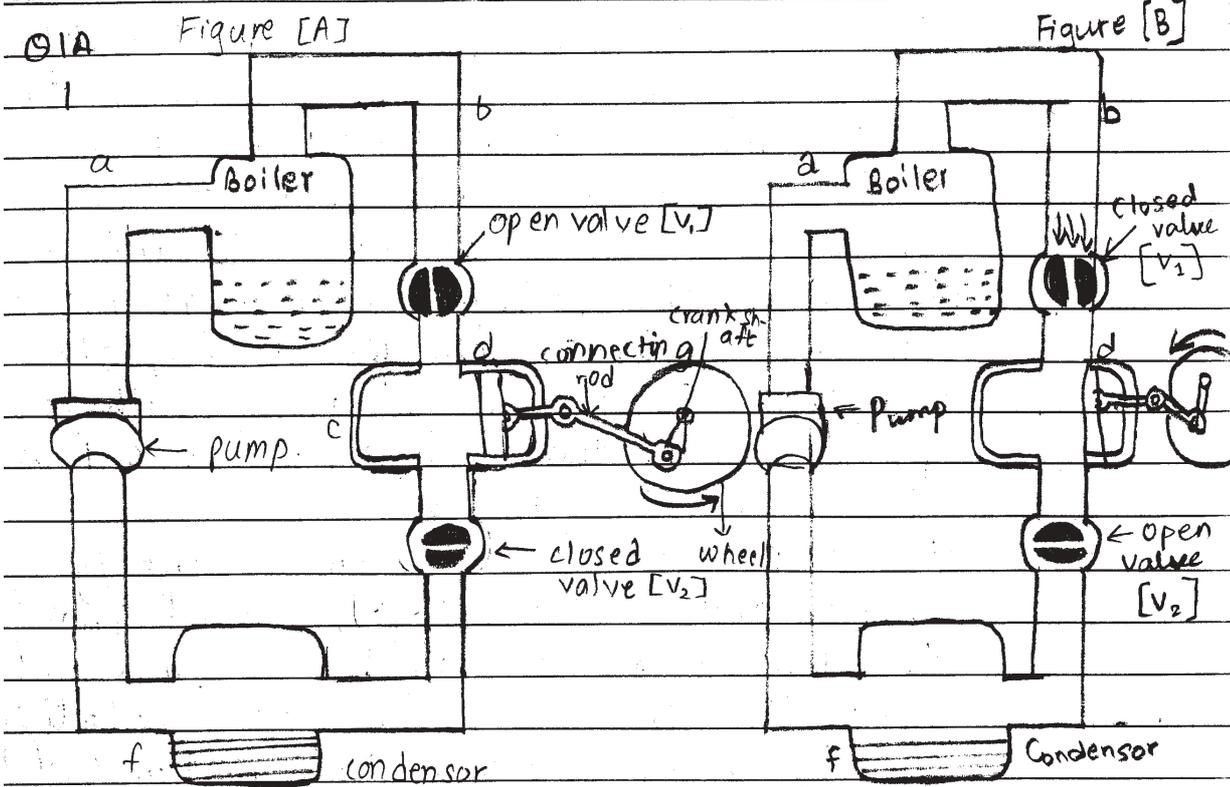
(1) What are asteroids

(2) What was the name of the first man made satellite.

(3) How much percentage of CO<sub>2</sub> is present on Venus.

(4) Which ~~ten~~ technique is used to estimate the age of the solar system.

Answers.



Principle - Conversion of heat energy into mechanical energy.

Working - (1) As shown in the figure water is heated in boiler [a] at high temperature and steam with high pressure is produced.

(2) Steam reaches to valve (V<sub>1</sub>) through pipe [b]

(3) Due to steam pressure, valve (V<sub>1</sub>) opens and steam enters the cylinder (c) with movable piston.

(4) Steam at high pressure exerts pressure on piston (d). So the piston moves to the outer side of the cylinder.

(5) Crankshaft (e) is connected to the piston (d). Crankshaft (e) is connected to a wheel. As steam exerts pressure on the piston (d) the piston moves to the outer

side (right) of the cylinder. So there is a thrust on the crankshaft due to the linear motion of the piston. Thus the linear motion of the piston is transformed into the rotational motion due to crankshaft. Thus mechanical work is obtained.

(6) When the piston reaches to the end of the cylinder, the wheel completes half rotation. So the cooled steam with low pressure in the cylinder moves out side as the valve (V<sub>2</sub>) opens.

(7) Because of the kinetic energy of the wheel, and due to the property of inertia, the wheel remains in the motion and completes the remaining half rotation. During half rotation the piston (c) moves to the inside of the cylinder. So the piston (d) come to its original position and one cycle of the wheel. When the piston comes to its original position, valve (V<sub>1</sub>) is closed and valve (V<sub>2</sub>) is open.

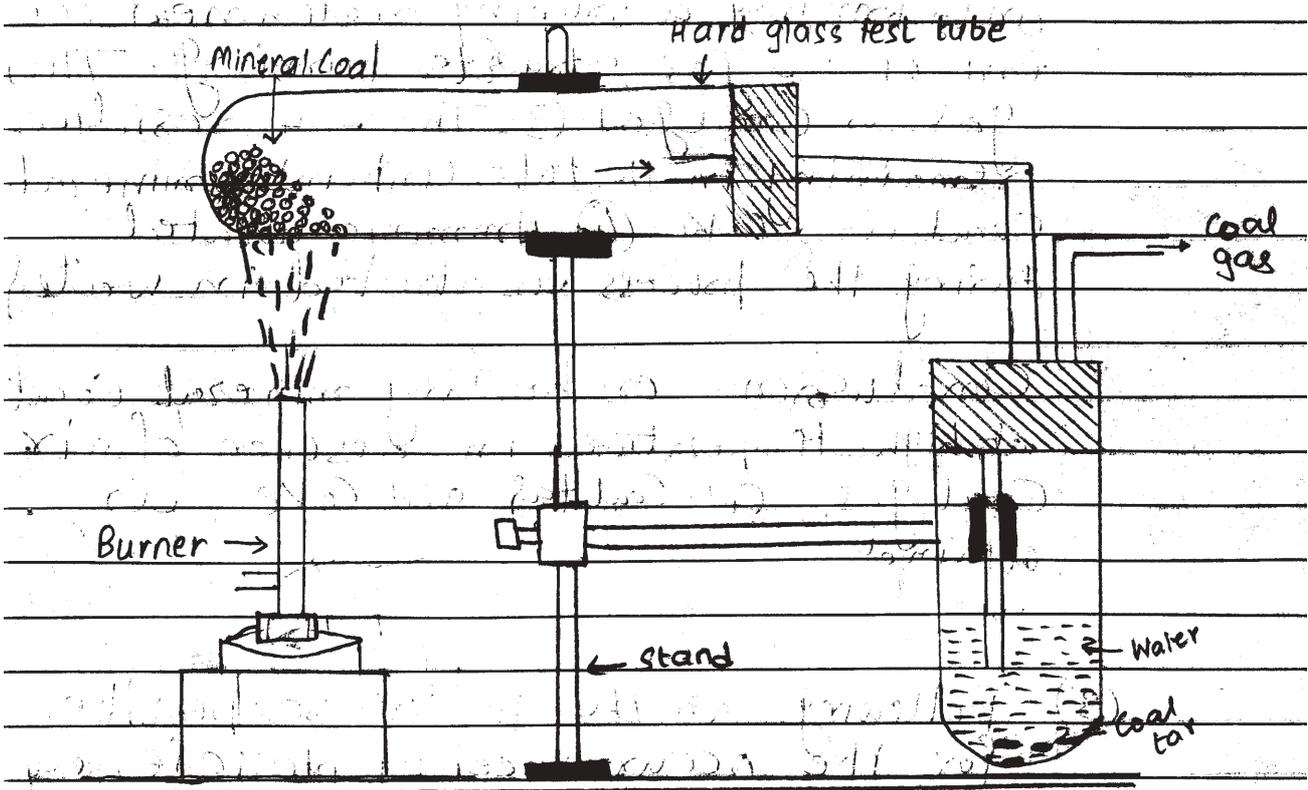
(8) On heating water in boiler (a) and producing high pressure steam. The above process is repeated, and continuous work is obtained.

uses of Steam engine - For running heavy vehicles like railway, steamers and steam roller.

② Aim - To perform the destructive distillation of coal.

Apparatus - Hard glass test tube, Cork with one hole, test tubes, Cork with two holes, delivery tube, Stand, burner.

Substance - mineral coal, coal, water.



Procedure - ① Arrange the apparatus as shown in figure. ② Take some coal in a hard glass test tube. Close this tube airtight with one hole cork. Fix one end of the delivery tube in the hole. Place the other end of the delivery tube in the test tube with some water in such a way that it remains unimmersed in water as shown in figure. ③ From the other hole of the cork on the test tube with water pass another glass tube. [P.T.O]

(1) Heat the hard glass test tube with coal at high temperature with a burner.

Observations:-

(1) Black coloured coal tar collects at the bottom of the test tube with water.

(2) Gas comes out through the end of the tube inserted in the test tube with water on bringing a burning match near the end of the gas starts burning. This gas is coal gas.

(3) The solid residue remaining in the tube containing mineral coal is coke. (4) Ammonia liberated during the process is absorbed in water.

Conclusion- on heating mineral coal at high temperature in absence of air coal gas, coal tar and coke is obtained.

(3) Following are the factors responsible for the occurrence of deficiency disorders.

(1) Because of poverty poor people are unable to take a enough and nutritive diet.

(2) Conventional ways of cooking are also responsible.

(1) On exposing fruits and vegetables for longer time the vitamins in them get oxidised.

(2) On over baking or over frying in oil the nutrients in food gets destroyed.

Q13) (3) On washing pulses and vegetables again again the water soluble vitamins in them are washed away. (4) On polishing rice protein and vitamin B<sub>2</sub> are removed. So there is a possibility of Beriberi.

(3) peeling of green vegetables is harmful because if the skin of vegetables like gourd, bitter gourd, perval, and fruits like banana, bananas, chiku is removed many nutritious substances are lost.

(4) Due to less breast-feeding by mother. ~~a~~ more amount of power powder milk or other baby food will be given to infants. This causes diseases like marasmus in children.

~~5~~

Q13 Answer the following in brief.

(1) Following are the uses of wind energy

(1) Boats and ships are navigated using a sail, using wind energy.

(2) For grinding grains.

(3) Water can be drawn from well.

(4) To generate electricity.

(5) For channelising salty water in to subbed.

(2) Distinguish between petrol - Diesel.  
Petrol Diesel.

(1) In the fractional distillation of petroleum, petrol is obtained in the temperature range of 30°C to 180°C.

In the fractional distillation of petroleum diesel is obtained in the temperature range of 260°C to 340°C.

Petrol.	Diesel.
1B (2) There are 5 to 10 Carbon atoms in its Composition.	These are 14 to 20 Carbon atoms in its Composition.
(3) Its heat energy is 47 KJ/g.	(3) Its heat energy is 45 KJ/g.
(4) uses - As a fuel in scooter, and Car.	uses - used as a fuel in <del>taxi</del> trucks and buses.

(3) State the  
Following are the advantages of Biogas plant.

- 1 (1) NO smoke is produced on burning biogas. So it is pollution free.
- (2) Its caloric value is as high as 35 to 40 KJ/g.
- (3) The material to be used in biogas plant is very cheaply and locally available.
- (4) Natural manure is an important by product of a biogas plant.
- (5) Biogas plant removes the filth of rural area.
- (6) Biogas plant can be prepared for a family or for collective use.

(4) (B) Following are the four causes of energy crisis.

- (1) Electrical energy is continuously used in hotels, shops and bank.
- (2) Electrical energy is used on large scale to run house appliances like air conditioners, refrigerators etc.
- (3) Due to increase of transport vehicles, the consumption of fuel also increased.
- (4) Due to industrial development, use of energy increased by leaps and bounds.

- Q1C.
- ① Silicon semimetal is used, in constructing a solar cell.
  - ② By Haber's process, ammonia is manufactured on large scale.
  - ③ The naturally degradable substance like excreta, urine, garbage and dead leaves are known as ~~bi~~biomass.
  - ④ 'Horse power' is the unit of power in British system.

Q1B(5) Following are the symptoms of Marasmus.

- ① The water content in the body of the child decreases.
- ② The weight of child decreases.
- ③ The skin becomes loose and wrinkled.
- ④ The child looks like a bag of skin and bones.

Q6) Following are the characteristics of Ideal fuel.

- ① The Caloric value of ideal fuel must be high.
- ② It should be quickly combustible.
- ③ It should be easily and cheaply available.
- ④ Its ignition point should be proper according to its use.

Q2A① Conditions responsible for Goitre -

- ① Goitre is caused due to deficiency of Iodine.
- ② The deficiency of Iodine affects the thyroid gland in the throat.
- ③ Due to this the thyroid gland secretes less thyroxin. So the thyroid swells. As a result a large ugly swelling in the neck develops which is known as Goitre.
- ④ Iodine is necessary for the synthesis of thyroxin.
- ⑤ In some regions of Himalayan terrain the amount of Iodine in land and water is less, so Goitre is much seen in these regions of the Himalayas.

Symptoms of Goitre.

- ① The thyroid gland in the throat swells.

(2) A large and ugly swelling develops on the neck. so the physical and mental growth of a person is inhibited.

measures to prevent Goitre.

- (1) Iodised Salt should be taken in food.
- (2) Fish and other iodine-rich sea food should be taken.

(2) Following are the preharvesting measures.

- (1) selection of proper crop
- (2) Procuring seed of high quality
- (3) proper processing of seeds before sowing
- (4) maintaining the fertility of soil through proper cultivation.
- (5) Deciding proper time of sowing
- (6) Adopting proper method of sowing
- (7) Sufficient irrigation at proper time
- (8) Control of weeds and pests.

(3) different methods of preservation of food are as follows

- (1) Disinfection
- (2) Cooling and refrigeration
- (3) Drying
- (4) use of Sugar and Salt
- (5) use of chemicals
- (6) Irradiation of food

Disinfection - This method is used to preserve fruits and for packing jam, pickles, tomato sauce etc.

- (1) The Containers are disinfected at the time of packing.
- (2) The place of work is kept very clean
- (3) Clean cloth is used for wiping jars.
- (4) Jars are filled in bottle jar. So the food does not get spoiled.

(2) Cooling and refrigeration:-

The growth of disease causing germs and the action of enzymes retards at low temperature. So refrigerator or cold-storage are used for preserving food.

(3) Drying - Removal of water from food without spoiling its taste or reducing its nutritive value is known as drying or dehydration. Unripe mangoes, amlas & grapes & walnut etc.

(4) Use of salt and sugar:- Amla, mangoes and lemon are kept immersed in salt water. Mango jam is preserved by the thick syrup of sugar. As the concentration of salt solution of sugar is not allowing a growth of fungus.

(5) Use of other chemicals - If substances are kept in an acidic medium, the germs do not survive in it. Benzoic acid and potassium meta bisulphite are used for this. Tomato ketchup, mango juice are preserved in this way.

(6) Radiation: Gamma rays are used for storage of potatoes and onions. By this method they can be preserved for a long time.

[B] Answer the following -

(1) Under following circumstances one should need extra nutrition.

(1) A pregnant woman requires more amino acid, calcium and iron for the development of the tissues of the embryo.

(2) A lactating mother should take fortificed milk and vitamins for growth and immunity.

Q. 20B. (2) The following are the symptoms of Fluorosis.

- (1) Fluorosis causes damage to intestine, kidney.
- (2) It makes bones and teeth.
- (3) In mild form fluorosis there are dots on the teeth and the luster of the teeth vanishes.
- (4) There are tremours in the body.
- (5) There is pain in the lower region of vertebral column and gradually the back more stiff.
- (6) Hunger, decreases.

(3) Following are the biotic factors that spoil the stored food grains.

- (1) Rat — Rats do much damage to grains in the field, store houses. It drops its hair on the grains and contaminate it.
- (2) Birds: — Birds causes damage to fruits, vegetables and grains in the field. Birds excrete on the grain and drop their feathers on them.
- (3) Insects: — Some insects of stored food grains are ~~carry~~ carriers of fungi and bacteria. These insects lay eggs in the grain. The larvae coming out from them spread foul smell and spoil the food grain.

(4) Following are the various phases of agriculture farming.

- (1) Preparing the soil: — The soil can be prepared by using different types of plough i.e. tilling.
- (2) Levelling the soil — Levelling of the soil is done by using Harrows.
- (3) mixing fertilizers: — Fertilizers are mixed mixed by using wheel barrow.
- (4) Sowing — Sowing is done by sower.
- (5) Irrigation — Irrigation, is done by using spade by preparing out let for water.

Q2B (5) Deficiency diseases

- ① Disease caused by inadequate nutritive substances is known as deficiency disease.
- ② Ex. Kwashiorkor, Marasmus.

Communicable diseases

diseases which is spread through virus or person to person.

Known as Communicable diseases.

Ex. Small pox, Chicken pox.

(6) Following are the protective measures against chemical pollutants.

- ① The chemical industries like chemical fertilizers, colours and plastic industries should be set up away from city.
- ② The ~~water~~ chemical, like BHC, DDT, should be used as per requirement.
- ③ Waste water of chemical factory should be purified in the factory only.

(C) ① Unwanted plants growing with the standing crops are called weeds.

- ② Enzymes are chiefly made of protein.
- ③ Sodium Sodium mineral element in blood increases blood pressure.
- ④ Iron mineral element is present in the haemoglobin.

Q3A ① Carbon cycle:- The process of conversion of carbon dioxide of atmosphere into various carbonic compounds and then back again into carbon dioxide. From those compounds is known as Carbon cycle:-

Passing of Carbon dioxide in the atmosphere.

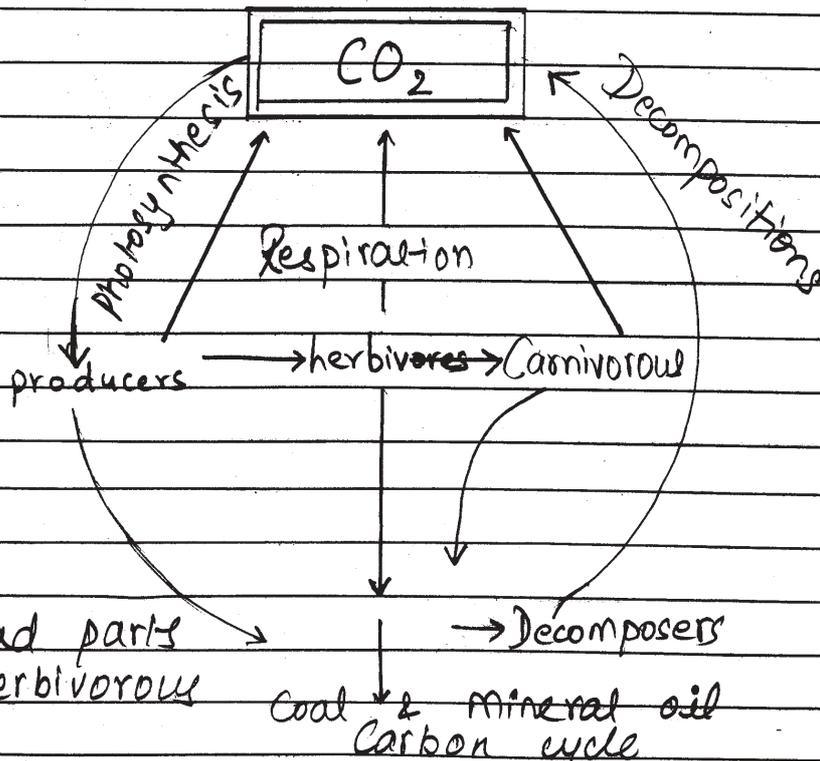
- ① There is 0.03% carbon dioxide in atmosphere
- ② In sea water, there is 50% more carbon dioxide than in the atmosphere

Q3A(1) (3) On exhalation of trees, and animal Carbon dioxide is given out and it passes into the atmosphere.

(4) Due to industrial enterprises like mills, factories etc. Carbon dioxide mixes with the atmosphere.

(5) On combustion of fuels like petrol diesel etc. in vehicles carbon dioxide is produced and it passes in to the atmosphere. Use of the Carbon dioxide of the atmosphere

(6) Green vegetable utilises carbon dioxide directly from atmosphere for photosynthesis and liberates oxygen. Thus air gets purified.



(2) An unwanted sound of very high intensity is known as noise.

Effects of pollution due to noise -

(1) Noise creates very adverse effect on the brain and physical health.

(2) The duration and intensity of sound decreases the efficiency of a man

Q3 A② (3) The hearing capacity of a man is temporarily or permanently damaged due to sound.

(4) The concentration of a man is disturbed.

(5) Noise creates nervous tension.

(6) Noise causes can cause high blood pressure or problem of heart.

(7) Noise may cause headaches or giddiness.

(8) Noise may disturb the mental piece of a man.

(3) Waste matter:- There are two types of waste matter.

(1) Biodegradable substances - Excreta, urine, garbage, paper etc etc.

(2) Non bio degradable - Plastic goods, Polythene bags etc.

Biodegradation - The comprehensive process of reducing the complex substances into simpler, elemental components by naturally occurring decomposers transformers. is known as bio-degradation.

Recycling of biodegradable substances -

(1) Compost manure should be made from biodegradable substances like excreta, urine and garbage.

(2) Paper is remade from agricultural waste.

Non biodegradation -

In many cases the form or structure of a substance or component cannot be altered by natural factors. This is known as non biodegradation.

Answer the following.

Q 3B (1) It is our moral duty to protect wildlife because

(1) Due to killing of wild animal these numbers will decrease. This creates an imbalance in the living world.

(2) If wild animal like lion, tiger is destroyed, man can not recreate them.

(2) Following points should be kept in mind to protect natural wealth.

(1) Control must be exercised on the use of natural resources.

(2) The area of forests and the number of trees must be increased by Vanmahotsav.

(3) Forest area should be reserved for conservation of wild animal.

(4) Tiger are protected by tiger project.

(3) Colloidal solution - Crystalline solution.

(1) The solutions whose permeation through membranes of animals is very slow is known as colloidal solution. - The solution whose permeation is very fast through membranes of animal origin are known as crystalline solution.

(2) The said permeation of gum, glue and starch is very slow. - The said permeation of salt sugar etc. is very fast.

(4) Following are the advantages of the balance of an ecosystem.

(1) Because of balanced ecosystem the proportion of gases is maintained on earth.

(2) The ratio of oxygen and Carbon dioxide in the atmosphere is maintained.

(3) Carbon cycle, nitrogen cycle and oxygen

3.B (4) go on continuously and ~~so~~ regularly.

(5) ~~de~~ The present methods of disposing off nuclear wastes is as follows  
At present such radioactive wastes are packed in strong and leak proof containers and are stored in specially constructed store houses of cement concrete.

(6) The cyclic flow of elements, like carbon, nitrogen, oxygen etc. between an ecosystem and its physical environment is called a biogeochemical cycle. The main elements enter in the producer and through them in the ecosystem.

C (1) The unit of for ~~measur~~ measuring radioactivity is 'Röntgen'.

(2) In 1972 UNO organised a world meet on environmental conservation at Stockholm.

(3) The percentage of nitrogen in the atmosphere is 78%.

(4) modern & Technique adopted by the forest division for rapid growth and development of trees is called silvi-culture.

Q4 A (1) Aim - To prepare methane gas

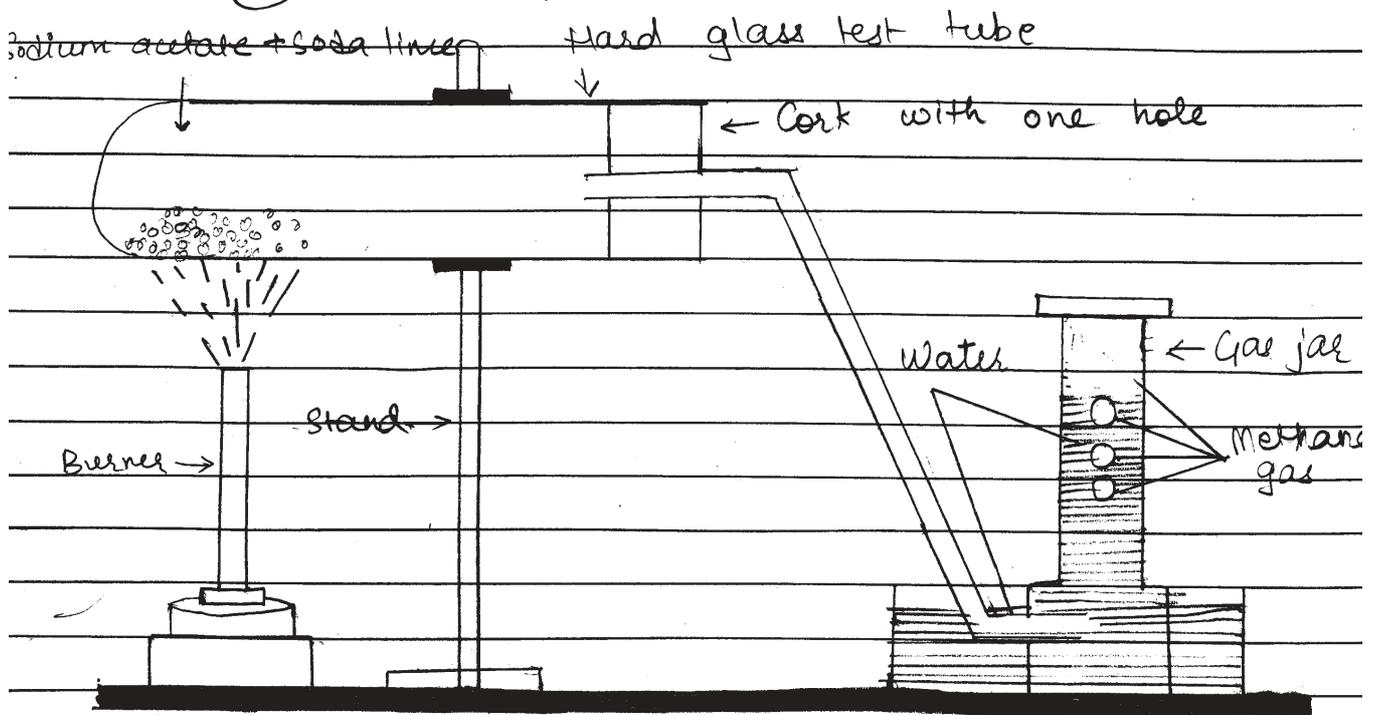
Apparatus - Hard glass test tube, burner, stand, delivery tube, beehive shelf gas jar. Cork with one hole.

materials - sodium acetate, soda lime, water -

Procedure - Take a mixture of 2g sodium acetate  $\text{CH}_3\text{COONa}$  and 2g sodium lime. Arrange the apparatus as shown in fig.

- ② Close the test tube with cork with one hole. Pass one end of the delivery tube through the hole in the cork. Place the second end in the beehive shelf kept in water.
  - ③ Put an inverted gas jar filled with water over the beehive shelf.
  - ④ Place burner below the test tube.
  - ⑤ When mixture in test tube is heated the chemical reaction takes place between sodium acetate and soda lime and methane gas is liberated.
- $$\text{CH}_3\text{COONa} + \text{NaOH} \xrightarrow{\Delta} \text{CH}_4 \uparrow + \text{Na}_2\text{CO}_3$$
- ⑥ methane produced is collected by downward displacement of water.

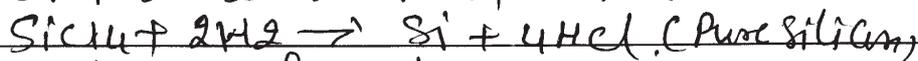
~~② Close the.~~



② The atomic number of Silicon is 14. The electronic configuration of Silicon is 2, 8, 4.

Sources of Silicon -

- ① In the earth crust silicon is second to oxygen.
- ② Silicon is found in clay, mica, granite and feldspar.
- ③ Method to obtain pure Si silicon, Silicon is reacted with first chlorine and then with hydrogen.



Properties of Silicon.

- ① It possesses external lustre.
- ② It forms alloys with metal like iron and aluminium.
- ③ Silicon is semiconductor.
- ④ melting point of silicon is high.

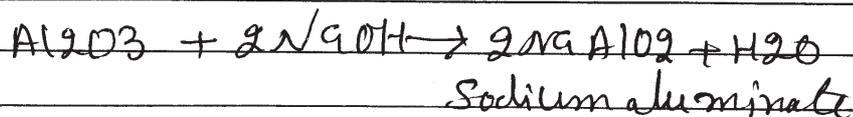
Uses of Silicon -

- ① It is used in preparation of solar cell.
- ③ It is used in preparing silicon polymer.
- ④ It is used as semiconductor.
- ⑤ Silicon carbide is used in making tools for cutting hard metals and rock.

③ The method for purification of bauxite ( $\text{Al}_2\text{O}_3 \cdot 2\text{H}_2\text{O}$ ) was discovered by Bayer. - So it is known as Bayer's method.

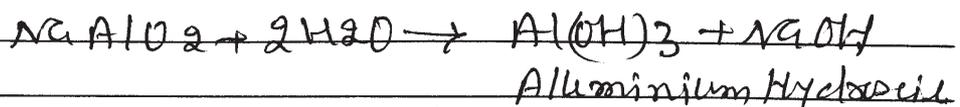
- ① First bauxite is powdered. On heating its ferrous oxide get converted into ferric oxide.

Q4A(3) (2) The waste bauxite is dried and ~~powdered~~ powdered. Then it is mixed with 45% solution of sodium hydroxide and heated in a closed vessel for 6 to 8 hours at 5 to 6 atmospheric pressure at about  $160^{\circ}\text{C}$ . So bauxite gets converted into sodium aluminate according to the following reaction.

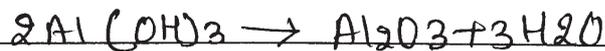


Insoluble ~~impurities~~ impurities collect at the bottom of the vessel. On filtration, soluble sodium aluminate is obtained in the filtrate.

(3) Excess water is added to sodium aluminate and stirred. So white gelatinous precipitates of aluminium hydroxide is obtained.



(4) These precipitates are repeatedly washed with water and then dried. On heating them at  $1100^{\circ}\text{C}$   $\text{Al}_2\text{O}_3$  is obtained.



Thus by this process about 99.5% pure alumina -  $\text{Al}_2\text{O}_3$  is obtained.

QUBT metal

① There are 1, 2, 3 free electrons in the last orbit.

② They have tendency to lose electrons.

③ They form positive ions.

④ Their oxides are generally basic.

Non metal

There are 5, 6, 7 electrons in the last orbit.

② -

They have tendency to gain electron.

They form negative ions.

Their oxides are acidic.

② Saturated hydrocarbon

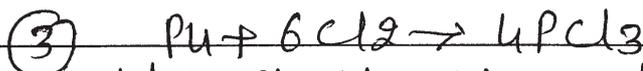
① The hydrocarbon in which one single covalent bond between carbon-carbon atoms are known as saturated hydrocarbon.

② Their general formula is  $C_nH_{2n+2}$

Unsaturated hydrocarbon

The hydrocarbon in which there are double or triple bond between carbon-carbon atoms are known as unsaturated hydrocarbon.

Their general formula is  $C_nH_{2n-2}$  and  $C_nH_{2n}$



When phosphorus reacts with chlorine phosphorus trichloride is formed.

④ When sodium <sup>bicarbonate</sup> carbonate heated it decomposes into sodium carbonate and carbon dioxide is liberated.



⑤ Four uses of Sulphur

① Sulphur is mainly used in the manufacturing of sulphuric acid.

② It is used in preparing ointment for skin diseases.

Q4B(5) (3) It is used in preparing solvent like carbondisulphide.

(4) It is used in making fireworks and in gun powder.

(6) Oxygenic Compound having -COO group are known as esters.

Name                      formulae.

methyl formate -  $\text{HCOOCH}_3$

methyl acetate -  $\text{CH}_3\text{COOCH}_3$

Q4C (1) Carbon black is added to natural rubber to make it hard.

(2) methanol is prepared when wood is heated in the absence of air.

(3) The colour of Iron sulphide is black.

(4) magnesium chloride makes the sea rocks hygroscopic

Q5A(1) Following are the uses of Artificial satellite.

(1) To make radio and T.V. programmes easily available.

(2) For Commercial Communication.

(3) For weather monitoring and forecasting.

(4) For detecting water and mineral resources.

(5) For detecting diseases in plant and trees.

(6) Useful in oceanographic studies.

(9) Protostar - If the cloud hydrogen gas in a galaxy is dense and large, the gravitation force of attraction

QSA 2) between its molecules is very strong and the cloud start contracting. On adequate contraction the cloud contracts into a dense gas material, which is known as a protostar.

Protostar

① At this stage the protostar does not glow as sufficient energy is not emitted.

② The contraction of a protostar can continue for million years. In due course it start glowing and star form.

Protostar

① The contraction of a protostar continue for about 1 million years. During this inner temperature increases from  $-173^{\circ}\text{C}$  to  $10^7^{\circ}\text{C}$ .

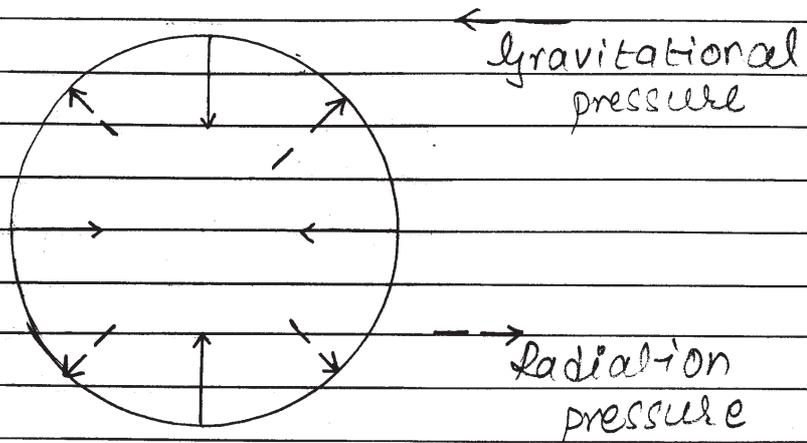
② At this temperature thermonuclear fusion starts. In this four hydrogen nuclei combine to form helium nucleus.

③ The mass of helium nucleus so formed is slightly less than the mass of four hydrogen nuclei. This mass which is lost is converted in to heat energy according to Einstein's formula  $E = \Delta m \times c^2$  where  $E$  is the heat energy,  $\Delta m$  is lost in mass and  $c$  is the velocity of light.

④ The energy released due to fusion exerts radiation pressure. It acts in the opposite direction of Gravitational force which acts towards the centre.

⑤ The tremendous amount of heat energy released during this process is in the form of visible light and other electromagnetic waves. So the protostar starts glowing. Thus star is

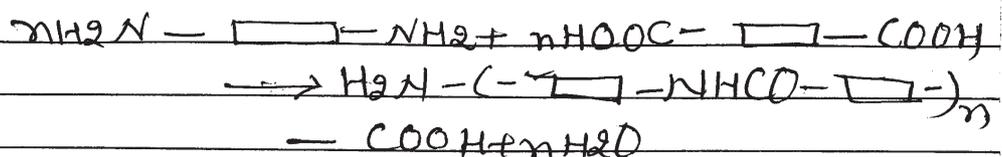
Q.5.A.2 (a) (b) It takes billions of years to get a protostar converted into star.



- (3) Short note on :- nylon fibres
- (1) Nylon are manmade fibres
  - (2) In nylon, there are innumerable monomers of amides - NHCO group linking continuously
  - (3) Nylon is polymer of polyamide type.

Method to prepare nylon :-

- (1) Nylon is obtained by the process of polymerisation between two organic compounds having (-NH<sub>2</sub>) and two carboxylic (-COOH) groups. The appearance of nylon is like silk.



where 'n' shows very large number of Carboxylic Compounds and  $\square$  denotes the hydrocarbon of the compounds.

Q5A(3) (1) In 1935 American Scientist first obtained nylon on the basis of amid group. But the Commercial production was started in 1939.

Q5B(1) + Short note on meteor.

- (1) meteor is a space objects.
- (2) when it enters the earth's atmosphere it starts burning and bright streak of light is seen in the sky.
- (3) In Common language it is known as shooting star.
- (4) most of the meteors burn out but some do not burn.
- (5) The left over part of a meteor falls on the earth is known as a meteorite.

(2)

solid fuel rocket

- (1) Solid fuel is used.
- (2) After stopping it, it can not be restarted again.
- (3) It can be preserved long time.
- (4) It is used for military purpose.

liquid fuel rocket.

- (1) Liquid fuel is used.
- (2) It can be started or stopped as per our desire.
- (3) It can be preserved long time after filling.
- (4) It is used for space research.

(3)

mass

- (1) It is outer planet
- (2) It has two satellites
- (3) It has thin atmosphere.
- (4) daytime temperature is  $40^{\circ}\text{C}$

mercury

- (1) It is inner planet
- (2) It has no satellite.
- (3) It does not have atmosphere.
- (4) daytime temperature is  $525^{\circ}\text{C}$

Q3 R(4) Life is not possible on the moon.

(1) There is no atmosphere on the moon.

(2) The temperature on the moon, during daytime is  $100^{\circ}\text{C}$  and during night is  $-115^{\circ}\text{C}$ . Thus difference is  $215^{\circ}\text{C}$ . So life is not possible on the moon.

(5) If a star moves towards the earth the frequency of light emitted by it when received on the earth gradually increases.

The position of the corresponding spectral line in the visible spectrum would shift towards violet.

This is known as violet shift.

(6) Thermoplastic

(1) In thermoplastic molecules of polymer form long chain.

(2) It can be moulded into desired shapes.

Thermosetting plastic

In thermosetting plastic polymers are cross linked.

once moulded, its shape cannot be changed on heating.

S-C. (1) Asteroids are the debris of small objects made of rocks and minerals, which fail to assemble into planet.

(2) The name of the first man made satellite was Sputnik-I.

(3) 97% of  $\text{CO}_2$  is present on Venus.

(4) Radiometric dating technique is used to determine the age of the solar system.